

Fuel Cycle – SULI PPPL

Dr. Emre Yildirim –
Assistant Chief Engineer



Presenter Slide

2015

Mechanical
w/Nuclear
Engineering *BEng*

2018

Materials Science
(Fusion Materials)
PhD

2023

PFC Development
Engineer –
Tokamak Energy

2024

Assistant Chief
Engineer – Tritium,
Power and
Utilities



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The biggest thing I have learnt
is to be interested in what
others are doing and take
opportunities!



Talk Outline

- Why the Fuel Cycle matters
 - And its link to the plasma
- What makes up a fuel cycle?
 - Inventory and flow
 - Technologies (there are many – we will cover *some*)
- Challenges in the Fuel Cycle
 - Where are we now and where do we need to be?
- Who works on the fuel cycle?



Why the Fuel Cycle Matters



The importance of the fuel cycle

Continuous,
efficient supply

Self-sufficiency

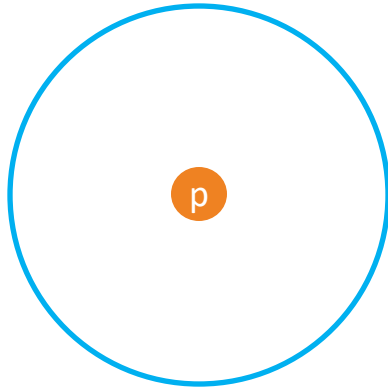
Low burn
fraction

Accountancy

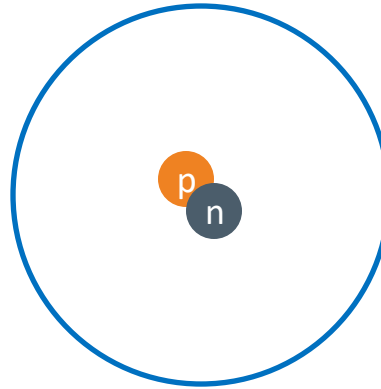
Safety



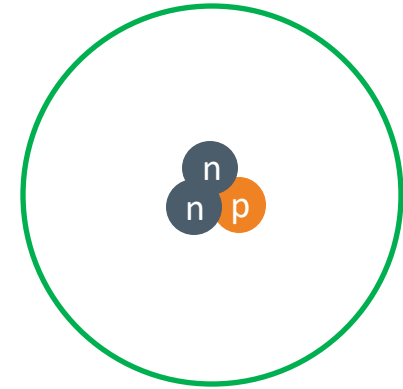
The hydrogen isotopes



- **Protium**
- Abundant
- Not used as a fusion fuel



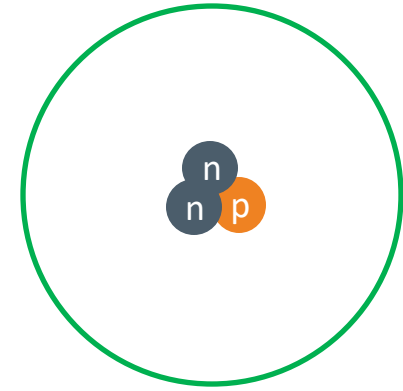
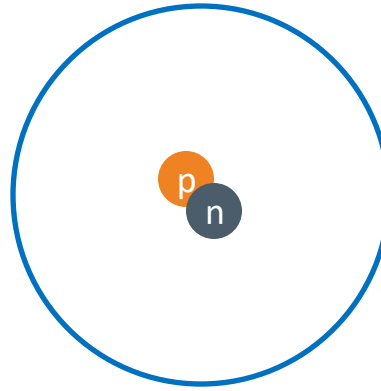
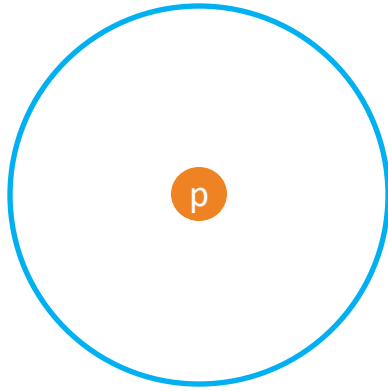
- **Deuterium**
- Naturally Occurring
– 130-160 ppm in H
- Continuously consumed



- **Tritium**
- Beta emitter (18.6 keV)
- 12.3 year half life
- ~30 kg total global inventory
- Need to produce this continuously!



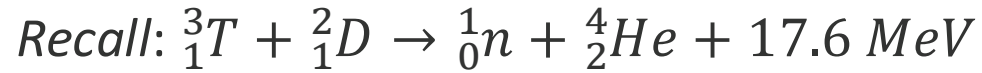
The hydrogen isotopes



Property	Hydrogen (^1H)	Deuterium (^2H or D)	Tritium (^3H or T)
Atomic number	1	1	1
Neutrons	0	1	2
Relative atomic mass	1.00784 u	2.01410 u	3.01605 u
Mass (kg/mol)	1.00784 g/mol	2.01410 g/mol	3.01605 g/mol
Boiling point (K)	20.271 K	23.67 K	~25 K
Radioactive?	No	No	Yes (β^- , $t_{1/2} \approx 12.3$ yrs)
Natural abundance	>99.98%	~0.0156%	Trace ($\approx 10^{-18}$ in nature)
Diatomic as a gas?	Yes	Yes	Yes



How much Tritium do we use?



1,000 MW_{fus}

$$E_{\text{reaction}} = 17.6 \text{ MeV} = 17.6 \times 1.602 \times 10^{-13} \text{ J} = 2.8195 \times 10^{-12} \text{ J}$$

$$1 \times 10^9 \text{ J s}^{-1} / 2.8195 \times 10^{-12} \text{ J reaction}^{-1} = 3.55 \times 10^{20} \text{ reactions s}^{-1}$$

M_T = 3.016 g/mol

$$3.55 \times 10^{20} \text{ atoms s}^{-1} / 6.022 \times 10^{23} \text{ atoms mol}^{-1} = 5.90 \times 10^{-4} \text{ mol s}^{-1}$$

$$5.90 \times 10^{-4} \text{ mol s}^{-1} \times 3.016 \text{ g mol}^{-1} = 1.78 \times 10^{-3} \text{ g s}^{-1}$$

~ 56.1 kg year⁻¹



Tritium Sources and Scarcity

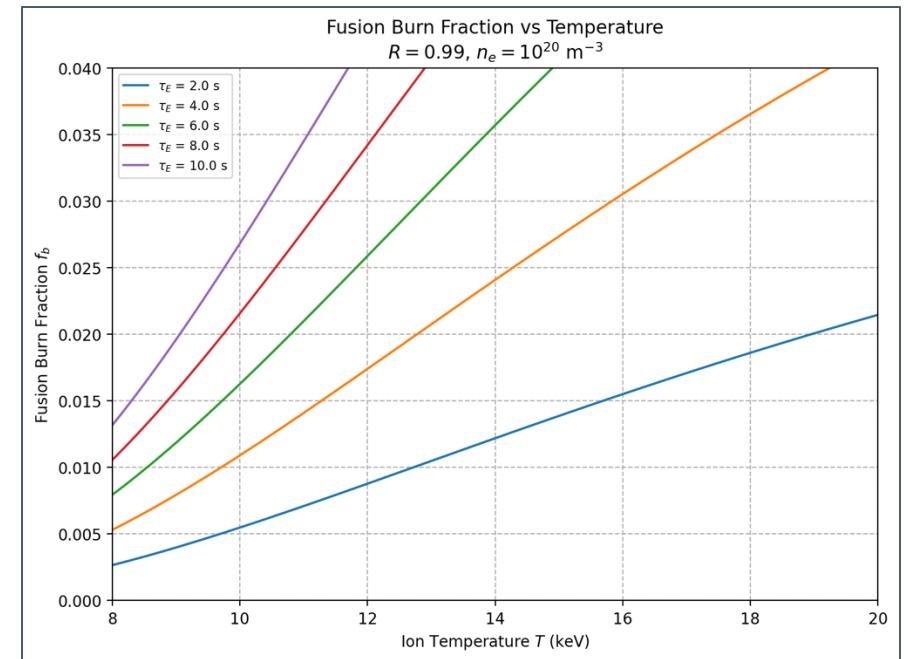
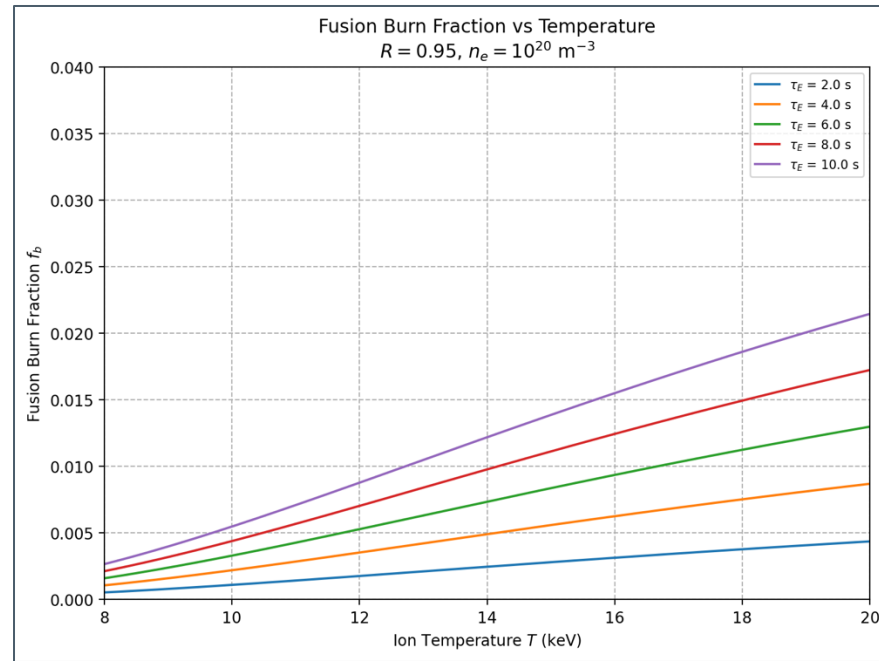
- Only current sources of tritium: Canada, South Korea and Romania
 - Through heavy water moderated fission plants
- With ~ 30 kg in commercial existence – cost is ~ 30,000+ \$/g
- Tritium use in defense – highly regulated commodity
- Produced in fusion through Lithium interaction with a neutron:
 - ${}^6_3\text{Li} + {}^1_0\text{n} \rightarrow {}^3_1\text{T} + {}^4_2\text{He} + 4.78 \text{ MeV}$ (requires thermal neutron)
 - ${}^7_3\text{Li} + {}^1_0\text{n} \rightarrow {}^3_1\text{T} + {}^4_2\text{He} + {}^1_0\text{n} - 2.47 \text{ MeV}$ (requires high energy neutron)
 - Natural lithium is 7.5 % Li-6 and 92.5 % Li-7



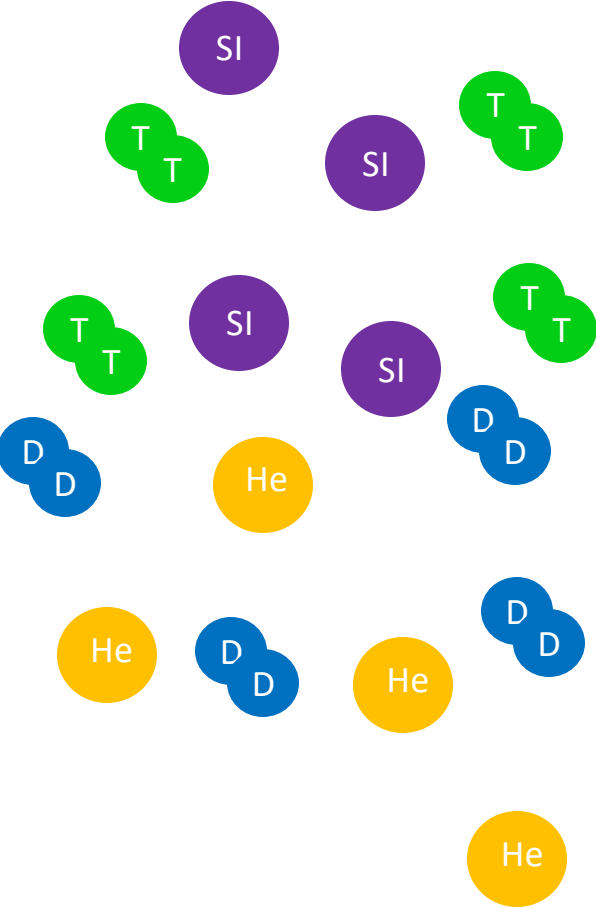
How much Tritium burns?

$$f_b = \frac{n\langle\sigma v\rangle\tau^*}{2 + n\langle\sigma v\rangle\tau^*}$$

$$\tau^* = \frac{\tau}{1 - R}$$

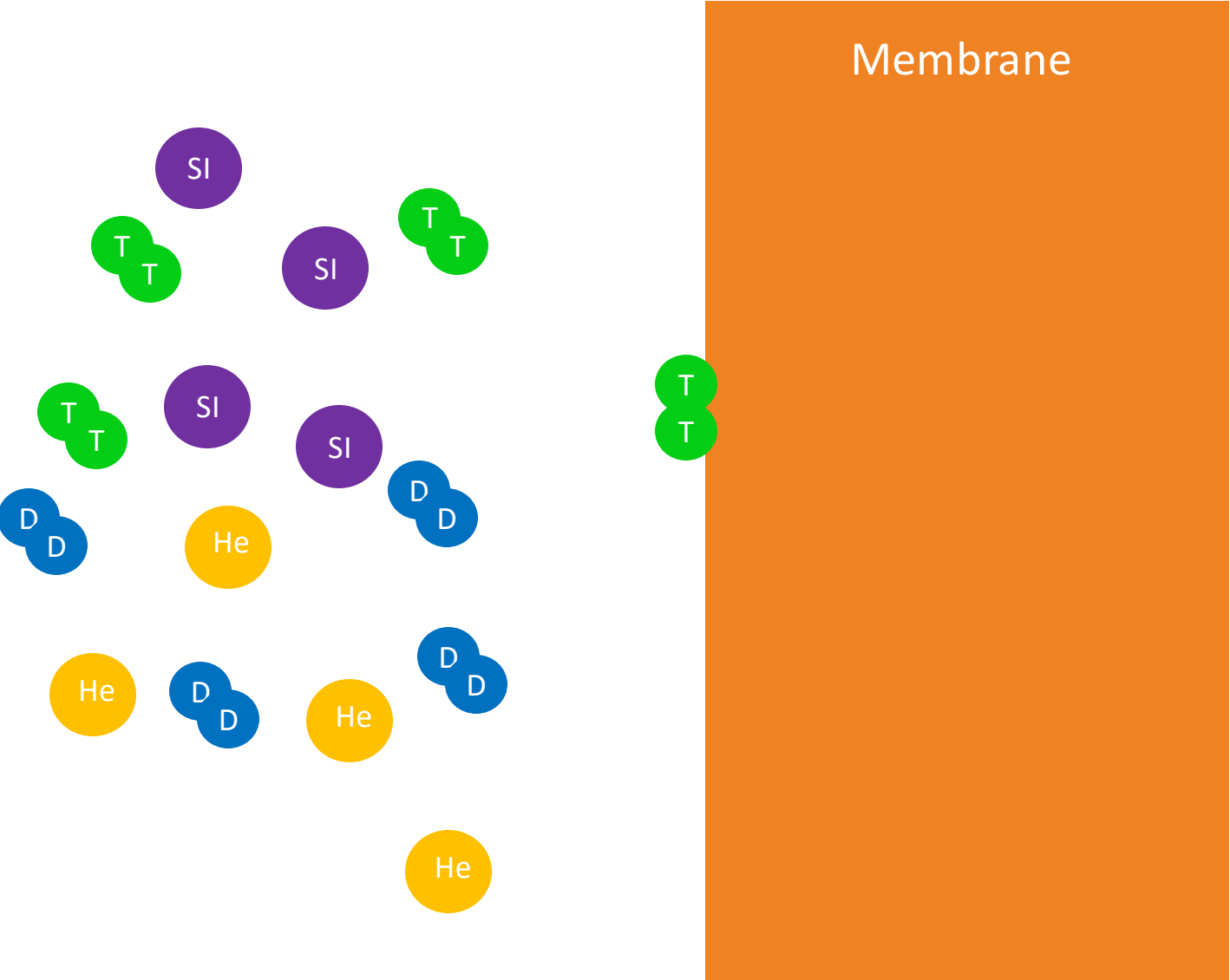


Key Concepts in the Fuel Cycle - Permeation



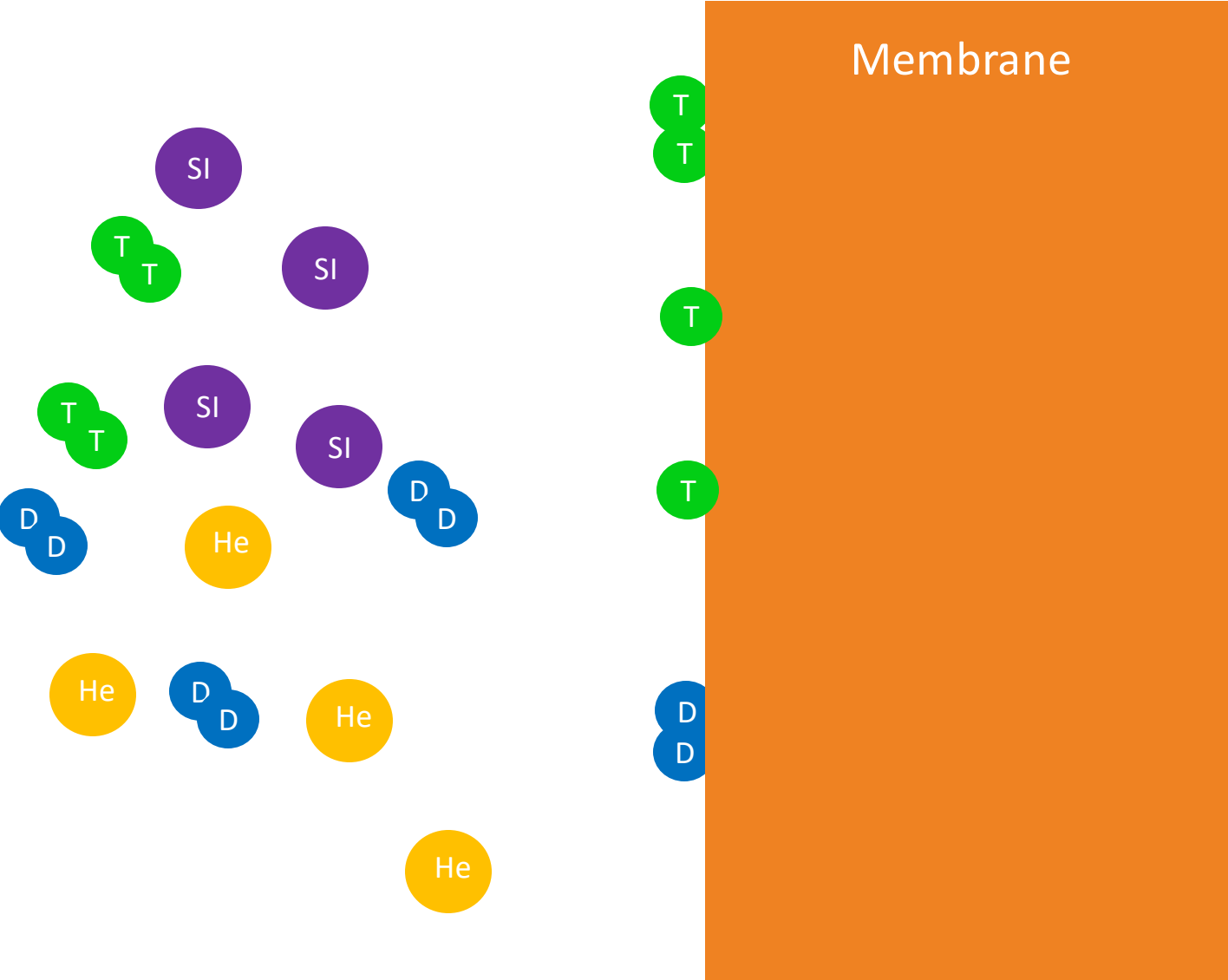
Key Concepts in the Fuel Cycle - Permeation

Adsorption



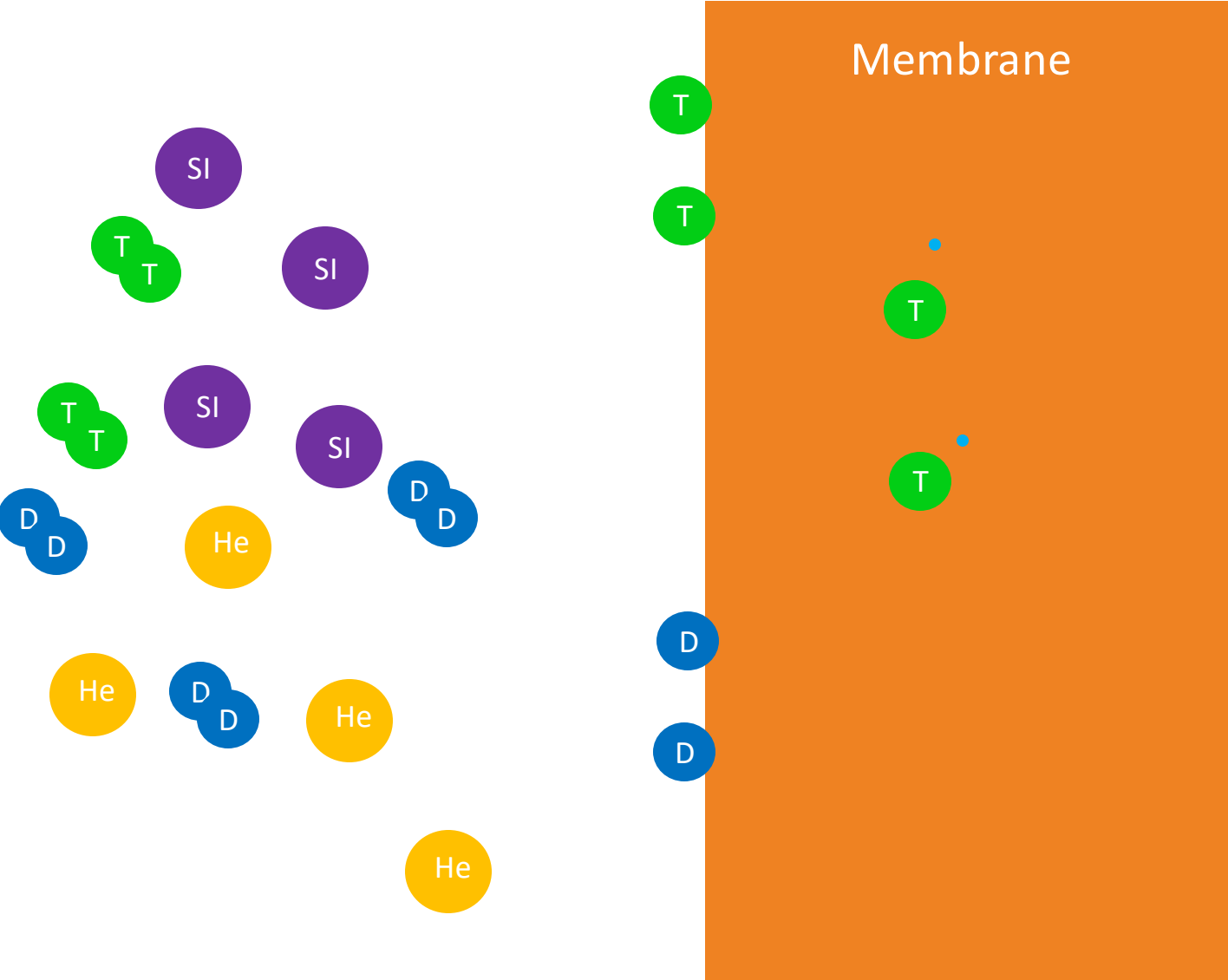
Key Concepts in the Fuel Cycle - Permeation

Dissociation



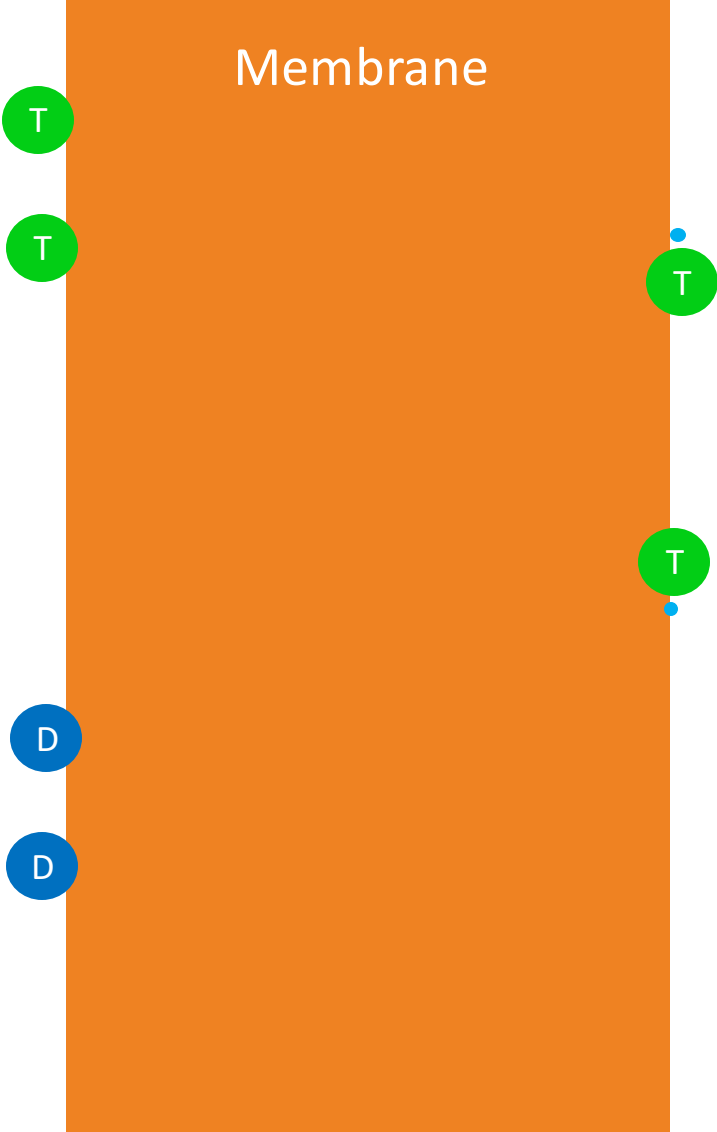
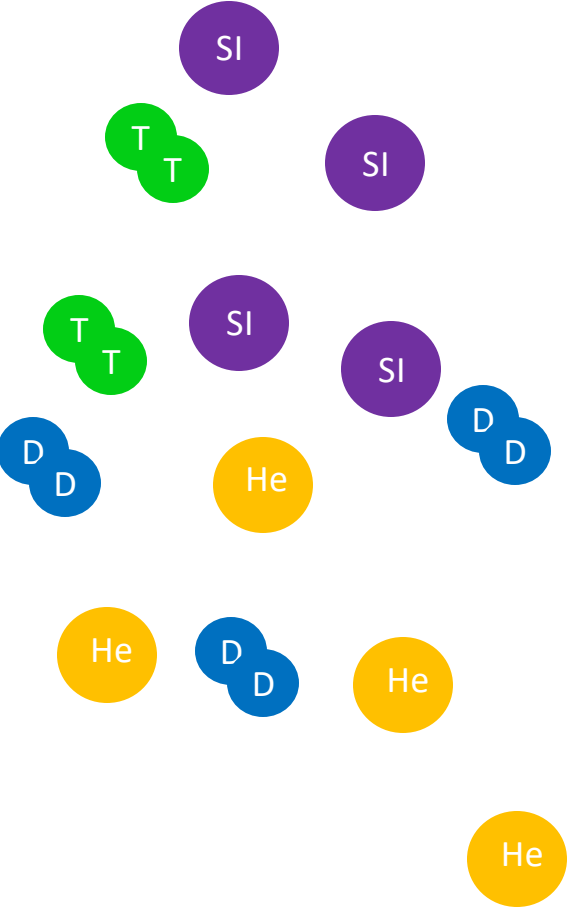
Key Concepts in the Fuel Cycle - Permeation

Ionization



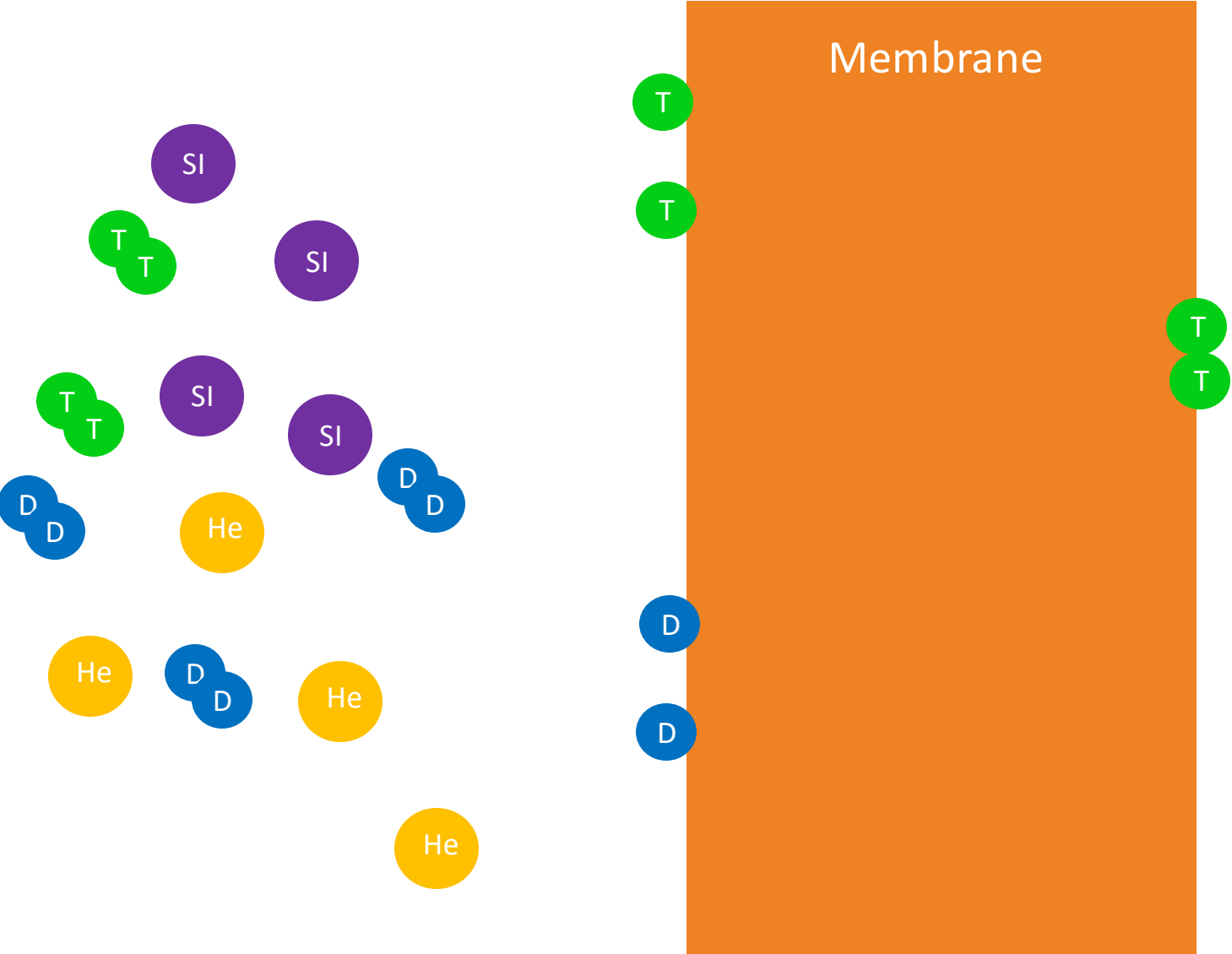
Key Concepts in the Fuel Cycle - Permeation

Diffusion



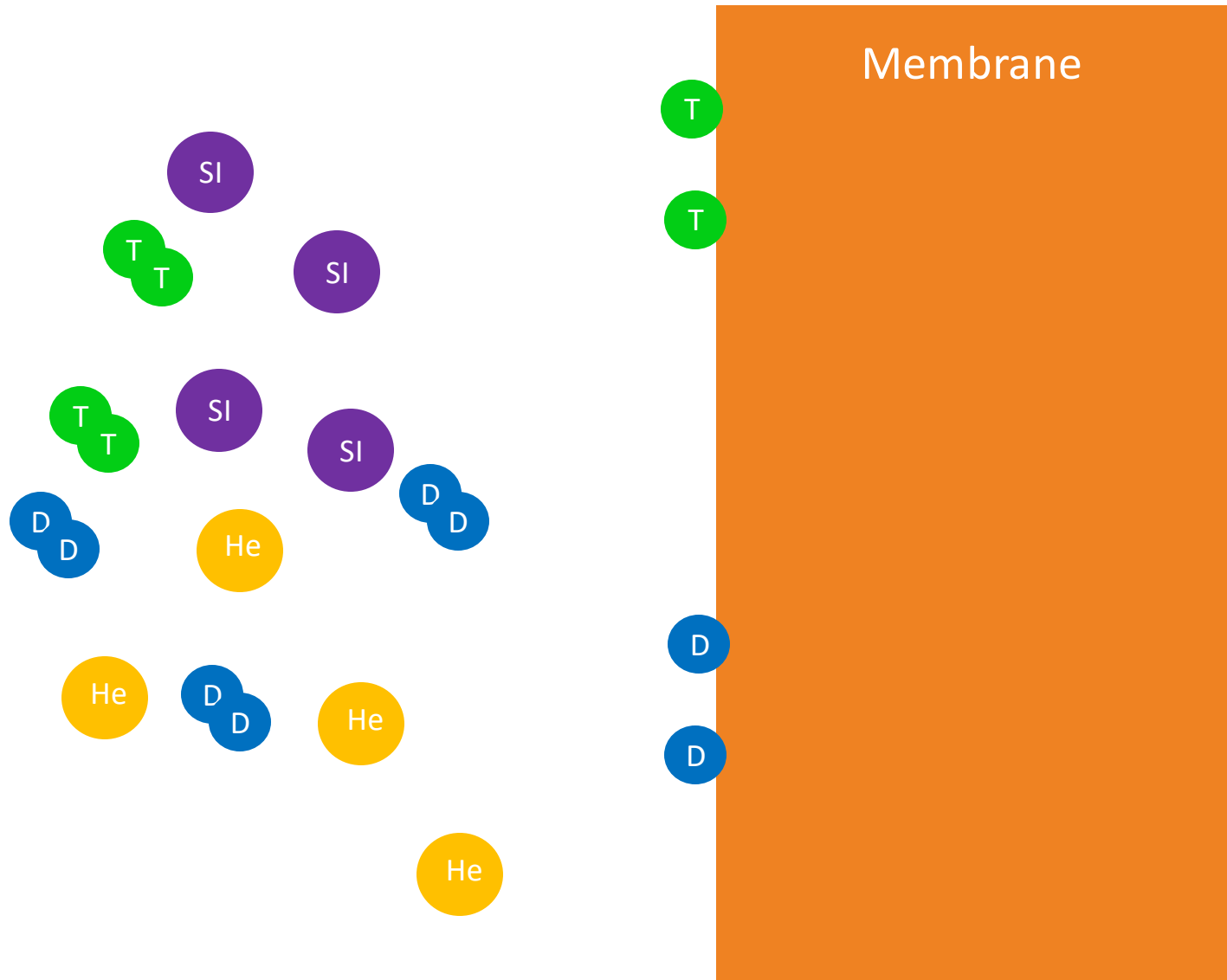
Key Concepts in the Fuel Cycle - Permeation

Recombination



Key Concepts in the Fuel Cycle - Permeation

Desorption



Materials with good adsorption properties: Pd

Materials with fast diffusive properties: V, Ta, Ti

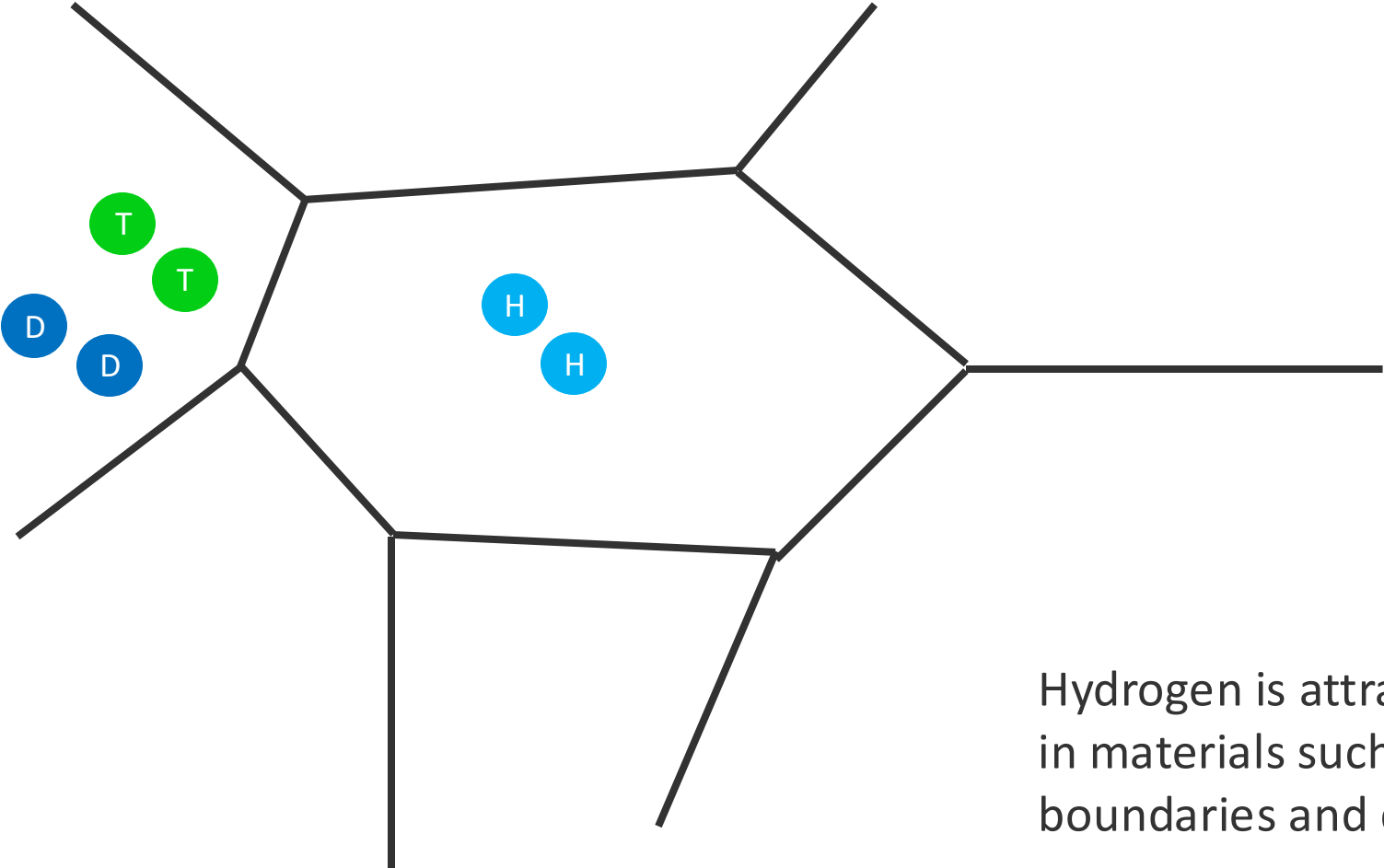


Key Concepts in the Fuel Cycle – Isotopic Differences

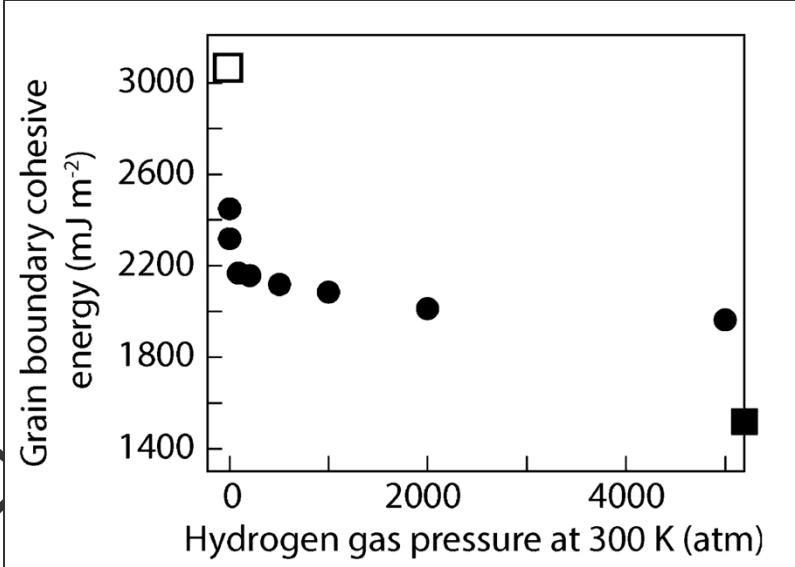
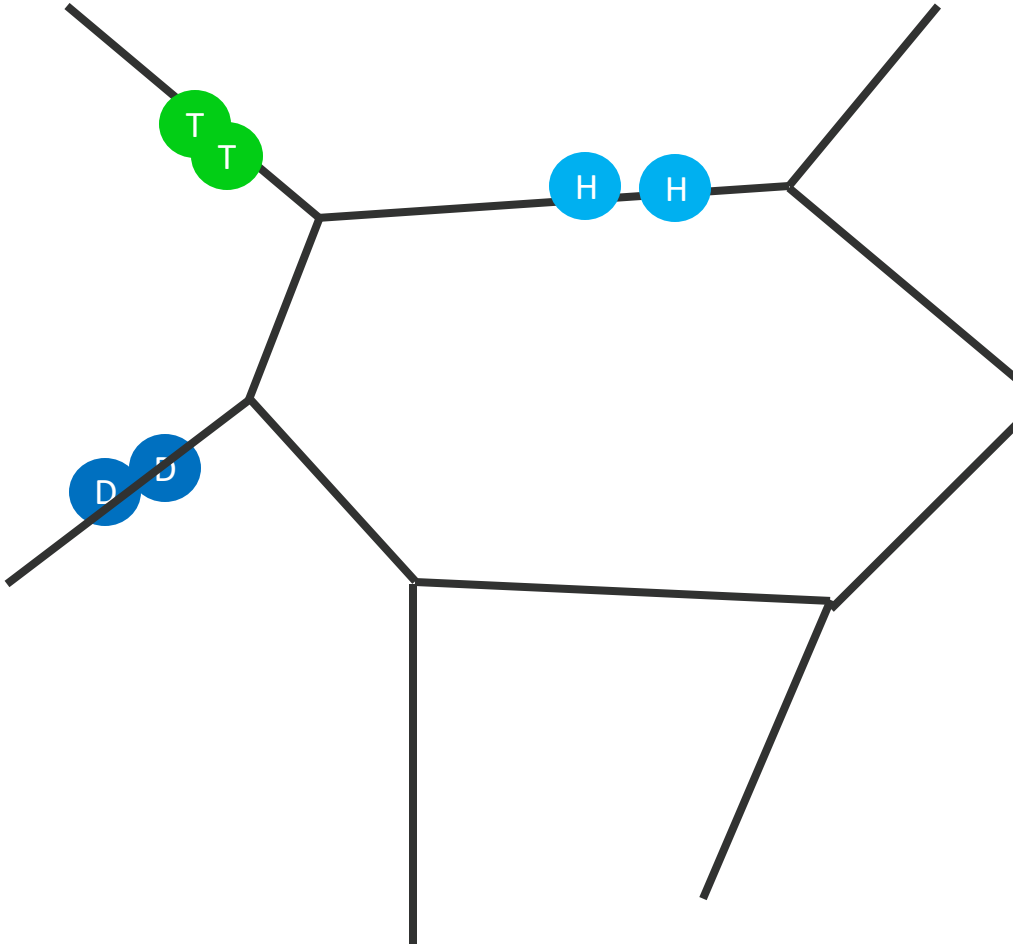
- Difference in mass leads to:
 - Differences in boiling points
 - Difference in diffusivity through materials
- Isotopic exchange can also occur
 - Tritium will prefer to be in bigger polar molecules like water vs. staying in HT form
 - This process is instantaneous
- These can help us separate out these molecules



Key Concepts in the Fuel Cycle – Hydrogen Embrittlement

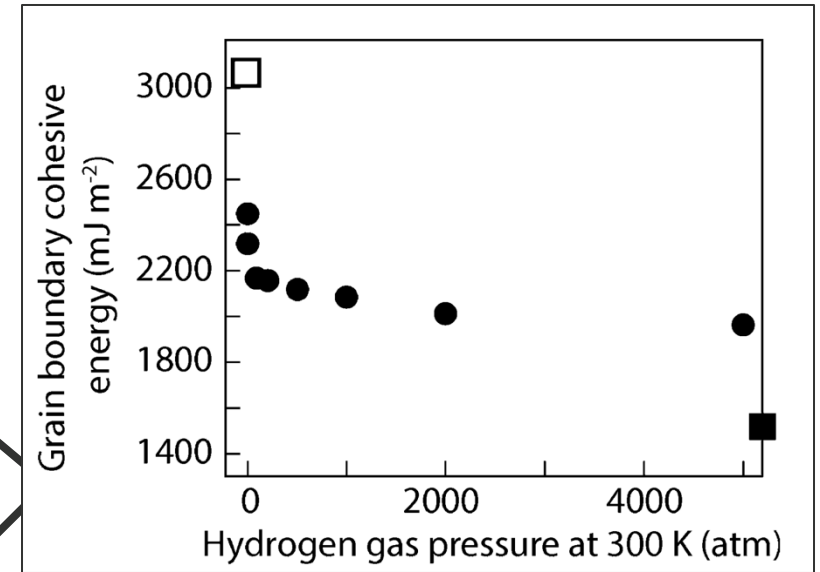
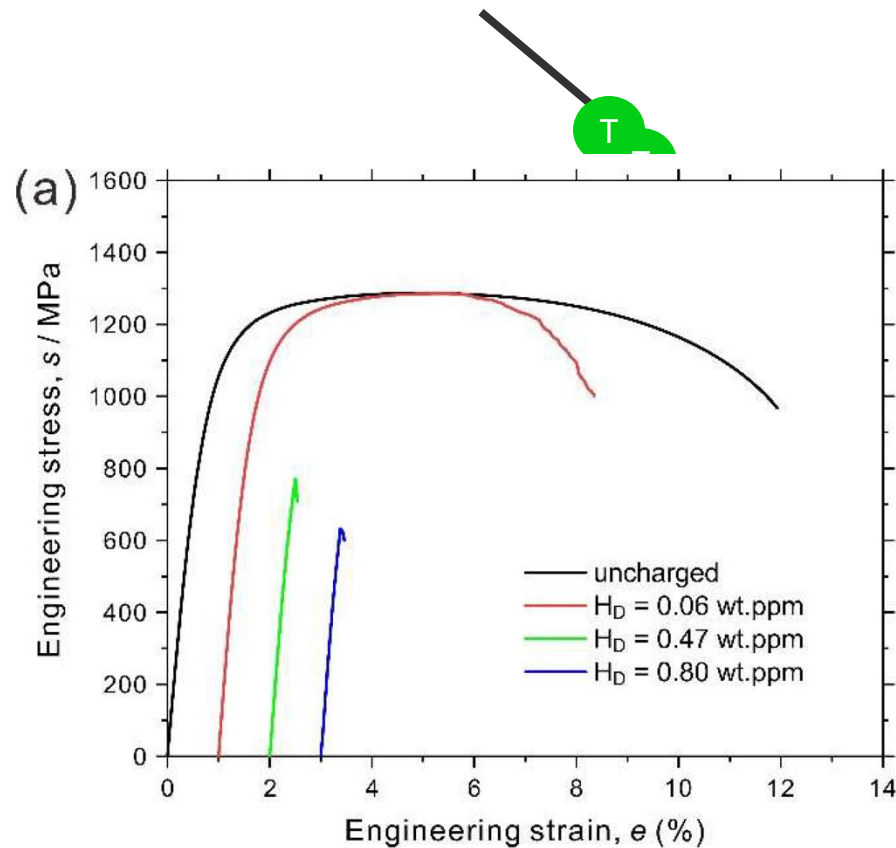


Key Concepts in the Fuel Cycle – Hydrogen Embrittlement



Robertson, I. A. *et al.*, "Hydrogen Embrittlement Understood", *JOM* 67(5), 1036–1046 (2015).

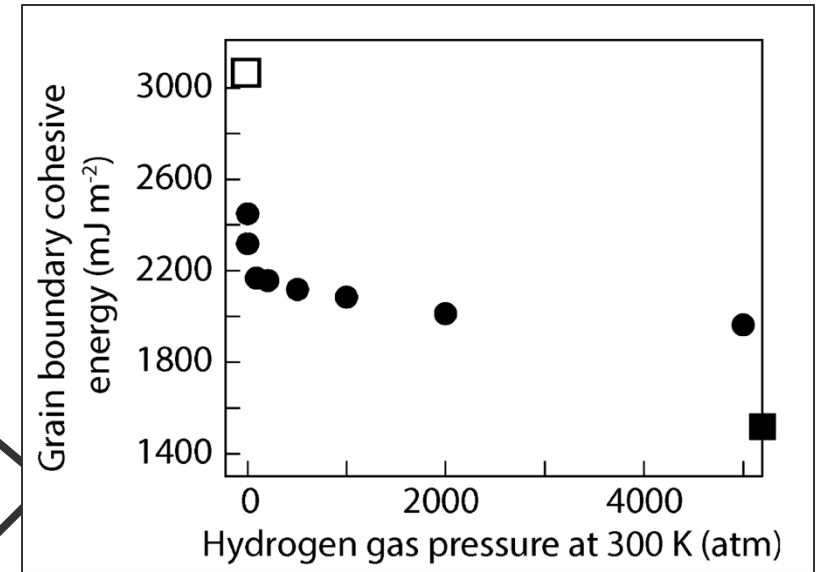
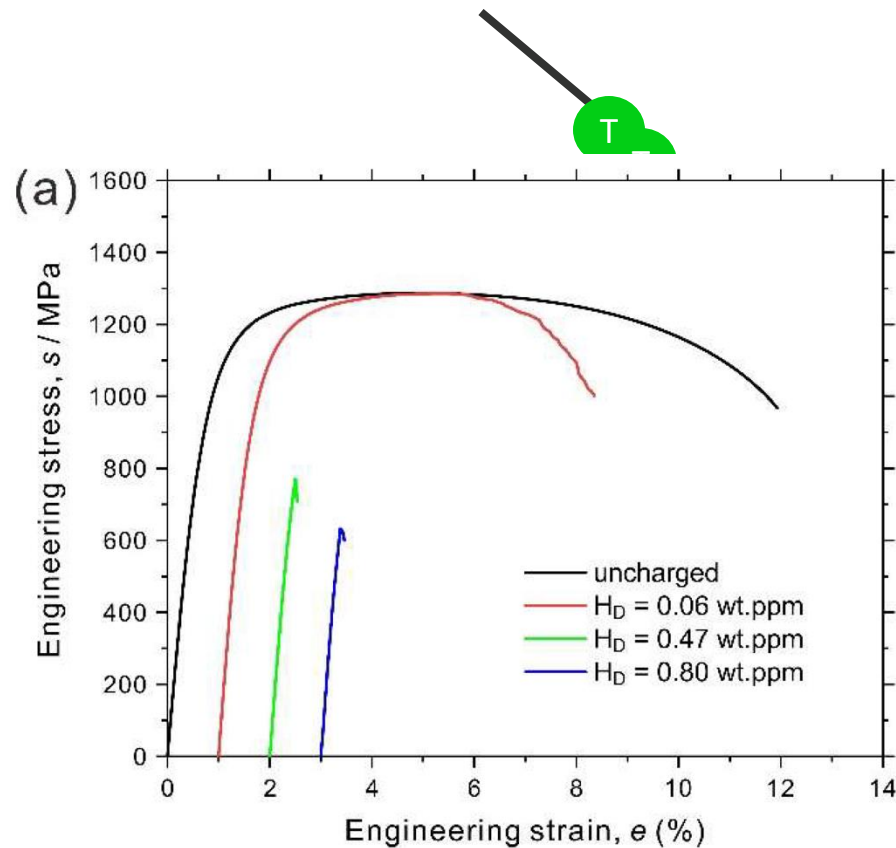
Key Concepts in the Fuel Cycle – Hydrogen Embrittlement



Shibata, A. et al. "Hydrogen-Related Fracture Behavior under Constant Loading Tensile Test in As-Quenched Low-Carbon Martensitic Steel", *Metals* 12(3), 440 (2022).



Key Concepts in the Fuel Cycle – Hydrogen Embrittlement



Mechanism	Description
Hydride formation	Brittle metal hydrides form inside grains
Decohesion	H weakens atomic bonds at grain boundaries
Dislocation interaction	H enhances crack growth at stress concentrators



Key Concepts in the Fuel Cycle – Polymer Challenges

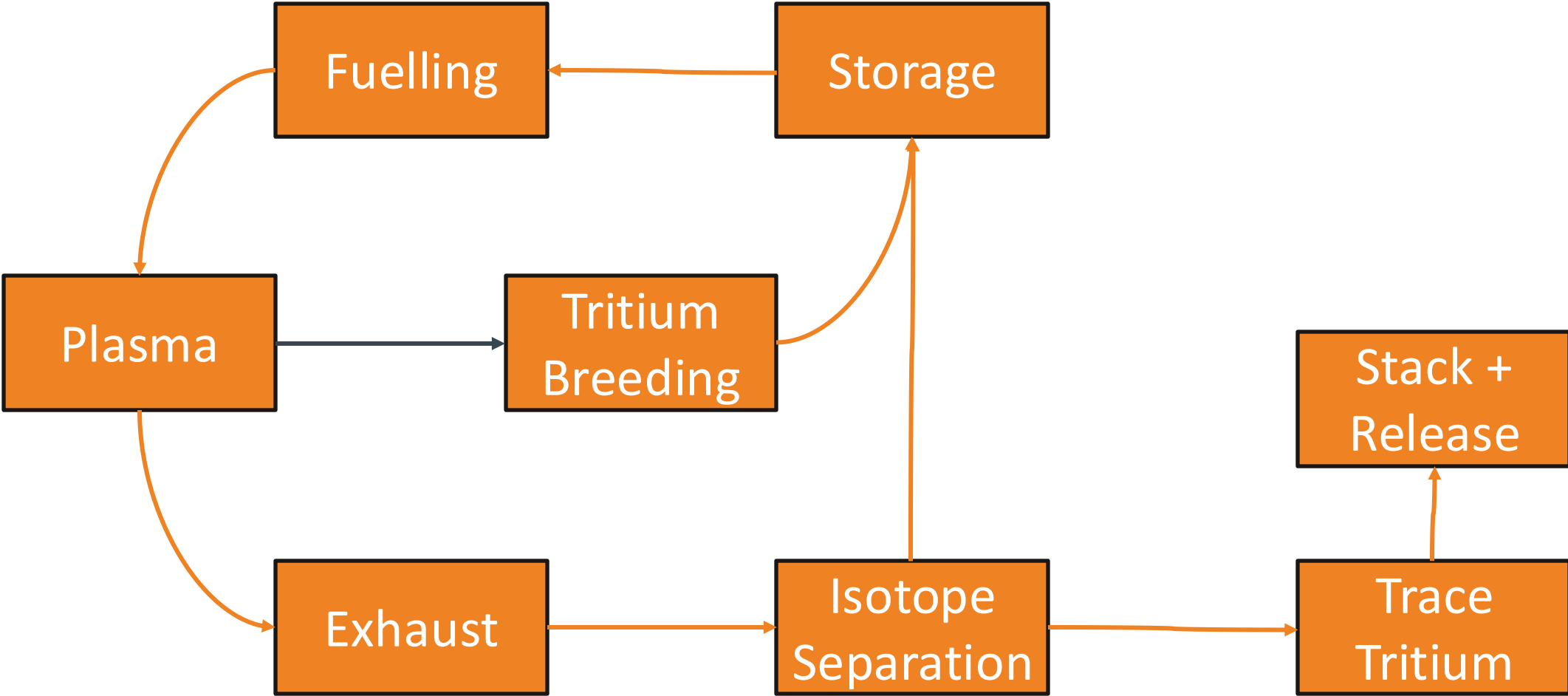
- Polymers (in general) are not preferred for tritium service but are normally needed in seals, gaskets, O-rings, valves, flexible tubing
- Tritium can degrade the material properties causing cracking and leaking through loss of elasticity

Effect	Mechanism
Radiolysis	Tritium beta decay releases ~5.7 keV -> ionizes/breaks chemical bonds
Gas evolution	Formation of H ₂ , CH ₄ , and other small molecules → blistering, pressure build-up
Embrittlement	Chain scission lowers molecular weight → brittle fracture
Crosslinking	Radiation-induced bonding between chains → stiffens material
Swelling and cracking	From accumulated gas or uneven degradation

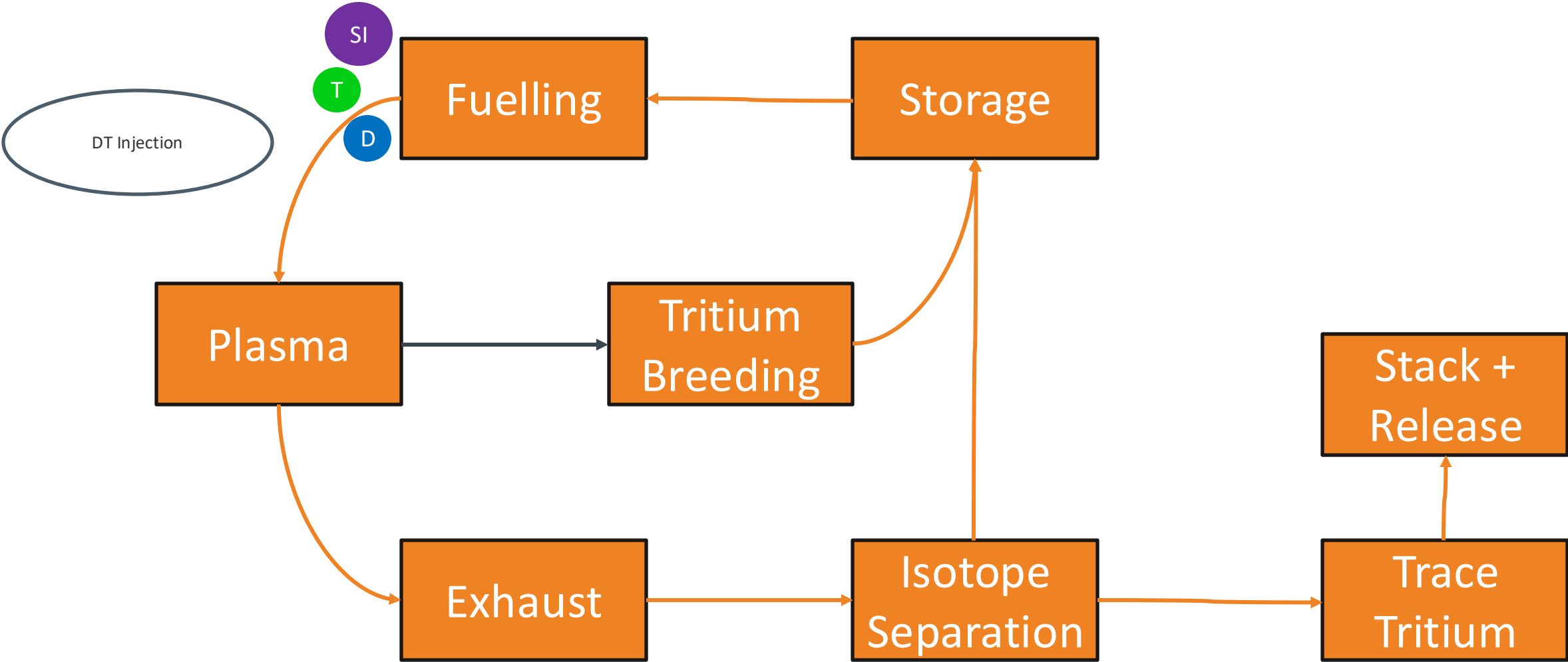


What makes up a fuel cycle?

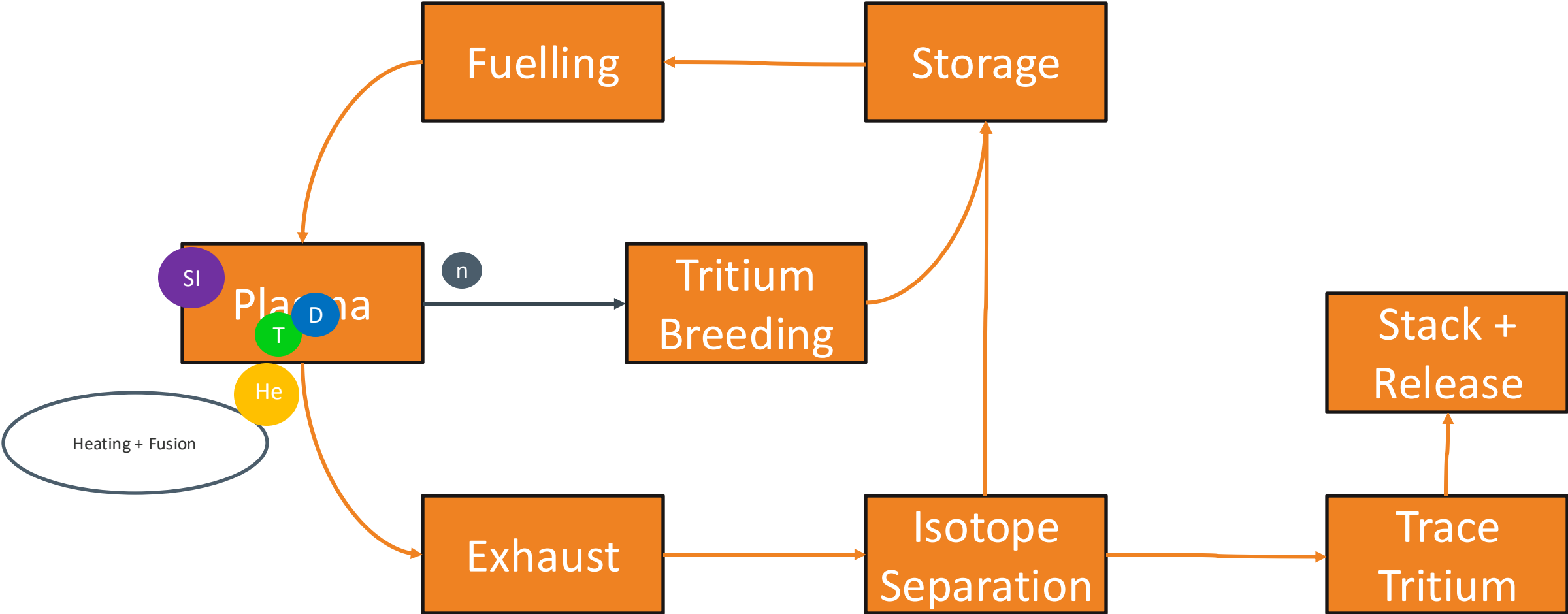
Simplified Block-Flow Diagram



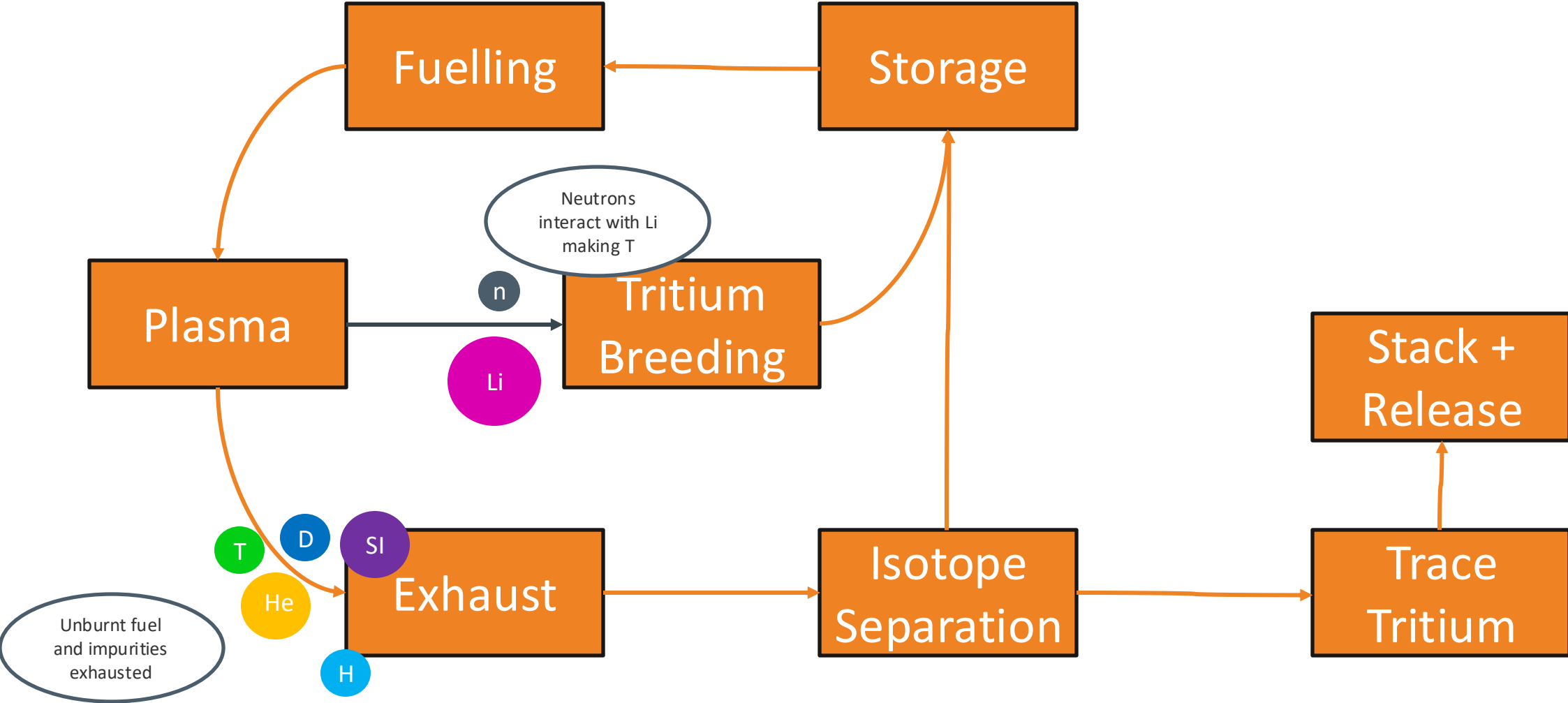
Simplified Block-Flow Diagram



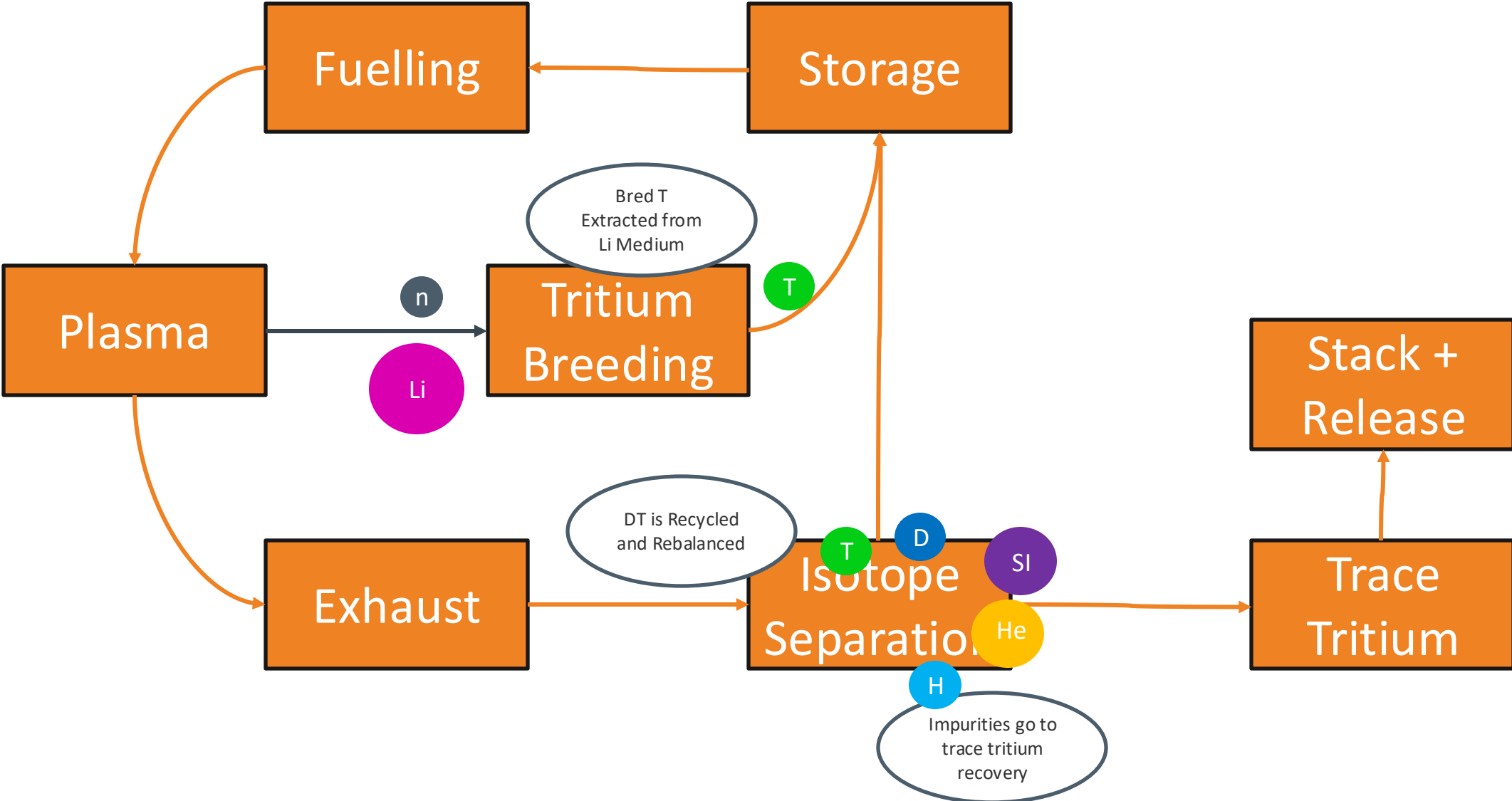
Simplified Block-Flow Diagram



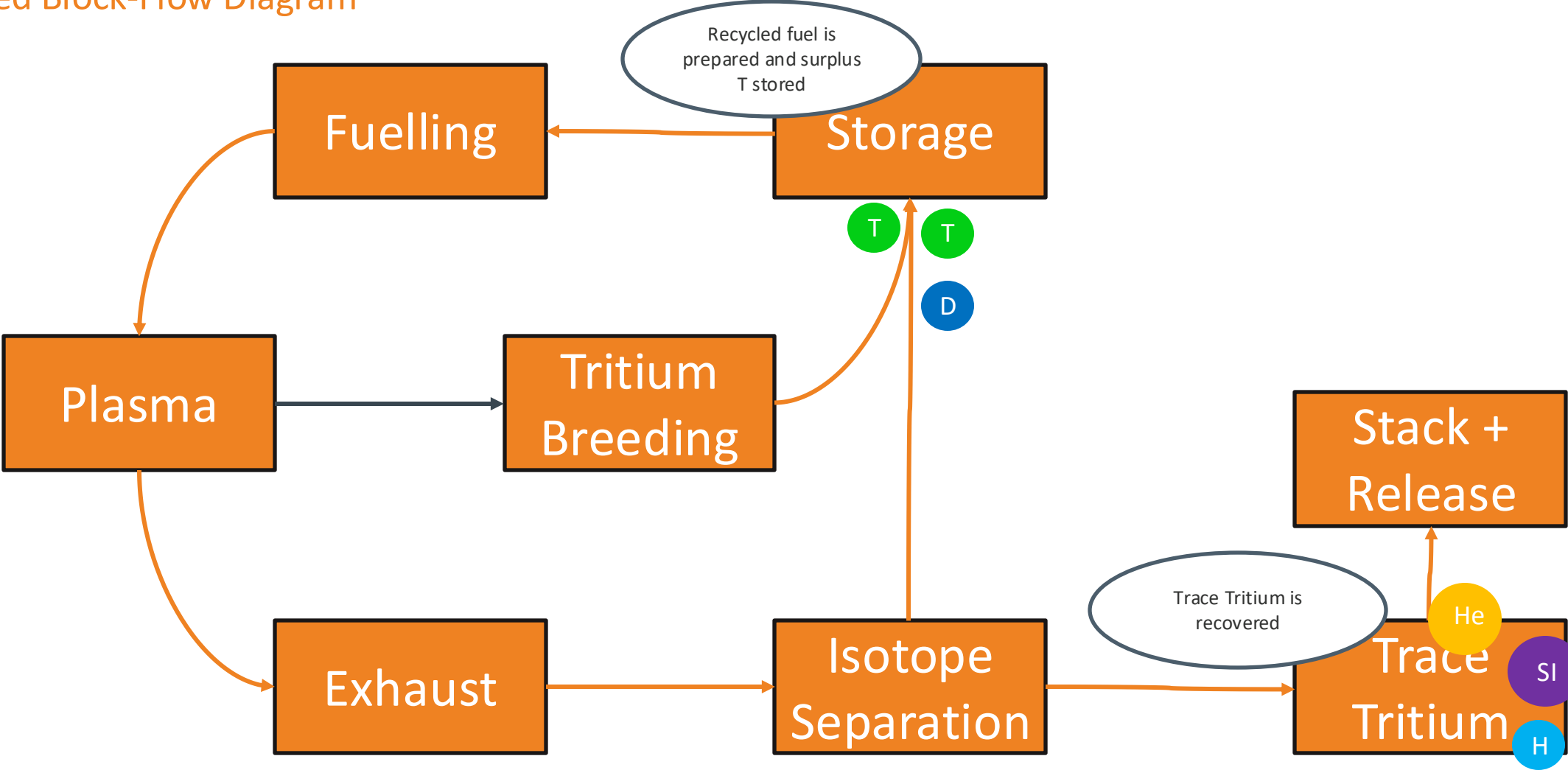
Simplified Block-Flow Diagram



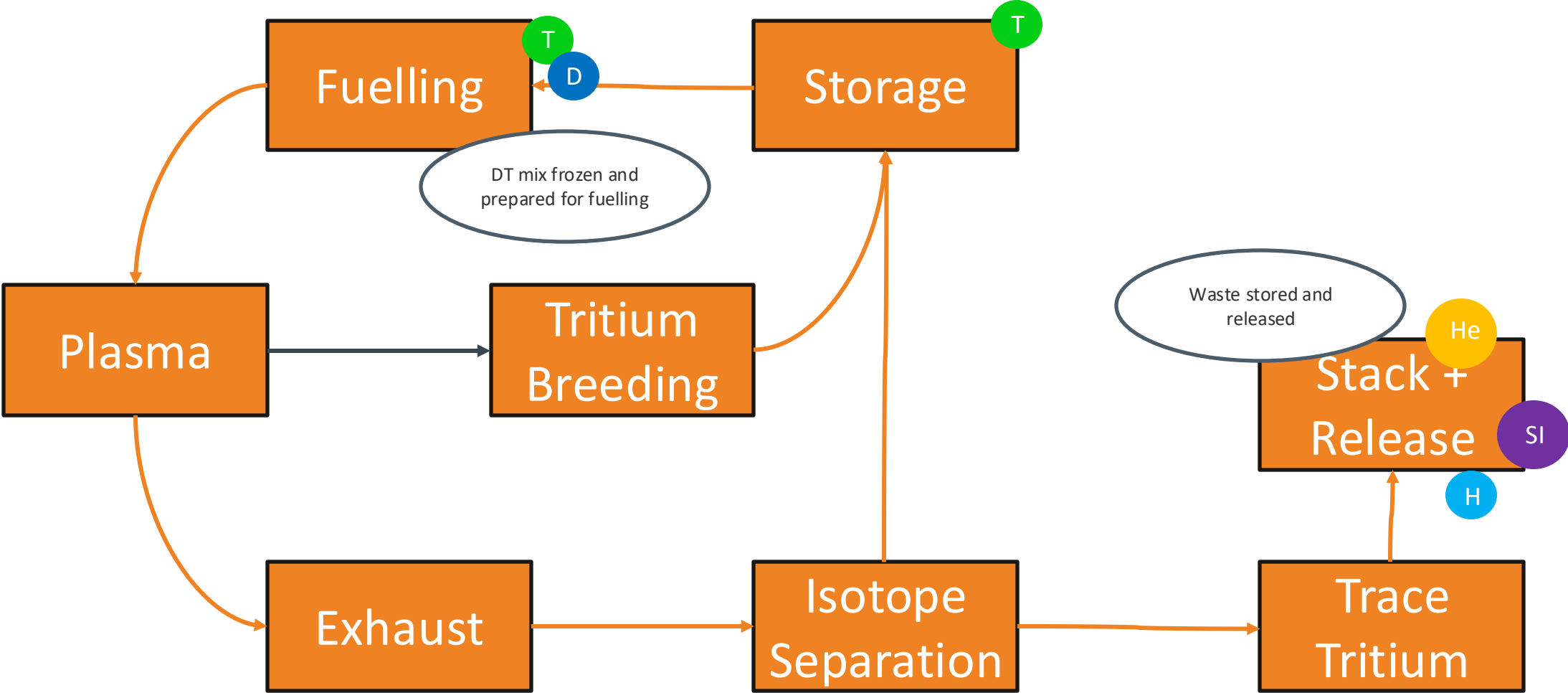
Simplified Block-Flow Diagram



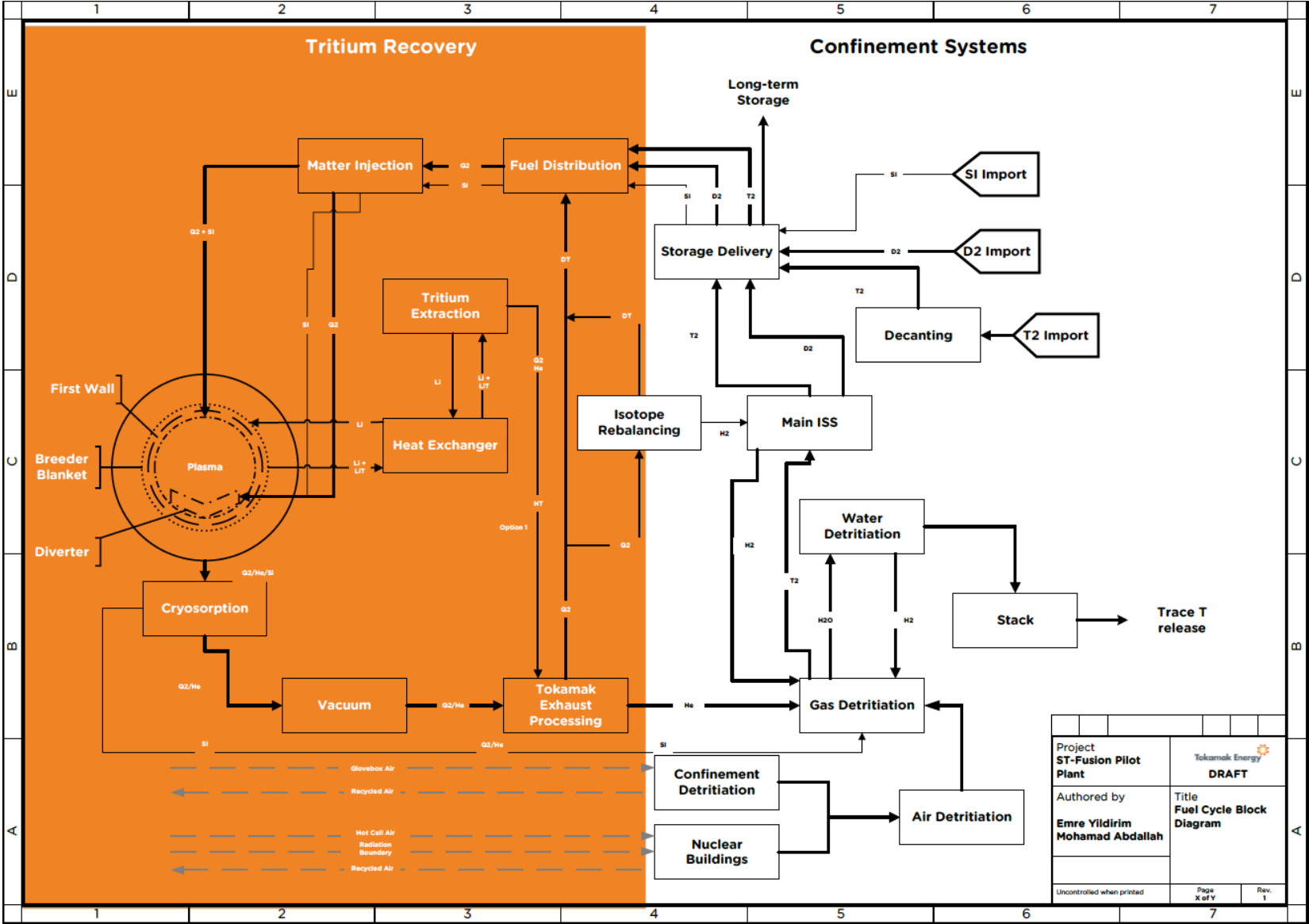
Simplified Block-Flow Diagram



Simplified Block-Flow Diagram

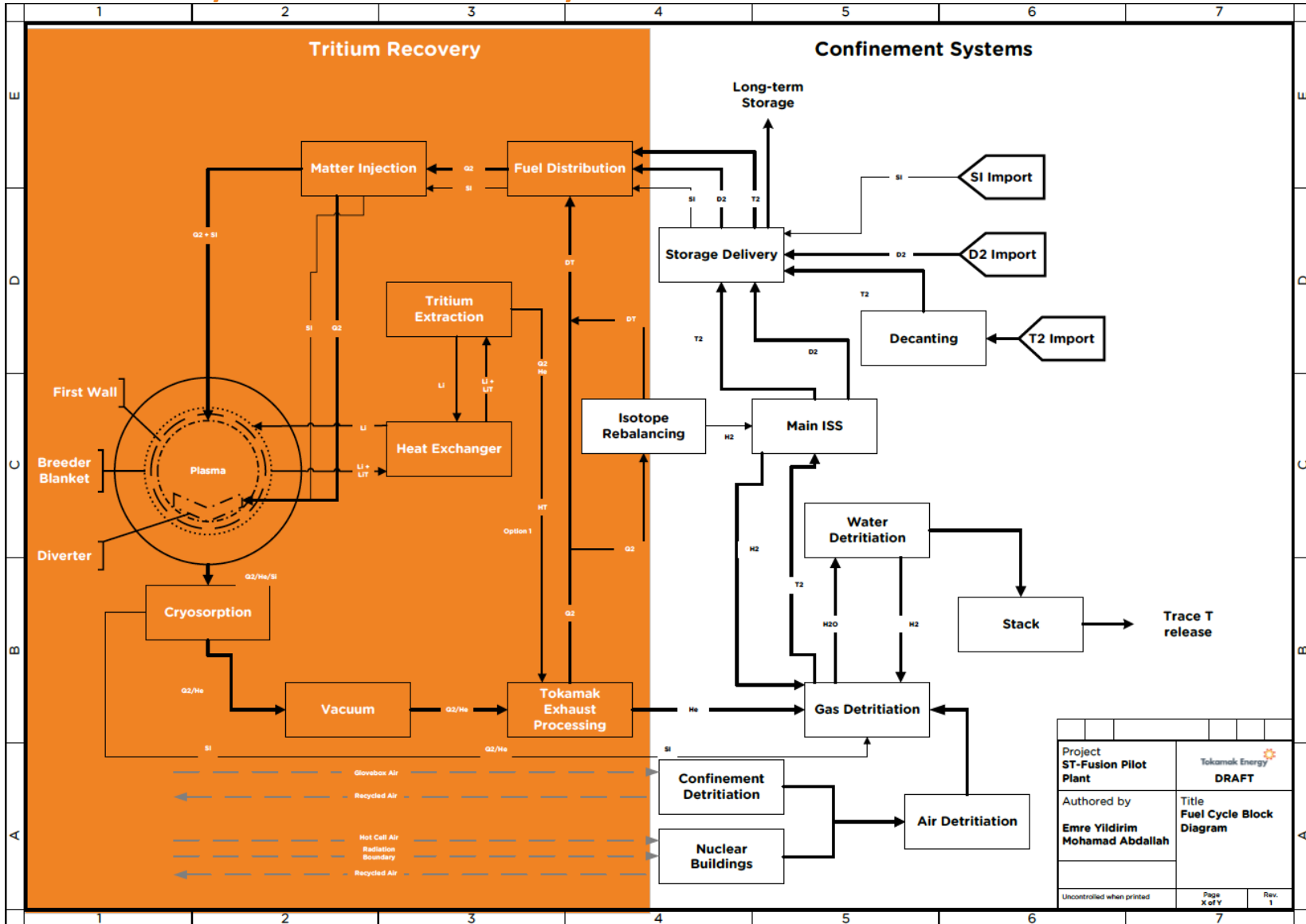


As always – there is always more!



And this is still just a Block Flow Diagram!

As always – there is always more!



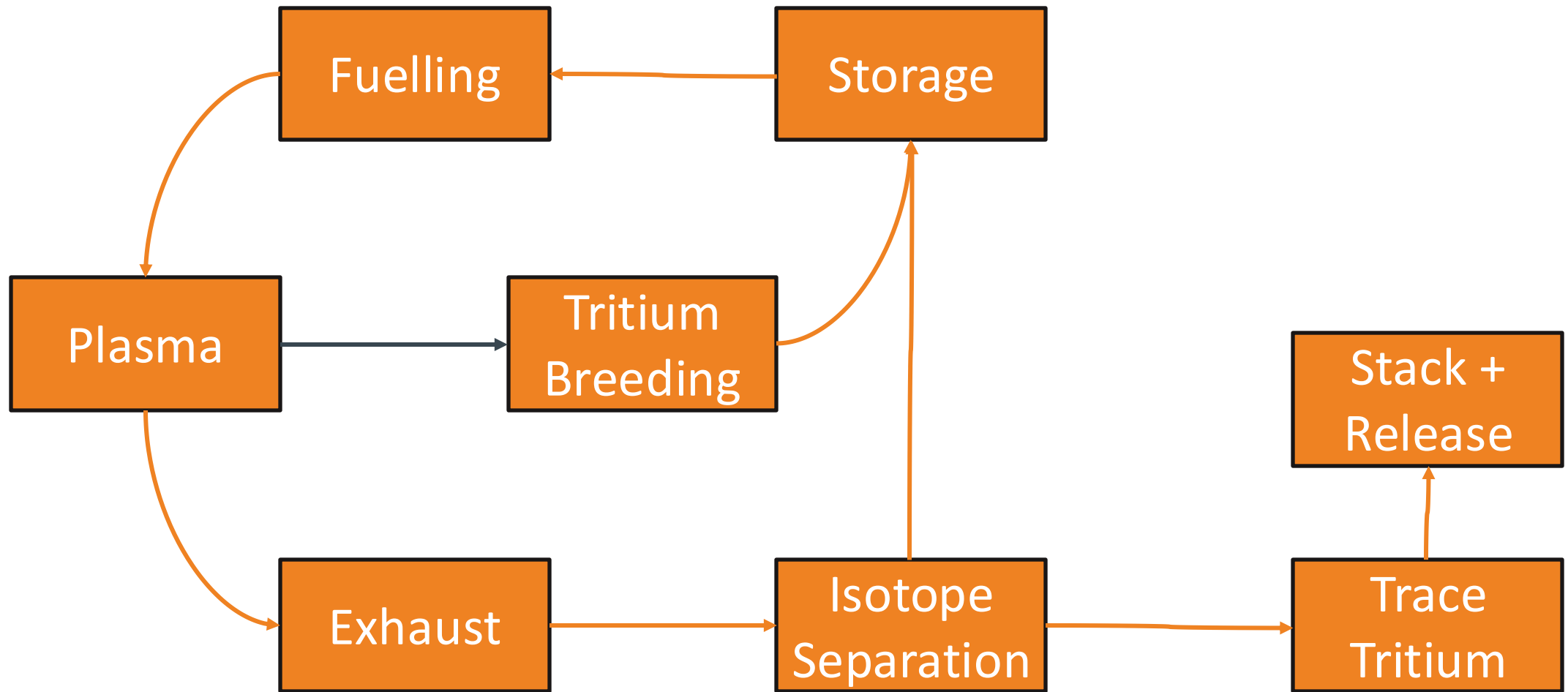
Questions:

The first wall and divertor are on this diagram – why? *Hint: remember your PMI lecture*

Why are we interested in highlighting heat exchangers in the fuel cycle?

Where could the need for air detritiation system come from?





Fuelling

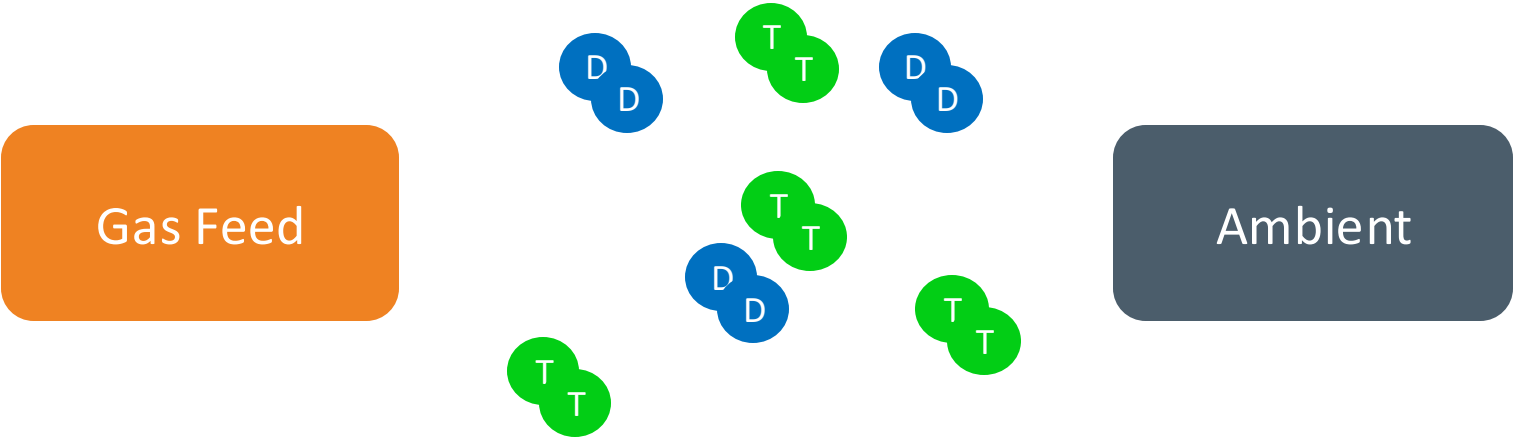
- Fueling controls the fusion power out and allows for the fusion reaction to be maintained
- Fueling must penetrate the core plasma – to allow as much of it to fuse as possible
- Fuel that does not make it to the core adds to the pumped flux out of the tokamak and must be recycled fast

Method	Description	Pros	Cons
Gas Puffing	Injects gas near plasma edge	Simple, robust, low-cost	Low fuelling efficiency (η), most fuel stays at the edge
Pellet Injection	High-speed frozen D/T pellets fired into plasma	Penetrates deeper, improves η and β	Cryo systems required, some mass loss during ablation
Neutral Beam Injection	High-energy neutral atoms (D or T) injected into core	Also heats plasma, can drive current	Requires ultra-pure gas, high tritium losses, complex system



Fuelling

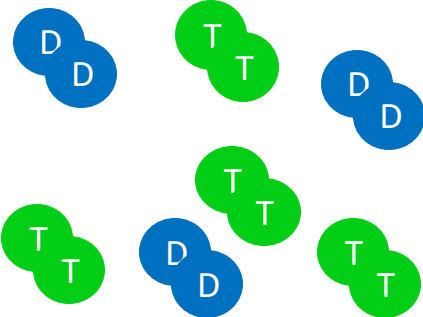
Pellet Injection – Pellet Injectors!



Fuelling

Pellet Injection – Pellet Injectors!

LN2 Precooler

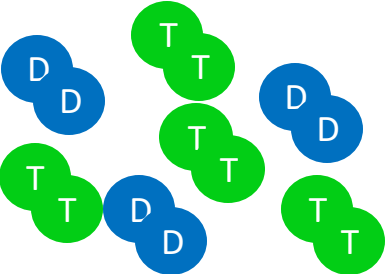


80 K

Fuelling

Pellet Injection – Pellet Injectors!

40 K Precooler

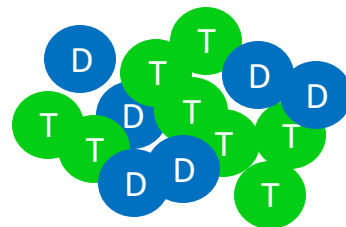


40 K

Fuelling

Pellet Injection – Pellet Injectors!

Liquifier



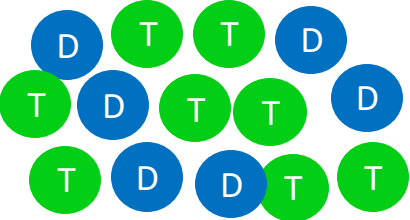
20 K



Fuelling

Pellet Injection – Pellet Injectors!

Extruder



~10 K

Fuelling

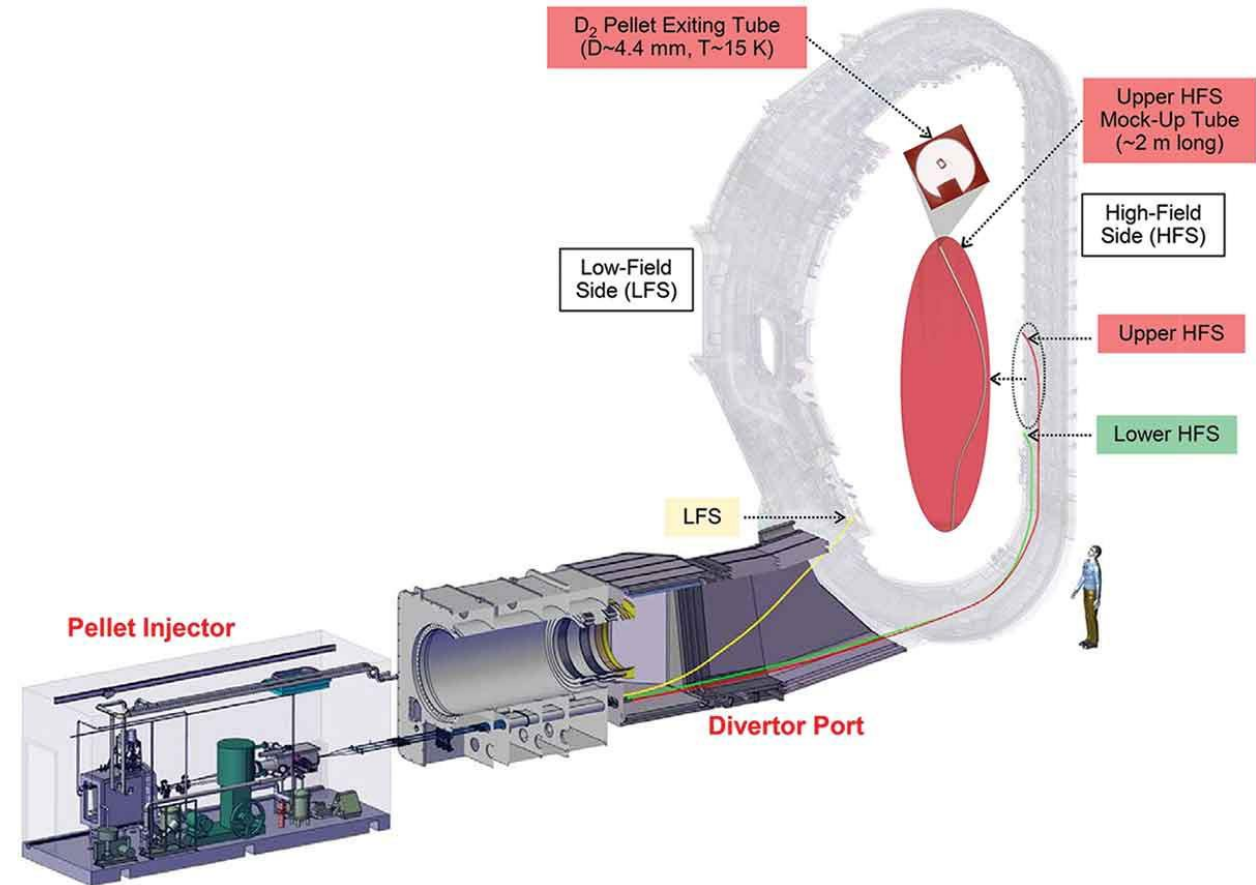


Type	How It Works	Speed
Gas gun	High-pressure gas propels pellet down a barrel (like a cannon)	~300–1000 m/s
Centrifugal launcher	Rotating arm flings pellet into guide tube	Up to ~1000 m/s



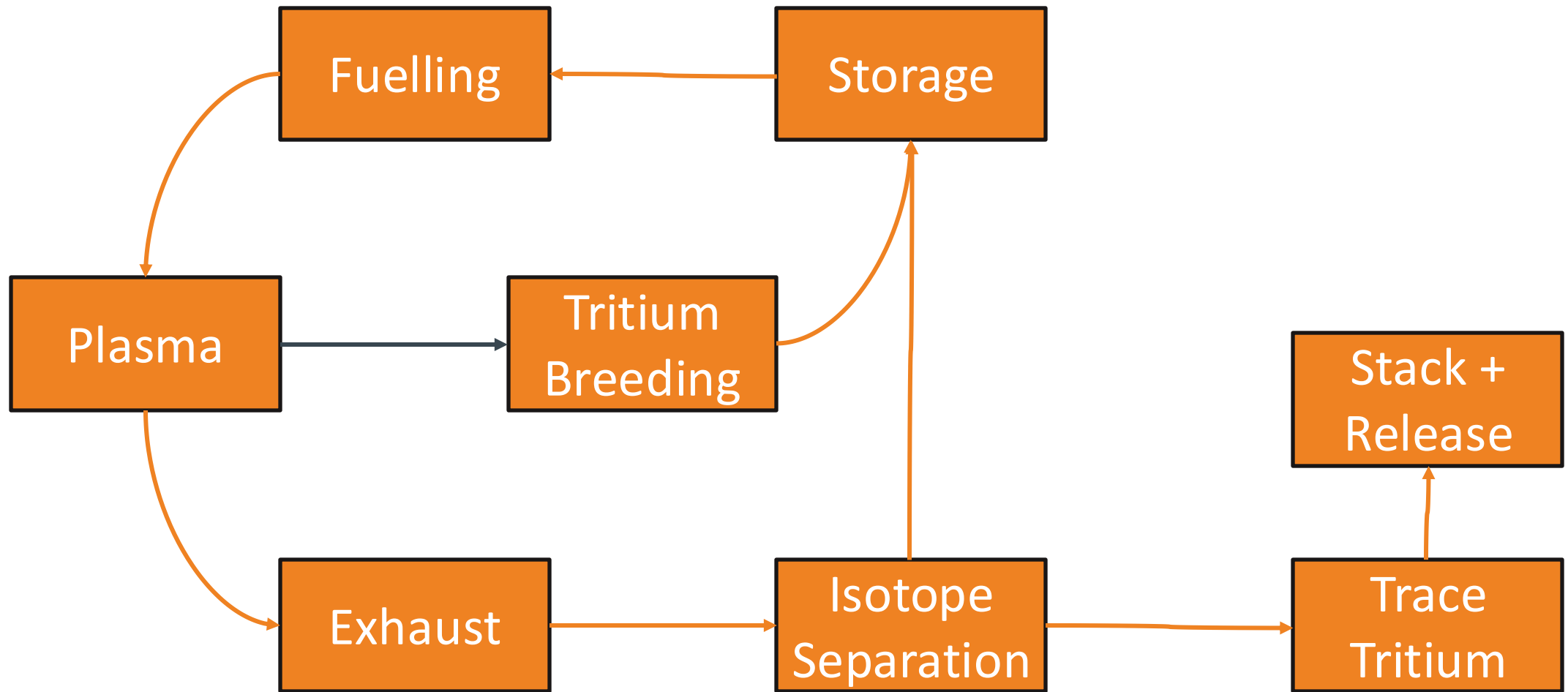
Fuelling

- There are large uncertainties around fuelling efficiencies, with these range from ~10-90 %
- Fuelling is critical for the fuel cycle requirements, but is more critical for the plasma itself
 - Aids control of plasma burn
 - Aids control of density profiles
 - Poor fuelling could also contribute to things like L-H transitions
- The pellet mass, injection speed and geometry are all parameters to tweak
 - Often the ExB drift is the biggest parameter - mean high field side fuelling is a must to improve efficacy and fusion burn



Combs, S. K., & Baylor, L. R. (2018). Pellet-Injector Technology—Brief History and Key Developments in the Last 25 Years. *Fusion Science and Technology*, 73(4), 493–518.

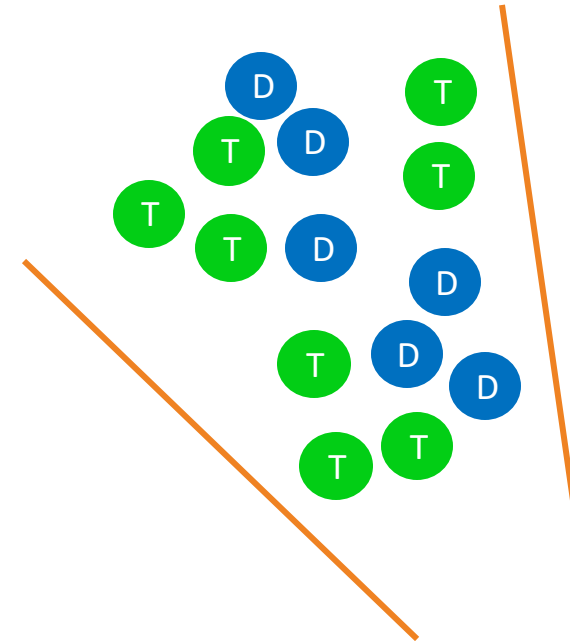




Exhaust

Exhaust – Vacuum Systems

- Vacuum acts to reduce the pressure at inlet
- Molecules move from high pressure to low pressure
- (usually) pressure is then increased after pump [compression ratio]
- This allows us to pump the species that have reached the SOL and been accelerated to the divertor to reduce pressure build up in the chamber

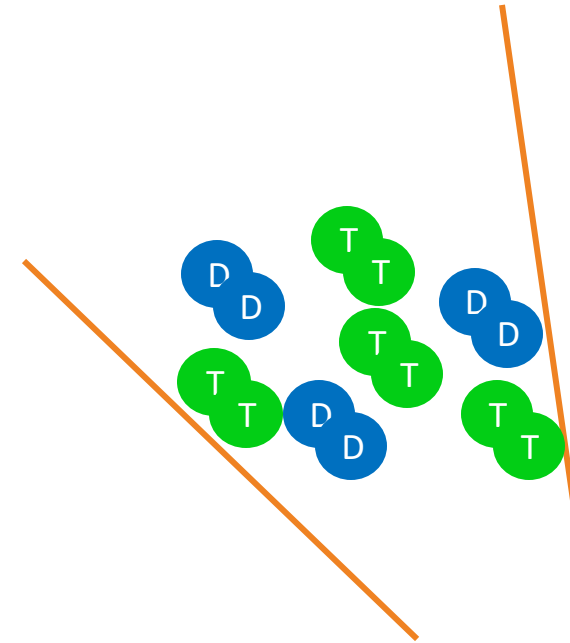


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Particles
neutralize on
divertor plates

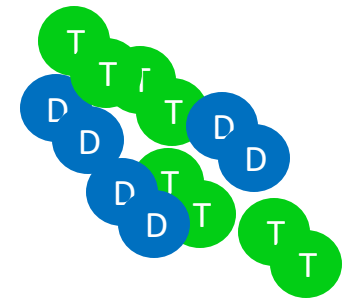


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Vacuum Pumps
pump out
particles



Exhaust

Exhaust Pumping

- Pump out unburnt hydrogen isotopes, He ash and impurities (both intentional such as Ar and residual such as W particles)
- Key to maintain a 'clean' plasma
- Depends on:
 - Conductance [C] - How easily gas flows through a duct or pipe to reach the pump (divertor geometry), in L/s
 - Pumping Speed [S] - The rate at which the pump can remove gas (pump type and design), typically in L/s or m³/s

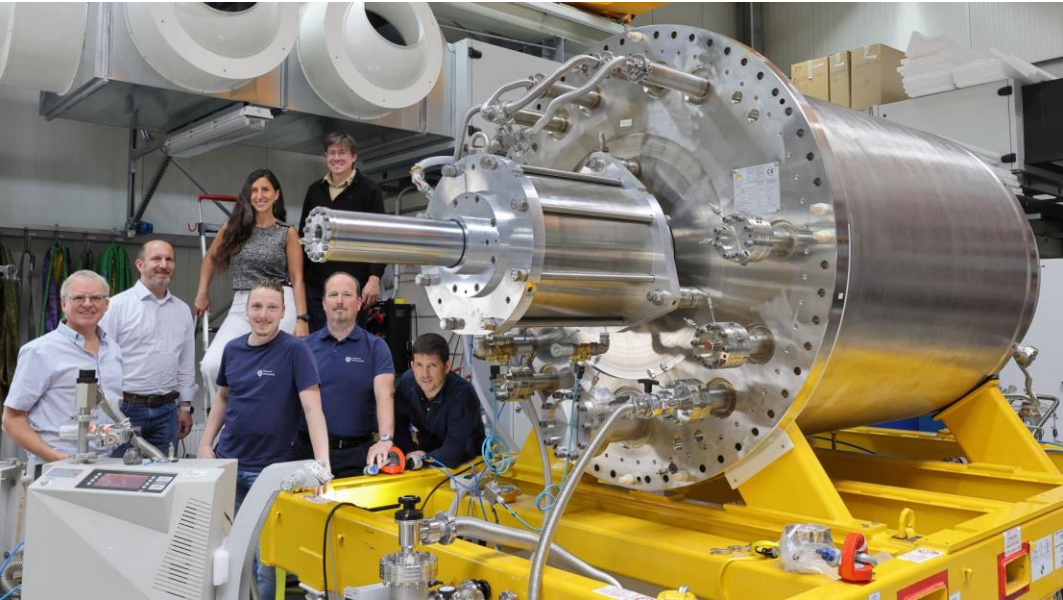
Pump Type	Operating Pressure Range (mbar)	Notes
Roughing Pumps	1e3 to 1e-1	Rotary vane or scroll; used for initial pump-down
Roots Blower	1e2 to 1e-2	Boosts flow between roughing and high-vacuum stages
Turbomolecular Pumps	1e-3 to 1e-9	Requires backing pump; excellent for high vacuum
Cryopumps	1e-3 to 1e-9	Ideal for hydrogen isotopes; based on cryocondensation/sorption
Diffusion Pumps	1e-3 to 1e-7	Oil-based; used less in fusion due to contamination concerns

NB: we use lots of these pumps all around the fuel cycle!



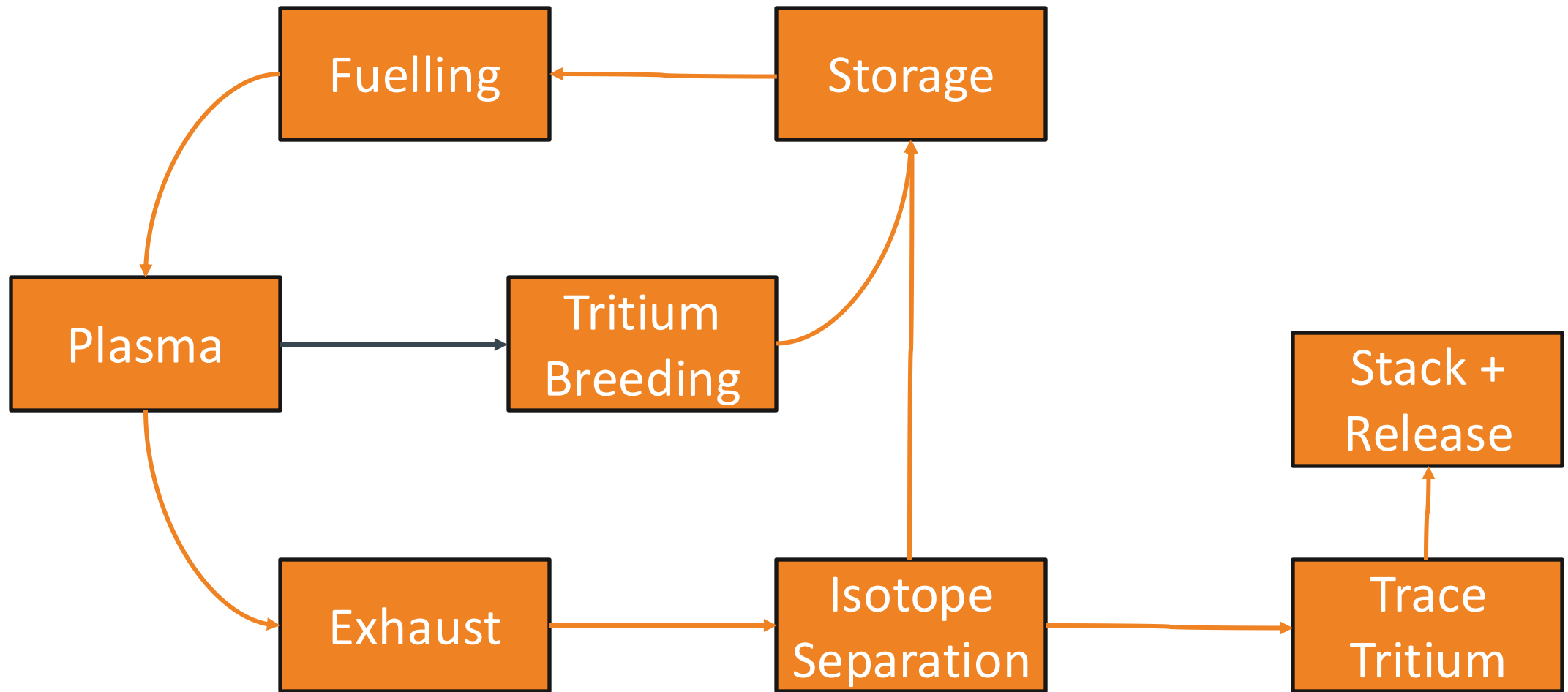
Exhaust

Exhaust Pumping



ITER Website

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Isotope Separation

Separating hydrogen isotopes from other gases
Separate Tritium from other hydrogen isotopes

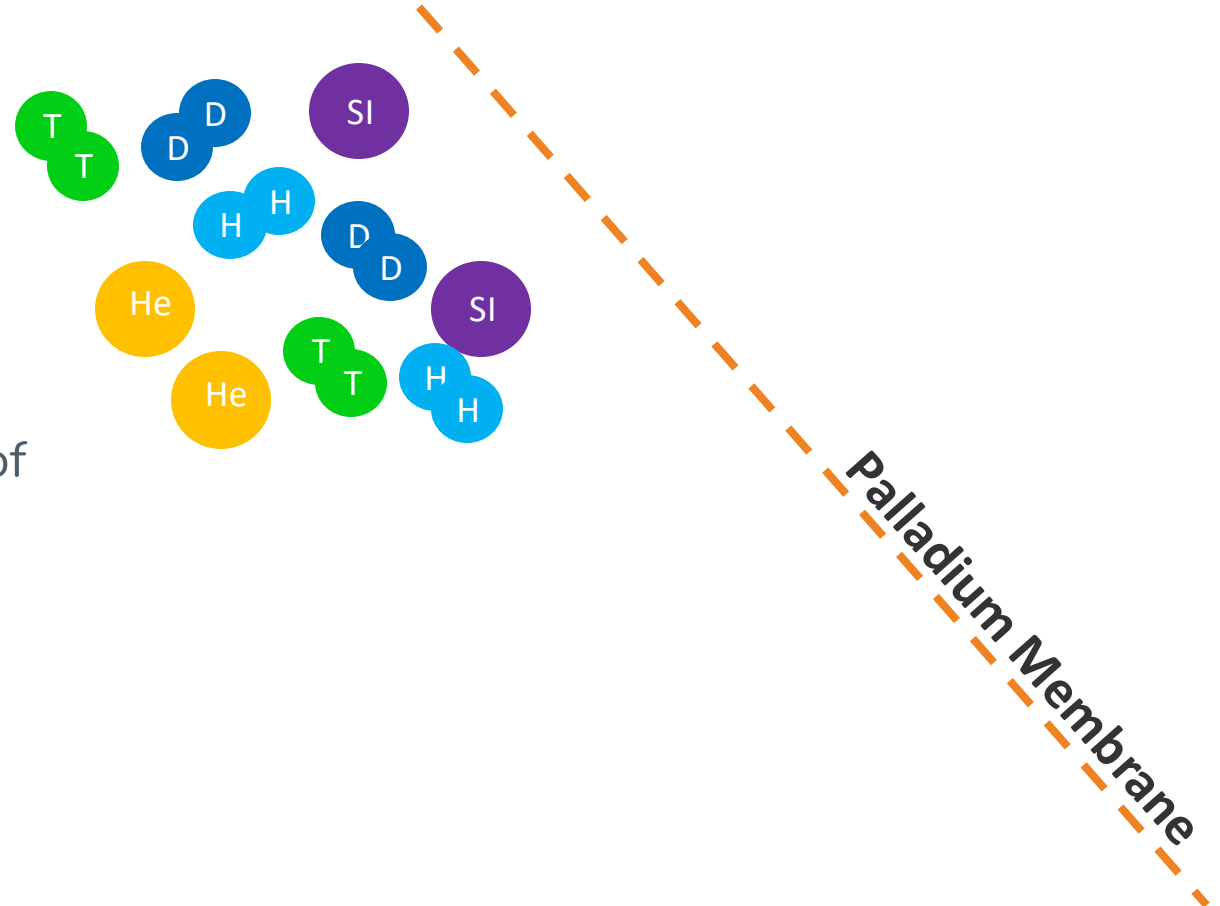


Isotope Separation

Separating hydrogen isotopes from other gases

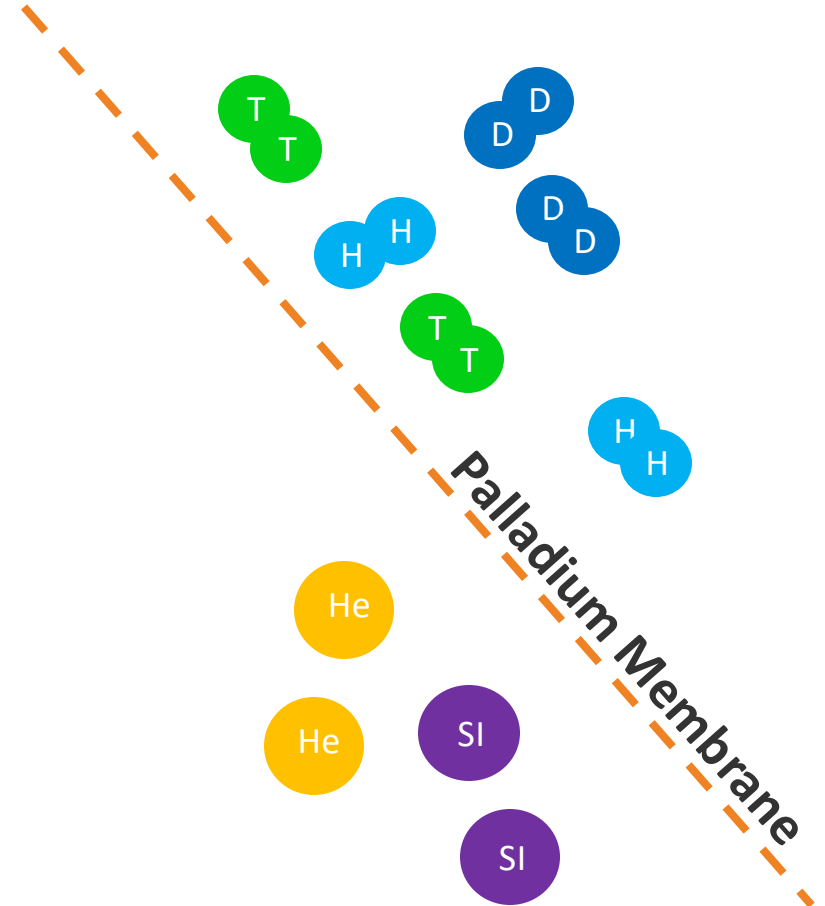
Palladium Diffusers

- Allows H_2 , D_2 and T_2 to pass through selectively blocking other gases
- Operate continuously (limitation is lifetime of the membrane)
- High temperature ($>300^\circ C$), pressure gradient across membrane, thin membrane



Isotope Separation

Separating hydrogen isotopes from other gases



Palladium Diffusers

Challenge	Notes
Embrittlement	Use Pd-Ag (or other Pd based) alloys to resist cracking
Contaminants	CO, H ₂ O can poison membrane
Cost	Pd is rare and expensive
Throughput limits	Temperature and surface area limited

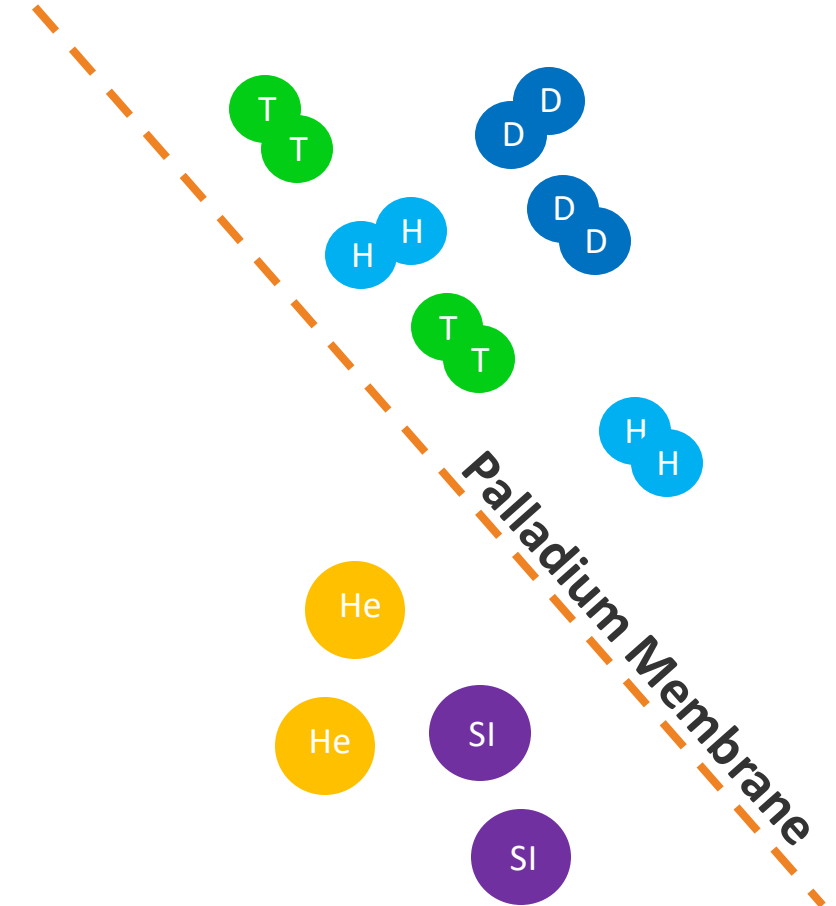


Isotope Separation

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Palladium Diffusers

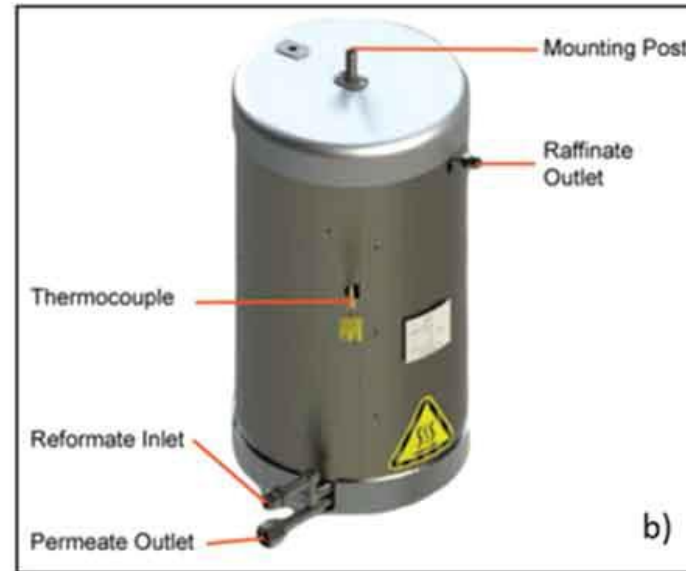
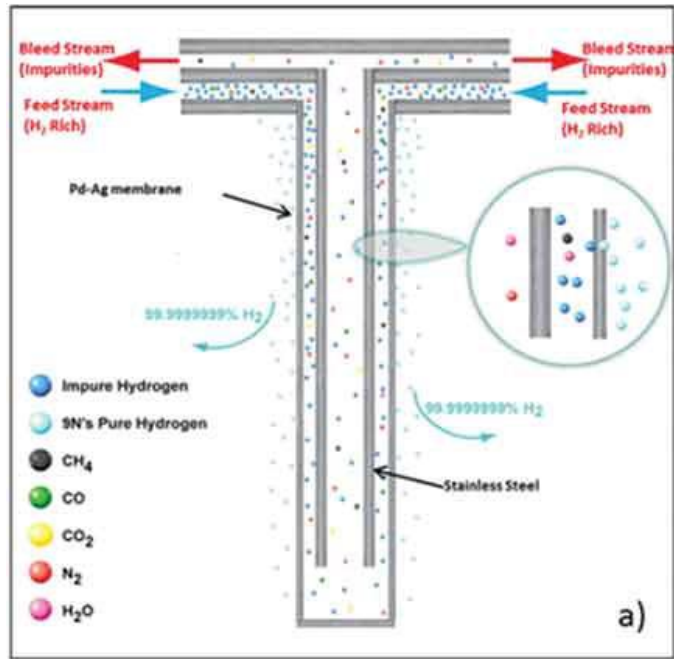
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Isotope Separation

Separating hydrogen isotopes from other gases

Palladium Diffusers - example



Morgan, G. A., Hodge, B. J., & Poore, A. S. (2021). *Performance Testing of a Palladium-Silver Diffuser for Tritium Processing*. *Fusion Science and Technology*, 77(6), 497–506.

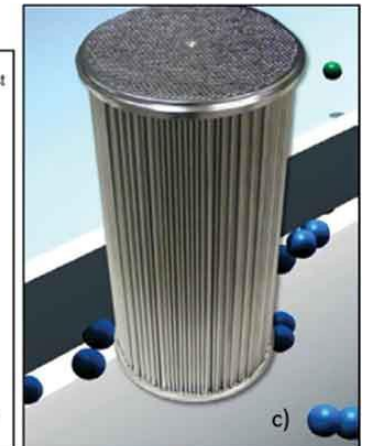
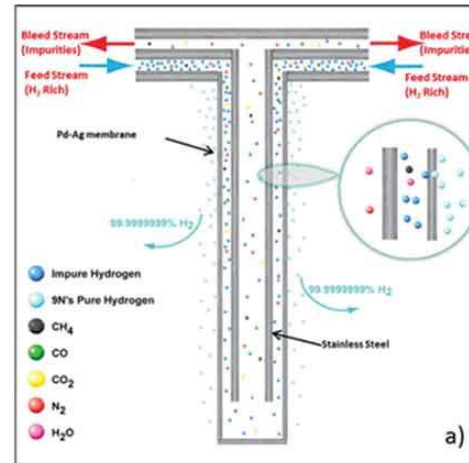


Isotope Separation

Separating hydrogen isotopes from other gases

Palladium Diffusers - example

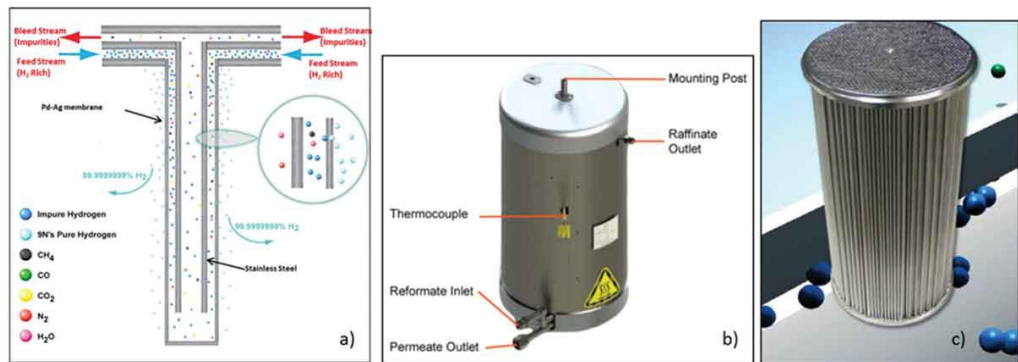
Parameter	Value
Temperature	400°C
Feed pressure	700–1140 Torr (~0.9–1.5 bar)
Feed gas compositions	96%, 50%, 2% H ₂ (balance N ₂)
Flow rates tested	100 – 3000 sccm
Membrane material	Pd-Ag alloy (microchannel tubes)
Membrane surface area	634 cm ²
Tube design	Inside-out flow, ~300 μm wall gap



Morgan, G. A., Hodge, B. J., & Poore, A. S. (2021). *Performance Testing of a Palladium-Silver Diffuser for Tritium Processing*. *Fusion Science and Technology*, 77(6), 497–506.



Isotope Separation



Separating hydrogen isotopes from other gases

Separate Tritium from other hydrogen isotopes

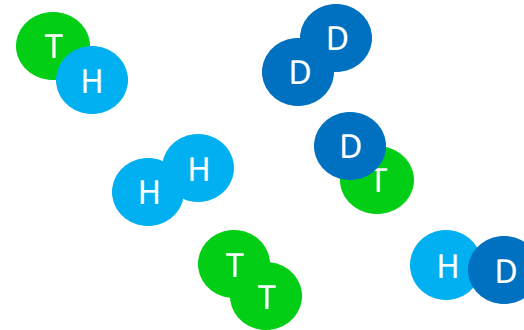


Isotope Separation

Separate Tritium from other hydrogen isotopes

Cryo Distillation Columns

- A cryogenic distillation column separates hydrogen isotopes (H_2 , D_2 , T_2) based on boiling point differences at very low temperatures.
- The top product is gas enriched in lighter isotopes (HH, HT), while the bottom is liquid enriched in tritium
- Multiple stages required



Isotope Separation

Separate Tritium from other hydrogen isotopes

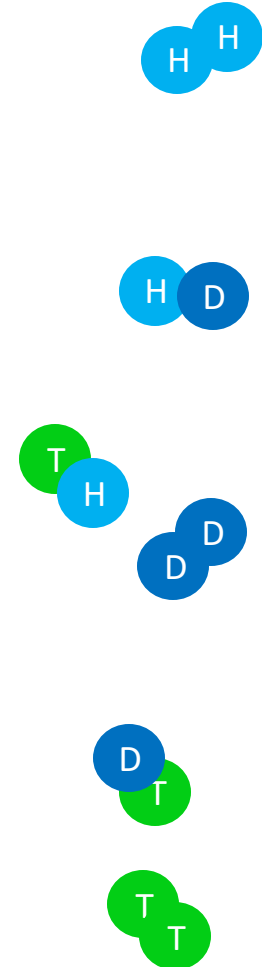
Cryo Distillation Columns

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- Multiple stages required

Full Reflux

Align and stabilize the isotopologue concentration gradient – no product removal

Condenser ~20 K



Reboiler ~25-27 K



Isotope Separation

Separate Tritium from other hydrogen isotopes

Cryo Distillation Columns

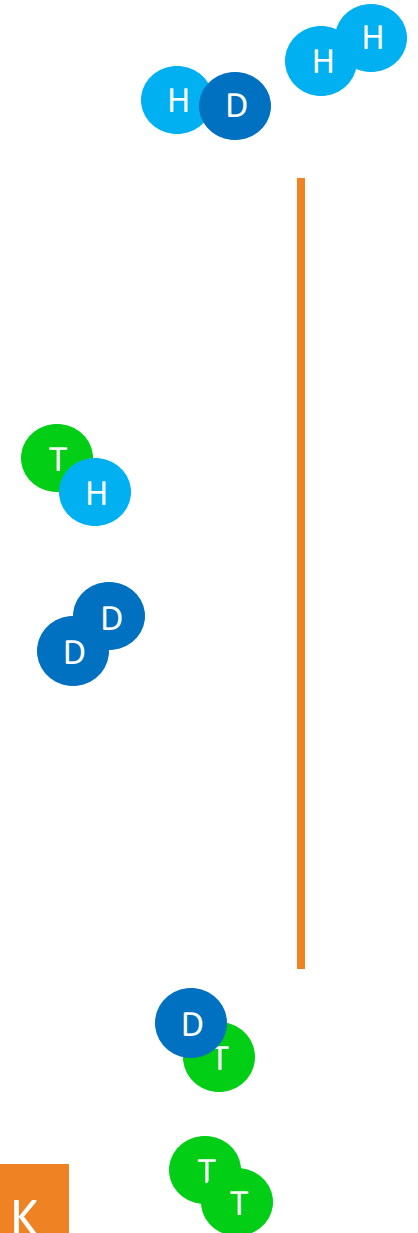
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- The top product is gas enriched in lighter isotopes (HH, HT), while the bottom is liquid enriched in tritium
- Multiple stages required

Partial Reflux

Start tapping off the top and bottom products (a fraction of the condensate reused as per reflux ratio)

Reboiler ~25-27 K

Condenser ~20 K



Isotope Separation

Pd Diffusers

Compound Cryopumps

Metal Foil Pumps

Getter Beds

Separating hydrogen isotopes from other gases
Separate Tritium from other hydrogen isotopes

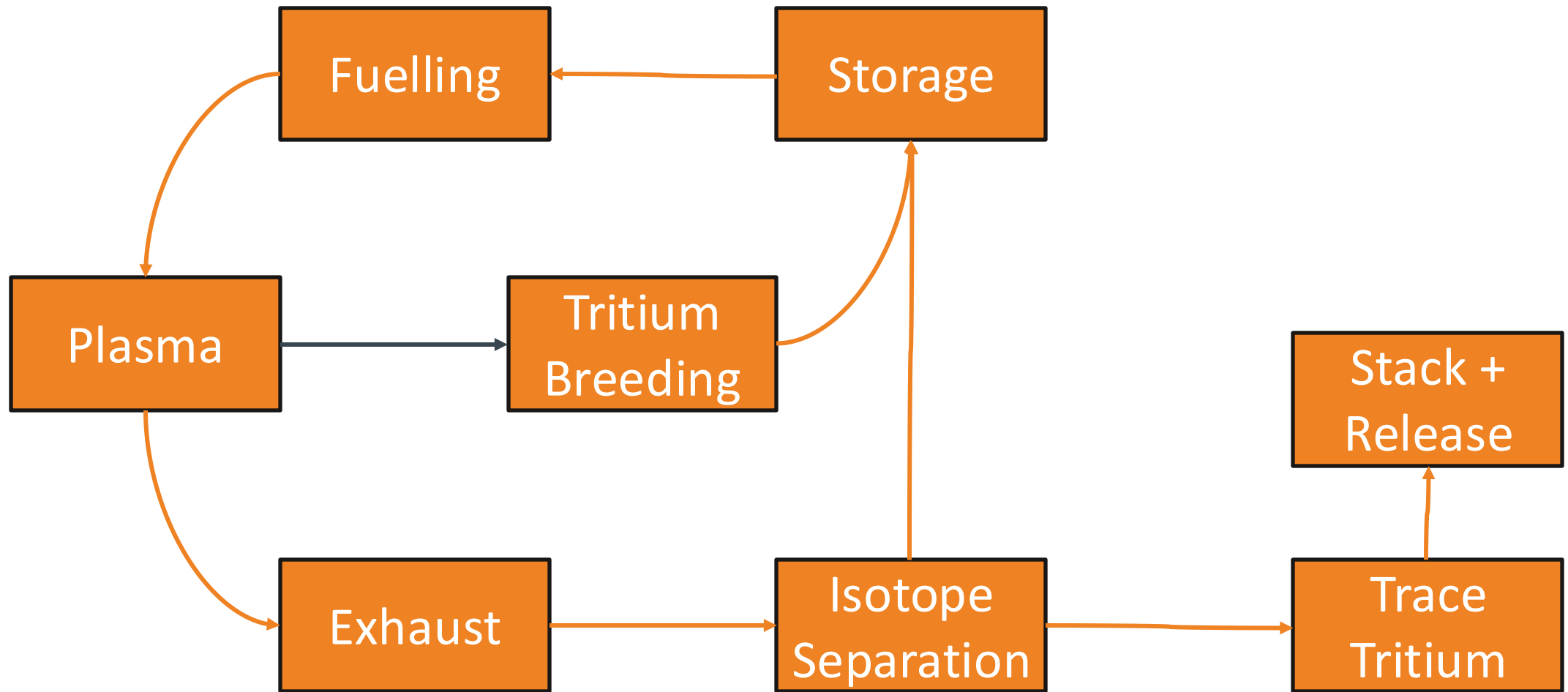
TCAP

PSA

Cryo distillation

Gas Chromatography



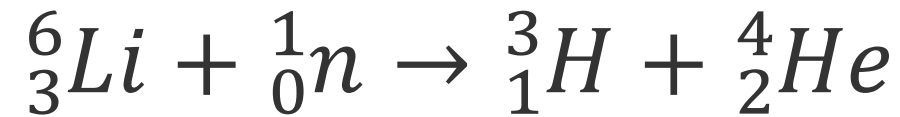


Tritium Breeding

Breed Tritium from Lithium Compounds (more in Lane's Talk!)



High energy
neutrons



Thermal
neutrons



Tritium Breeding

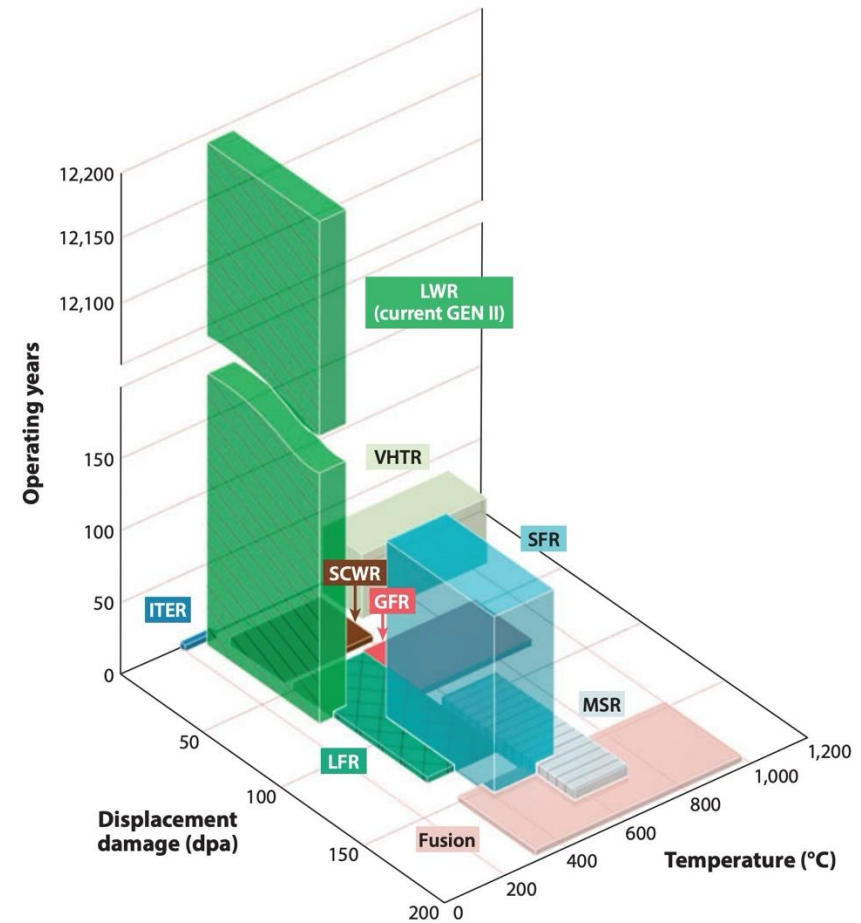
Breed Tritium from Lithium Compounds (more in Lane's Talk!)

Liquid

Solid

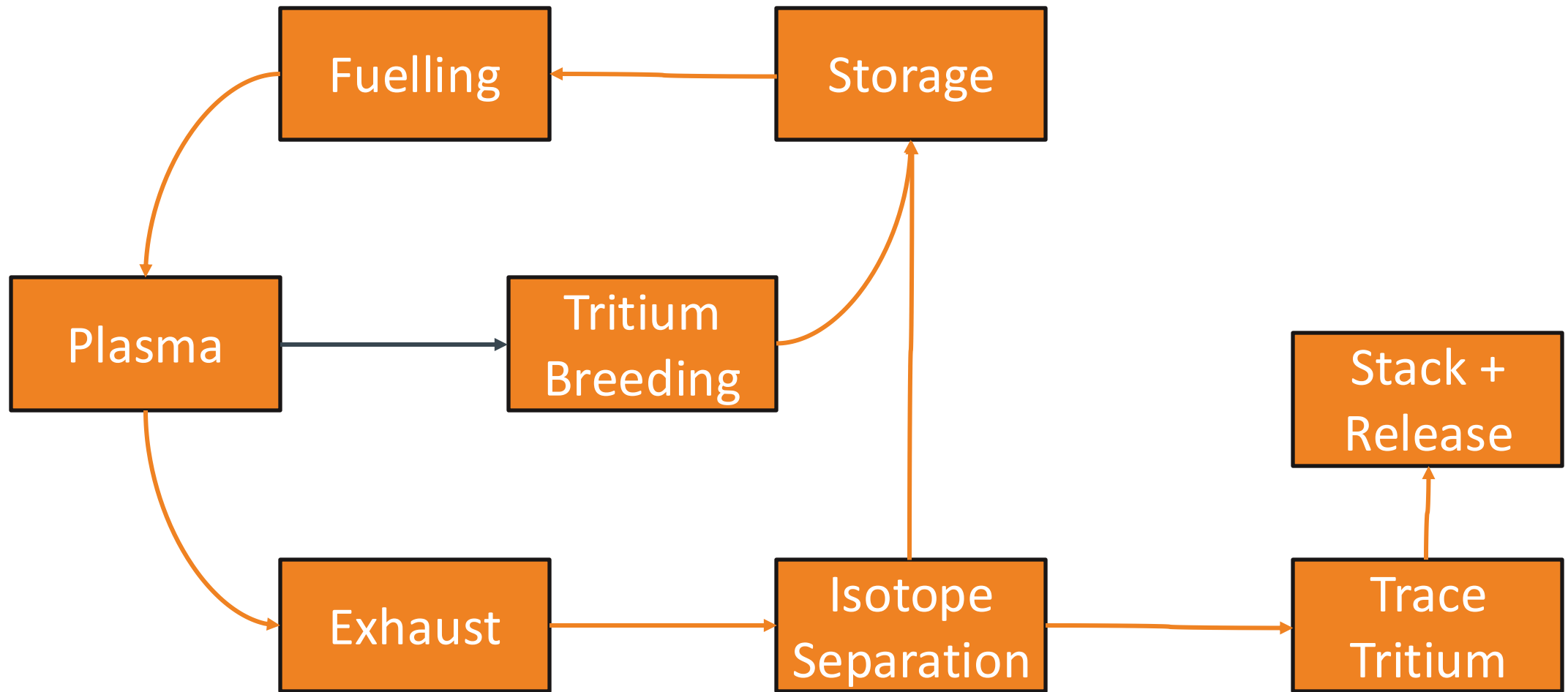
- Breeder: Li, Li/Pb, Molten Salt
- Coolant: Breeder, Water, Helium, Molten Salt
- Multiplier: Li-7, Be, Pb

- Breeder: Lithium Oxides, Ternary Oxides, Other
- Coolant: Water, Helium
- Multiplier: Be, Pb



Zinkle & Snead (2014) *Designing Radiation Resistance in Materials for Fusion Energy*, *Annu. Rev. Mater. Res.*, 44, 241–267



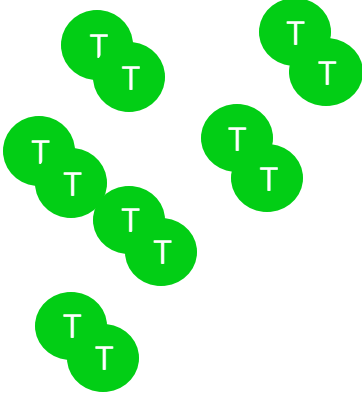


Storage

Metal Hydrides

- Beds that absorb and release hydrogen isotopes (T₂, D₂, H₂) by forming solid metal hydrides.
- Act like chemical sponges for tritium.
- Reversible storage: heat to release, cool to store.
- High volumetric density compared to gas cylinders.
- Passive safety: tritium is chemically bound in solid phase.

Material	Hydride
Uranium	UH ₃
LaNi ₅	LaNi ₅ H _x
ZrCo	ZrCoH _x
Ti, FeTi, Pd	various



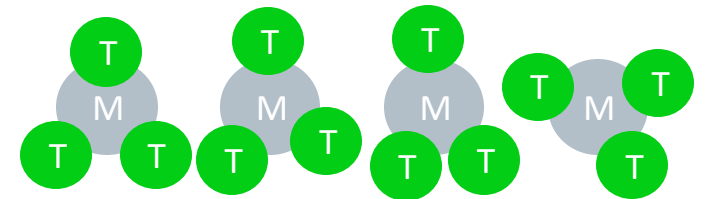
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Cool

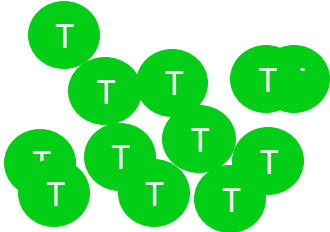


Storage

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Heat



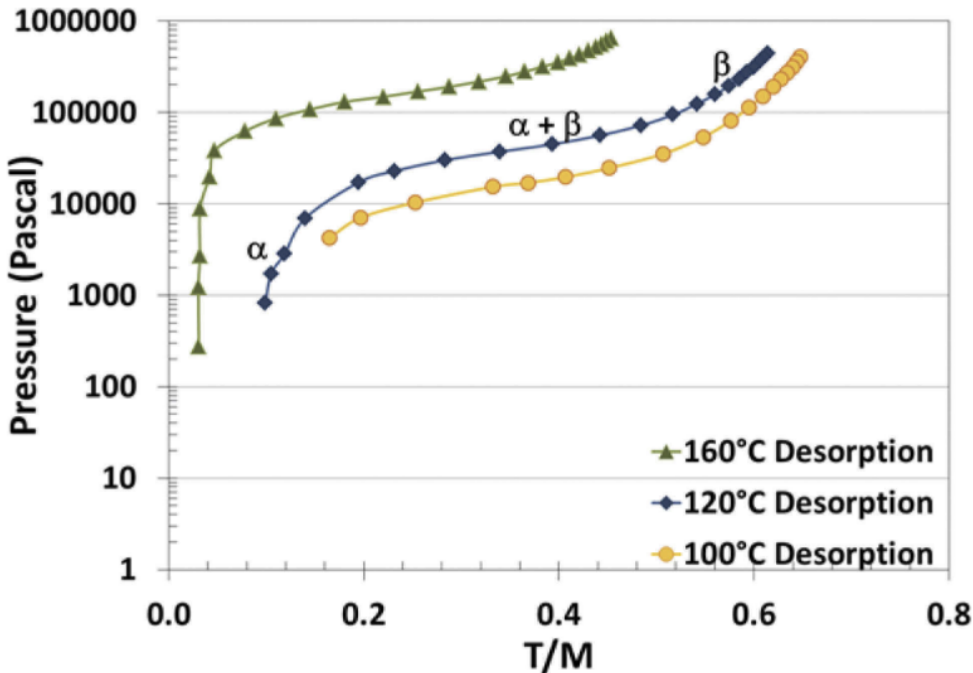
Storage

Metal Hydrides

- **α -phase (left side)**
Hydrogen is dissolved in the metal — easy in, easy out. Pressure rises quickly.
- **$\alpha + \beta$ Plateau (middle)**
Metal and hydride coexist. Large amounts of hydrogen are absorbed or released at nearly constant pressure — ideal for storage and delivery.
- **β -phase (right side)**
The hydride is full. Further hydrogen uptake requires much higher pressure — less efficient for operation.

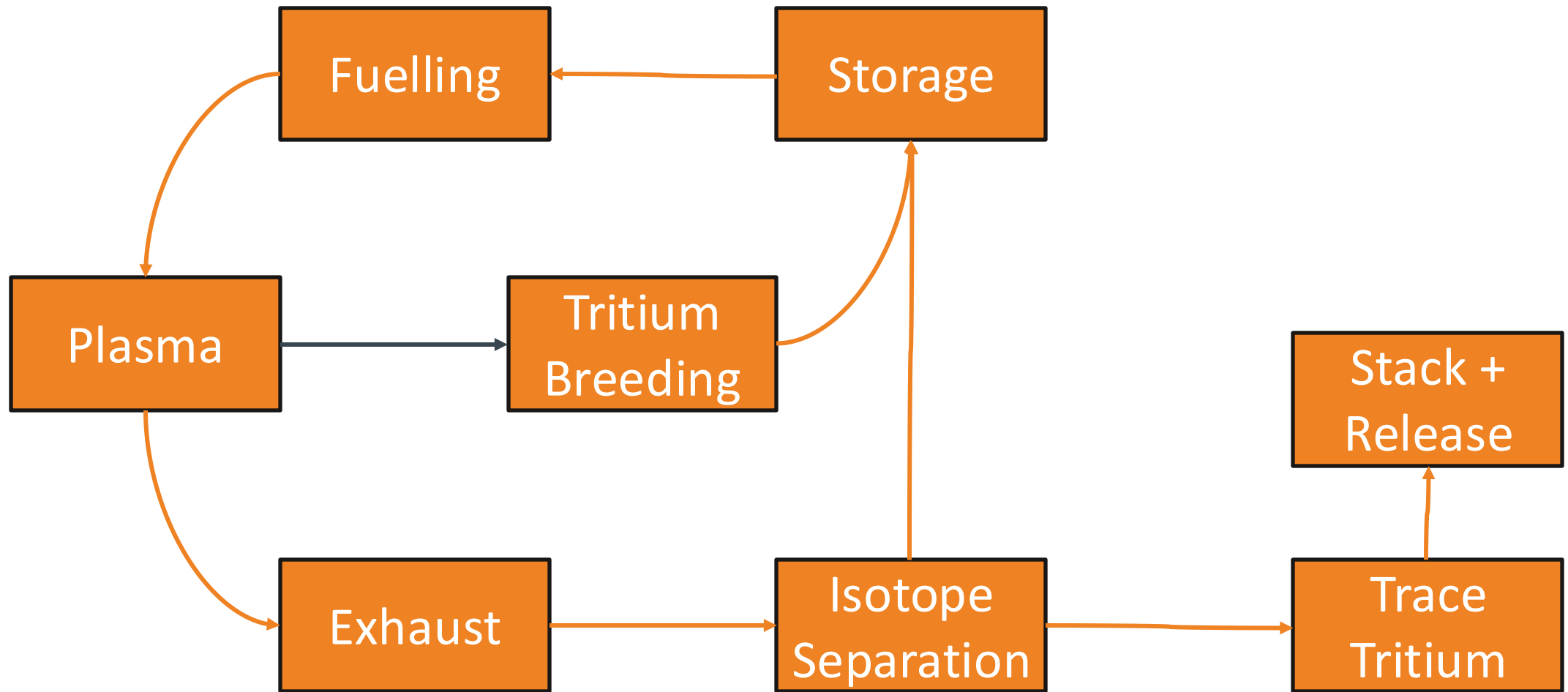


ITER Website – Uranium Beds



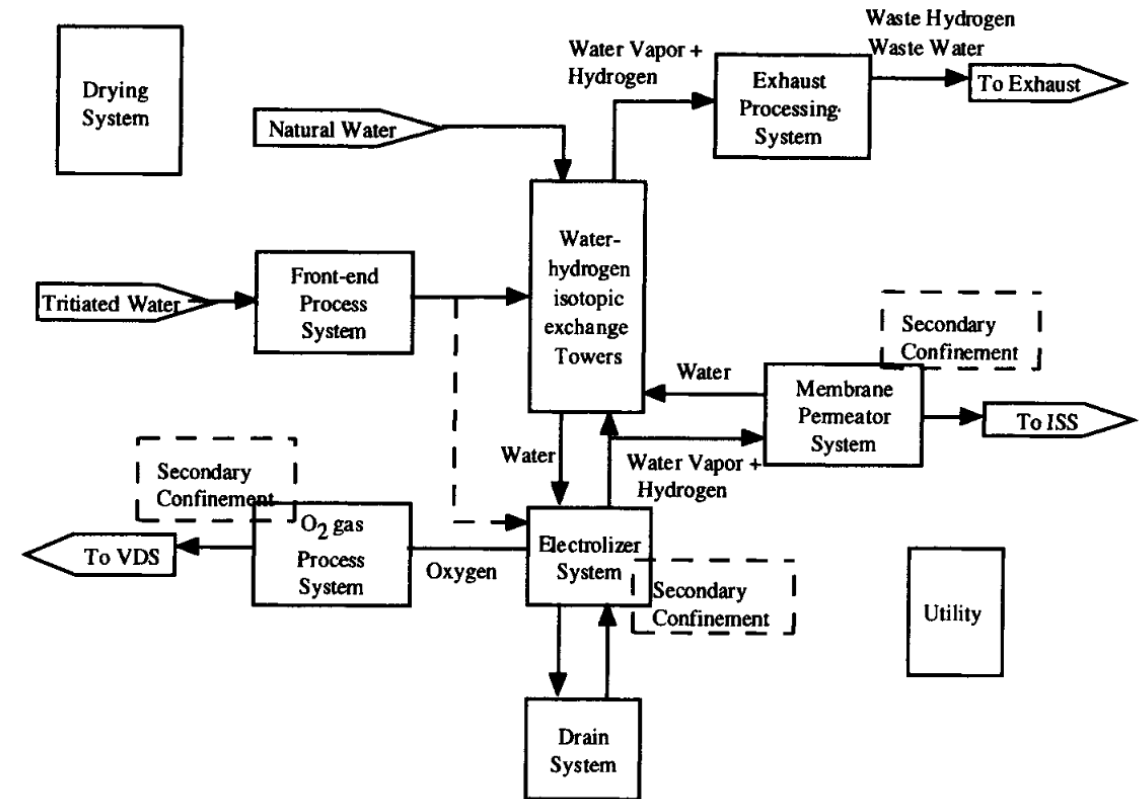
James et al. (2017), *Tritium Aging of LaNi_{4.15}Al_{0.85} (LANA.85)*, *Fusion Science and Technology*, 71(3), 565–569





Trace Tritium

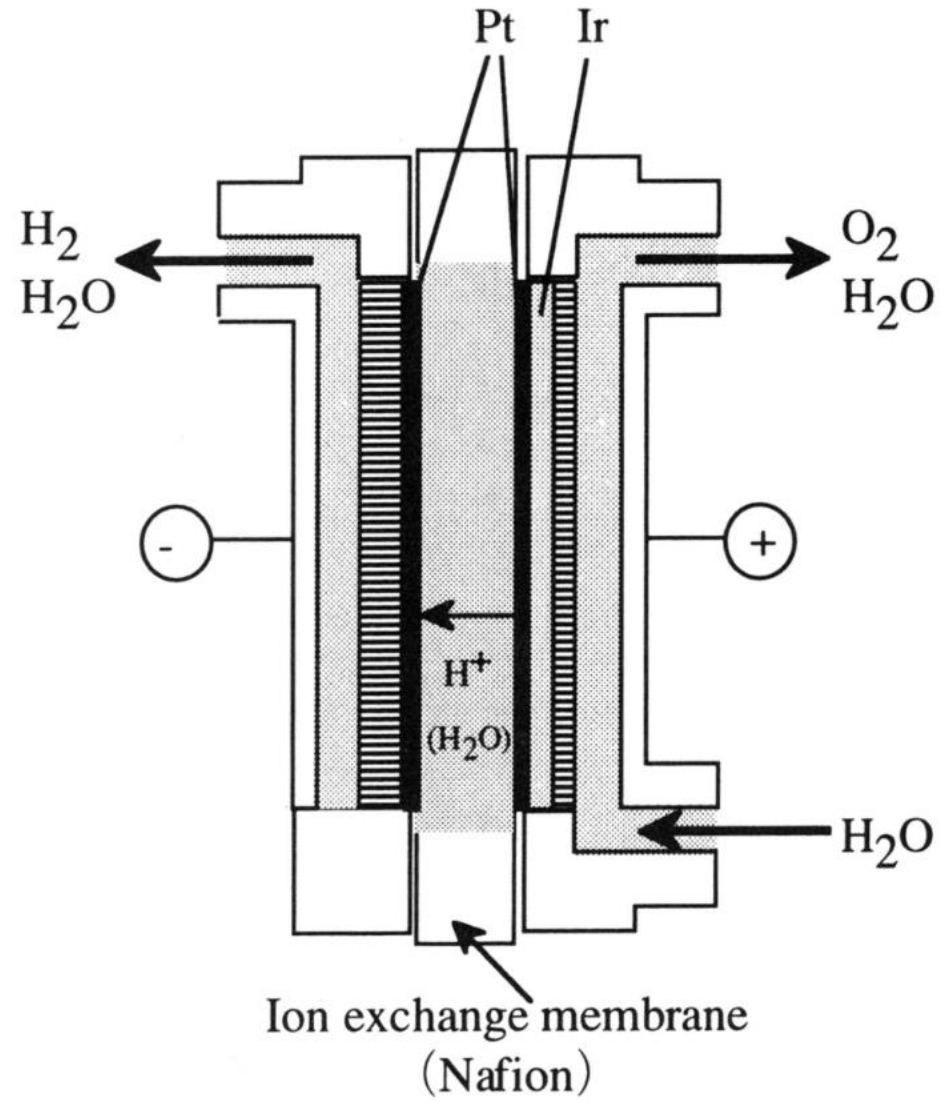
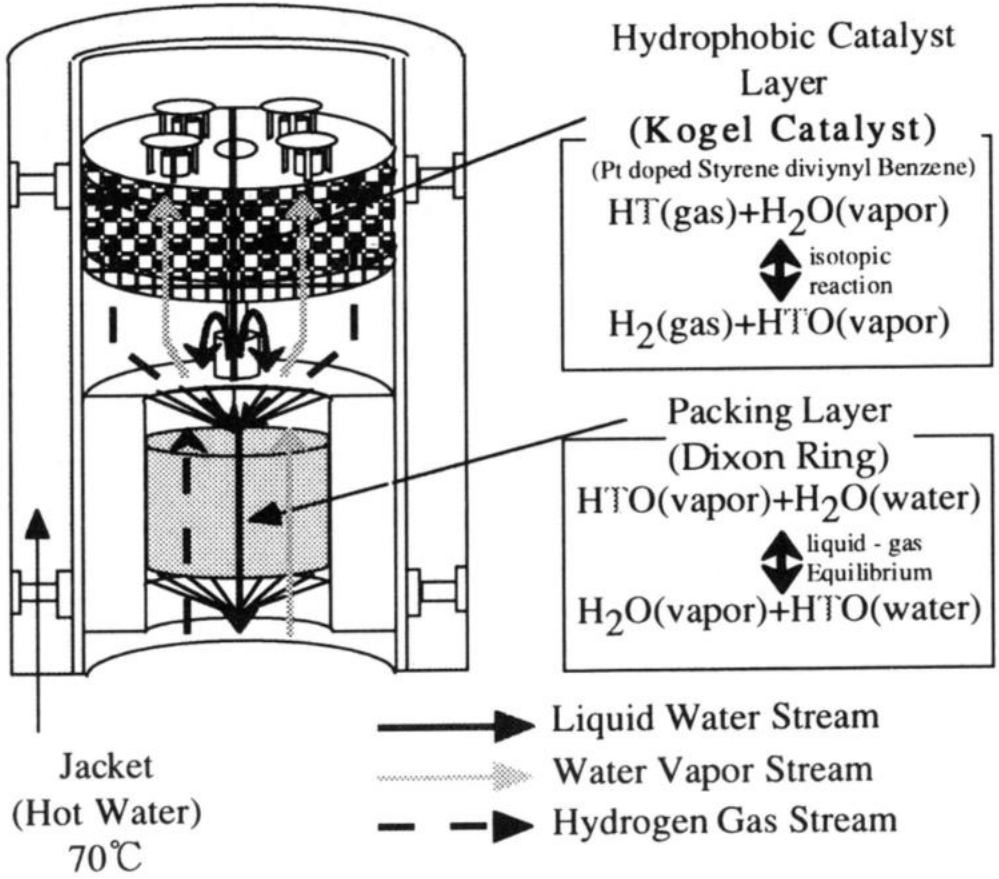
- Trace tritium recovery systems deal with trace tritium in systems:
 - Trace tritium in waste streams from isotope separation systems (tritium recovery is not 100 %)
 - Reactor coolant streams
 - Air from gloveboxes (especially due to tritium release during maintenance periods)
 - Process gases and liquids
- Example systems are electrolysis-based systems for water detritiation or getter beds to capture hydrogen or HTO from gas steams for gas detritiation



Iwai et al., 2002 "The Water Detritiation System of the ITER Tritium Plant"
Fusion Science and Technology, 41(3P2), 1126–1130



Trace Tritium



Fuel Cycle Technologies

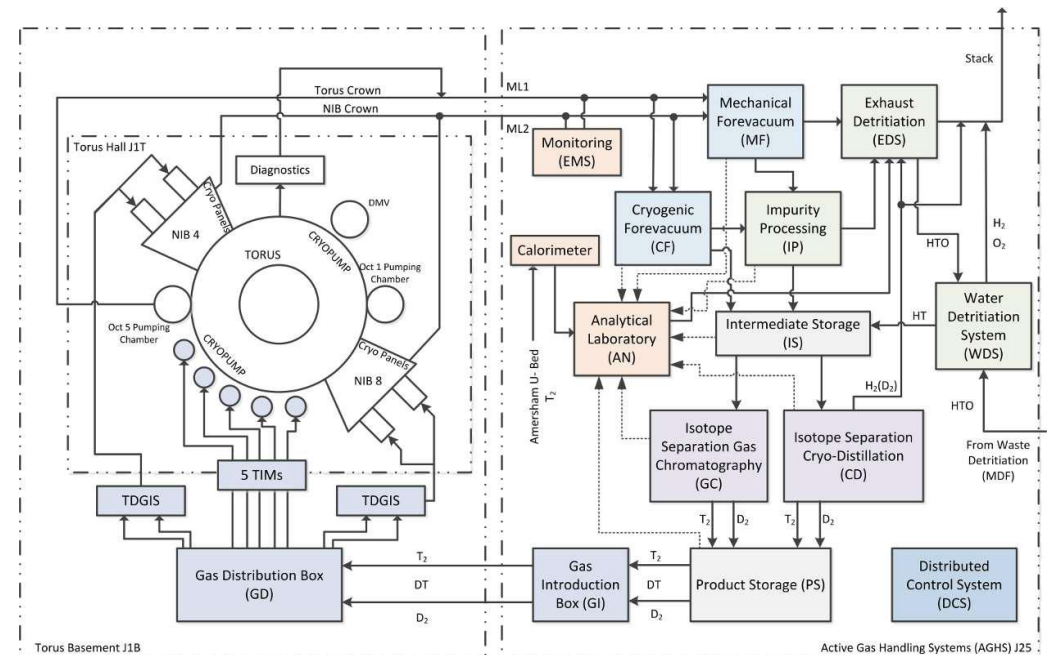
- Several technologies have been presented [we have gone in to varying detail on these!]
- It is important to consider the way we use tritium properties, rather than the technologies themselves!
 - There are many technologies that could be used for the different blocks you have seen
 - They utilize different things e.g.:
 - The permeation properties of hydrogen allow for hydrogen isotope separation from other gases
 - The boiling point differences allow hydrogen isotope separation from each other
 - Chemical reactions between hydrogen isotopes and other molecules
- The fuel cycle is a purification process, where we attempt to get as much deuterium and tritium back to the fueling systems as fast as possible



Challenges in the fuel cycle

Current Fuel Cycles

- There are not any continuous fuel cycles in operation today
- Historically, fuel cycles have been *batch* operated
- While the learning from these is critical – these systems will not be suitable for commercial fusion devices
- The only currently operating DT fuel cycle is at JET AGHS (Joint European Torus – Active Gas Handling System)

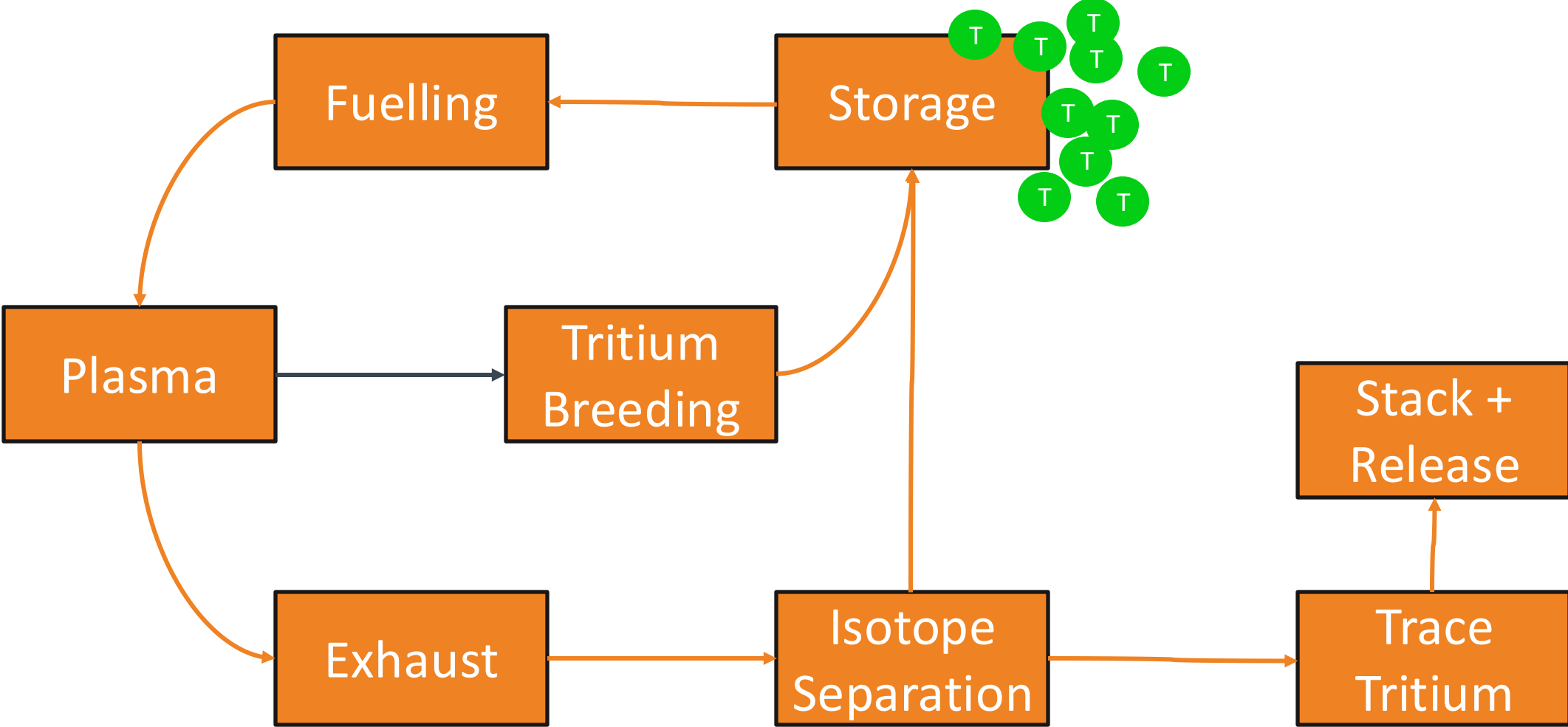


The key challenges

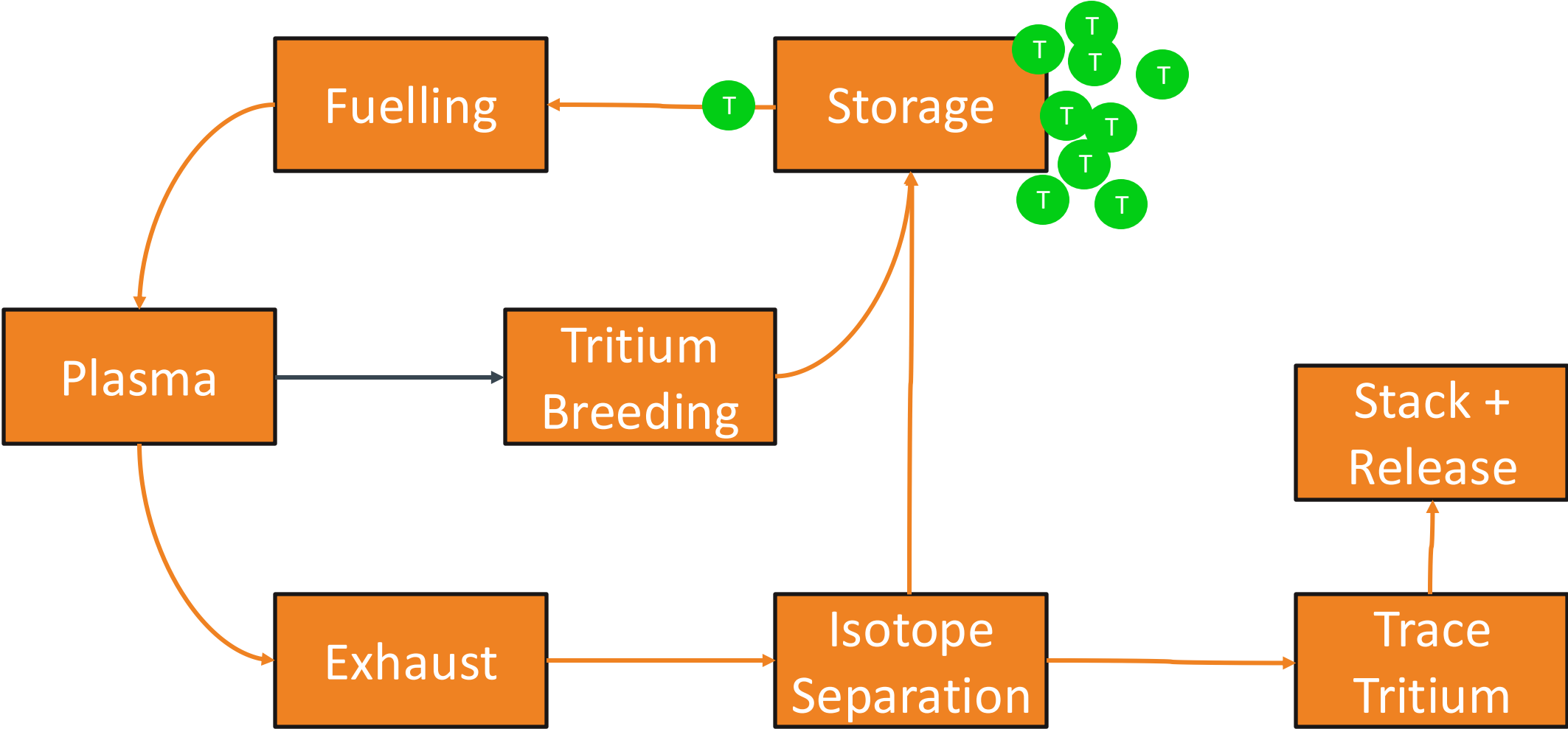
- Tritium Breeding Ratio – the ratio of tritium produced in the blanket to the tritium used must be >1 [more in Lane's talk after me!]
- Big startup inventories
 - It takes time to recycle unburnt and breed more tritium – in the meantime the plasma must be fuelled externally
 - As tritium is scarce, this initial *startup inventory* must be minimized
- Building an inventory for future plants – *doubling time*
- Tritium accountancy – how do we keep track of all the tritium
 - We need continuous methods of tracking tracking, and the concentrations in each of our Fuel Cycle systems
- Understanding losses in the system
 - We don't have a working continuous fuel cycle yet – how can we understand where those losses will be



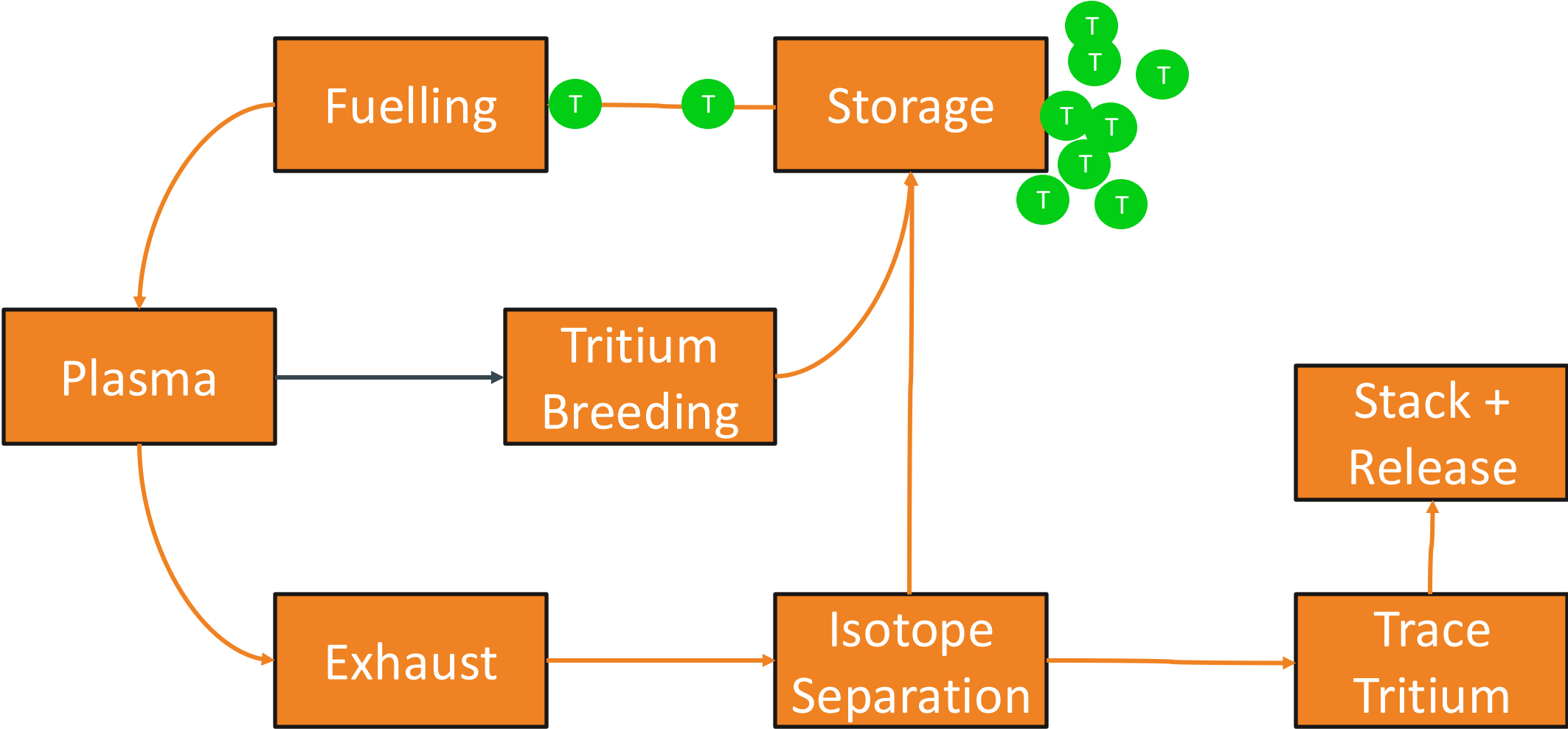
Startup Inventory



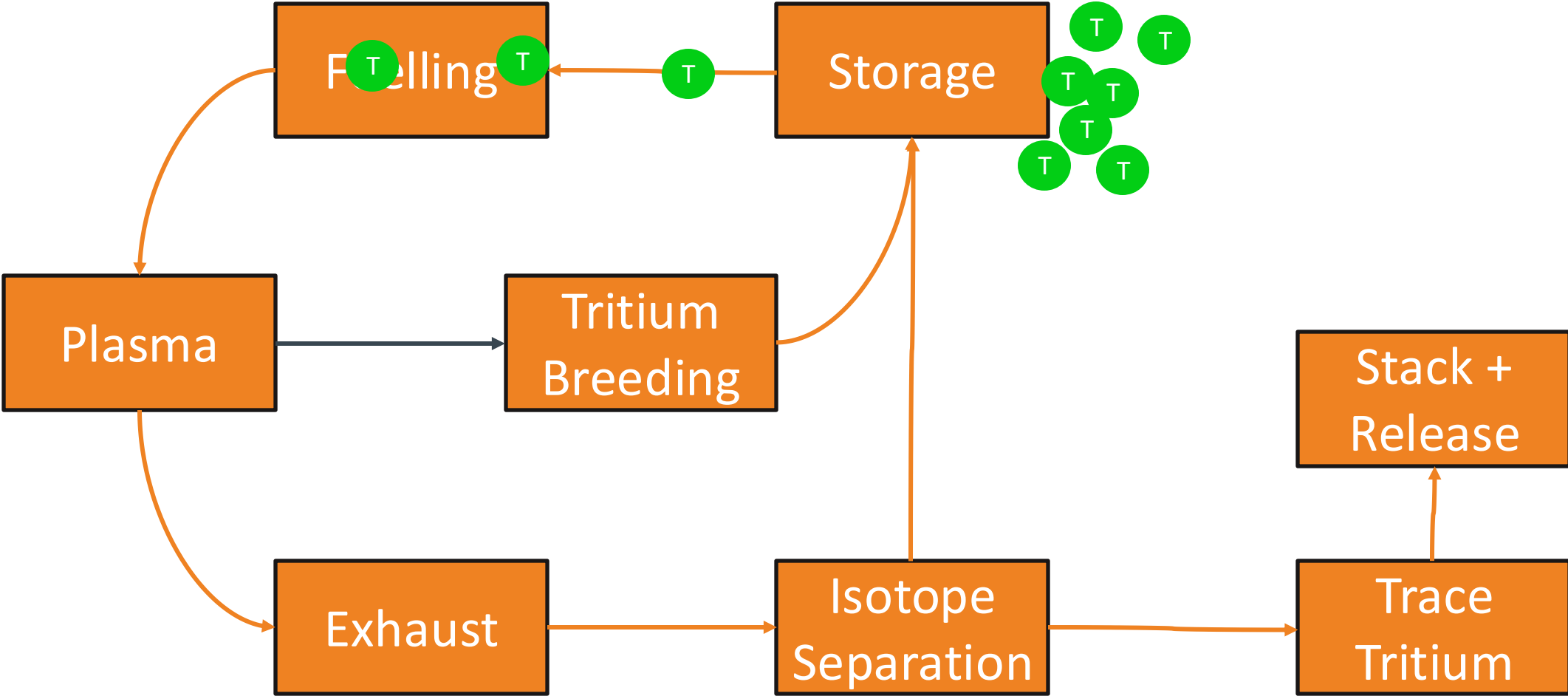
Startup Inventory



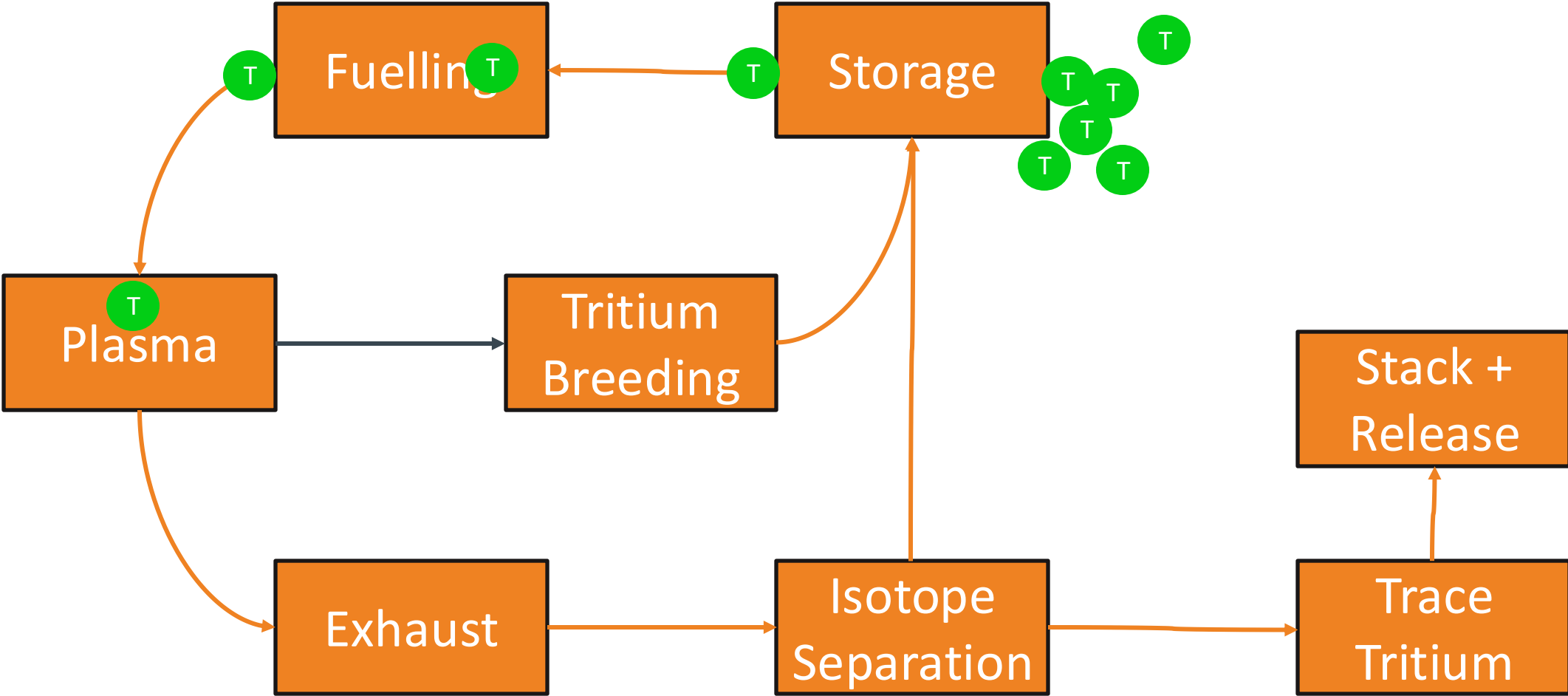
Startup Inventory



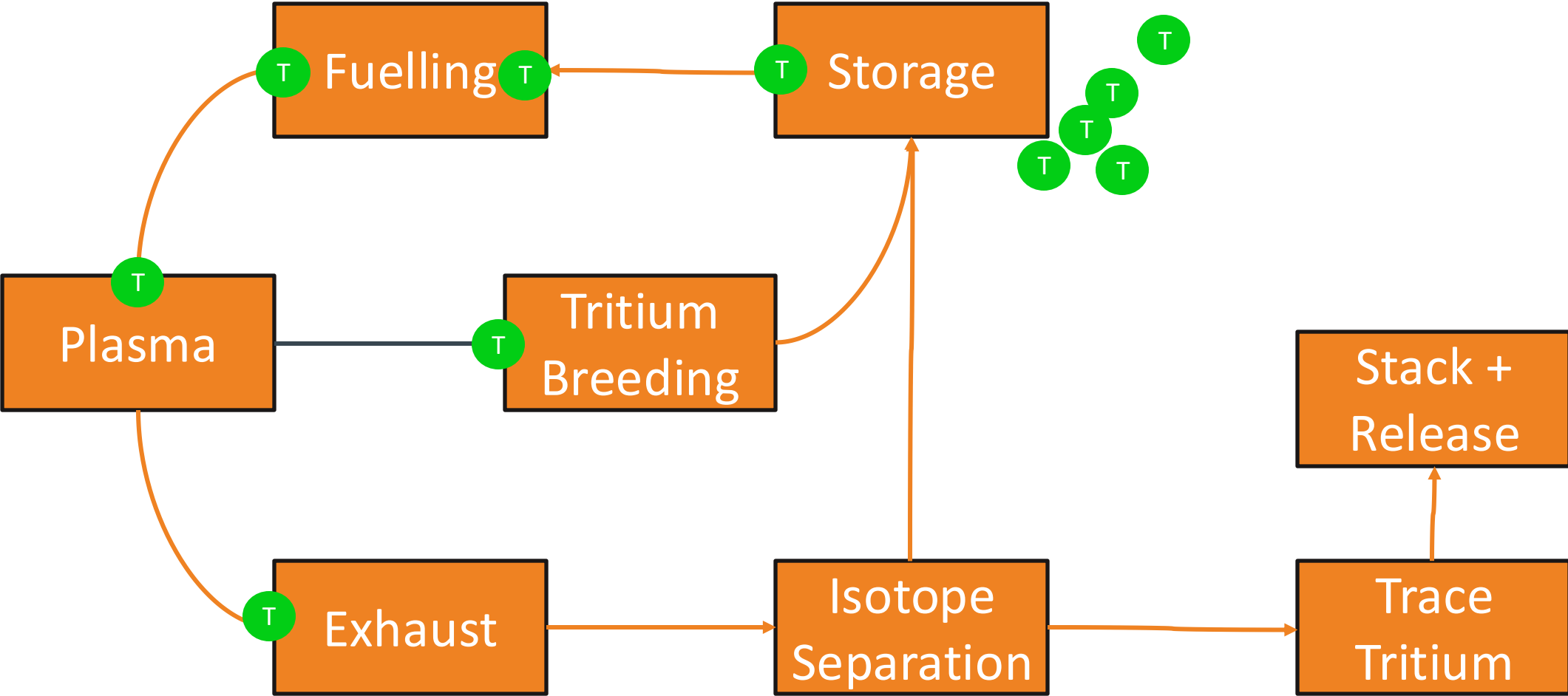
Startup Inventory



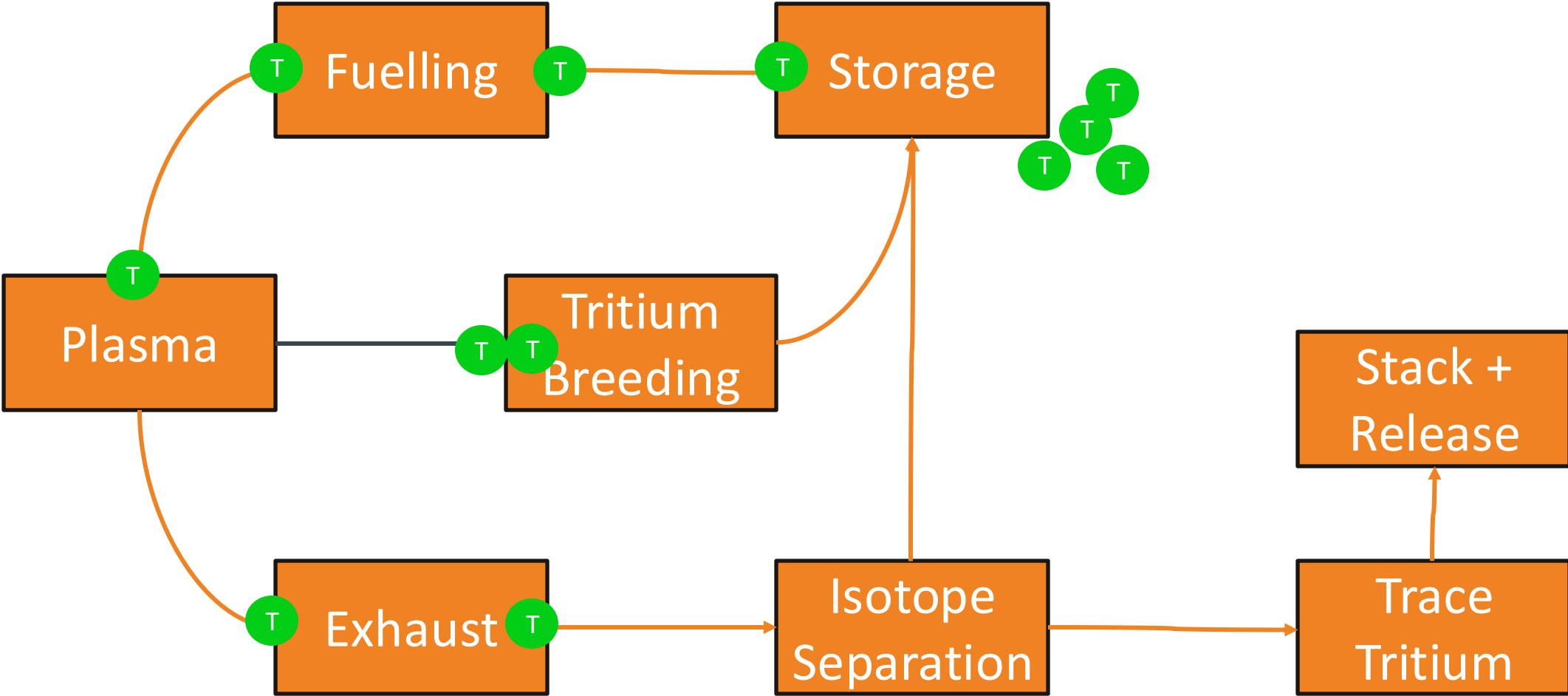
Startup Inventory



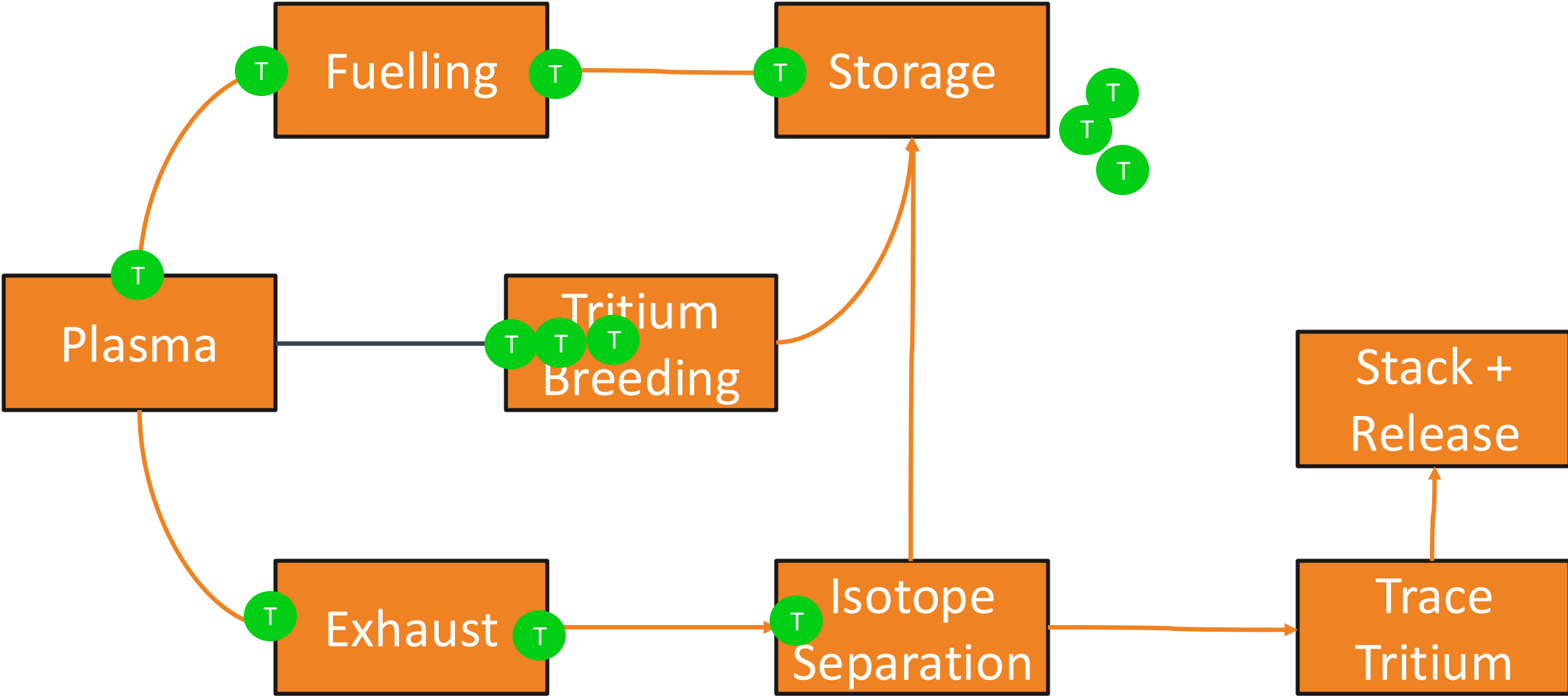
Startup Inventory



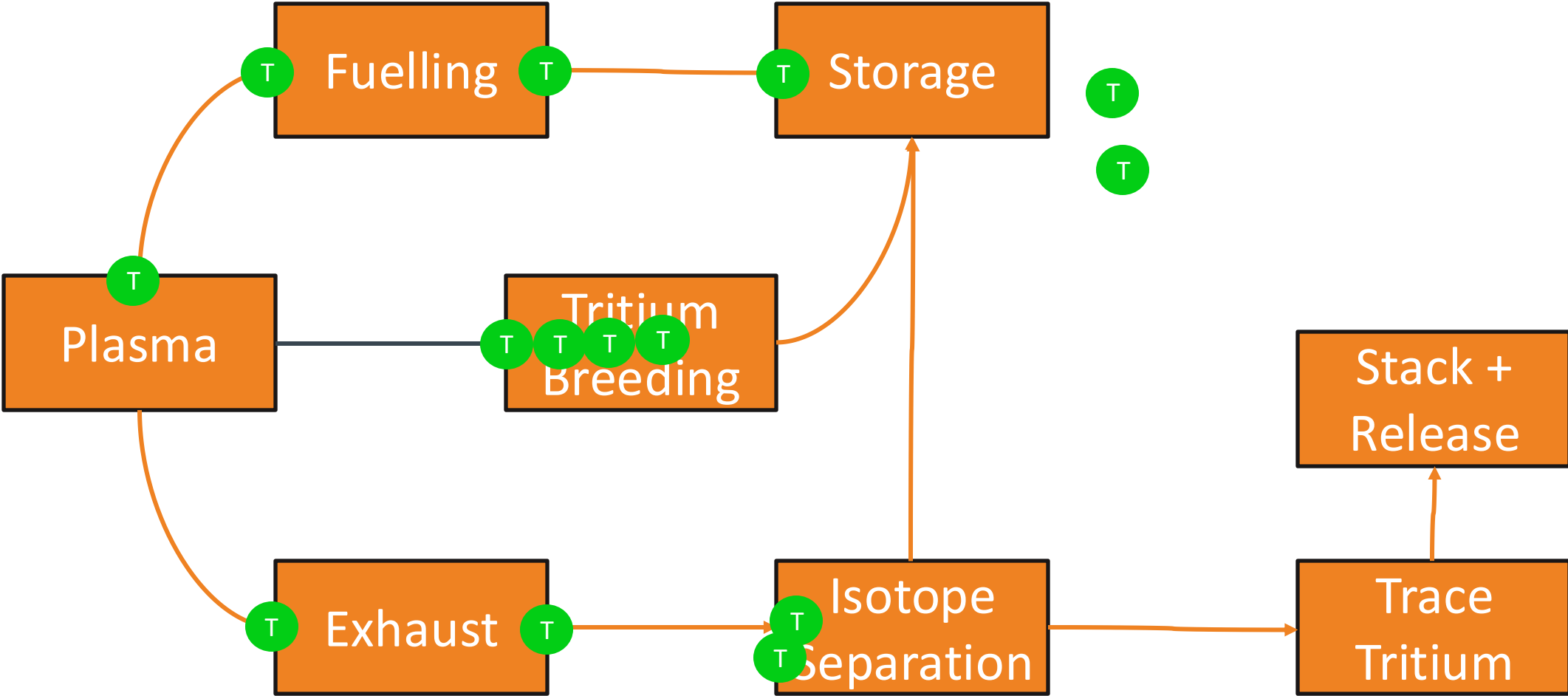
Startup Inventory



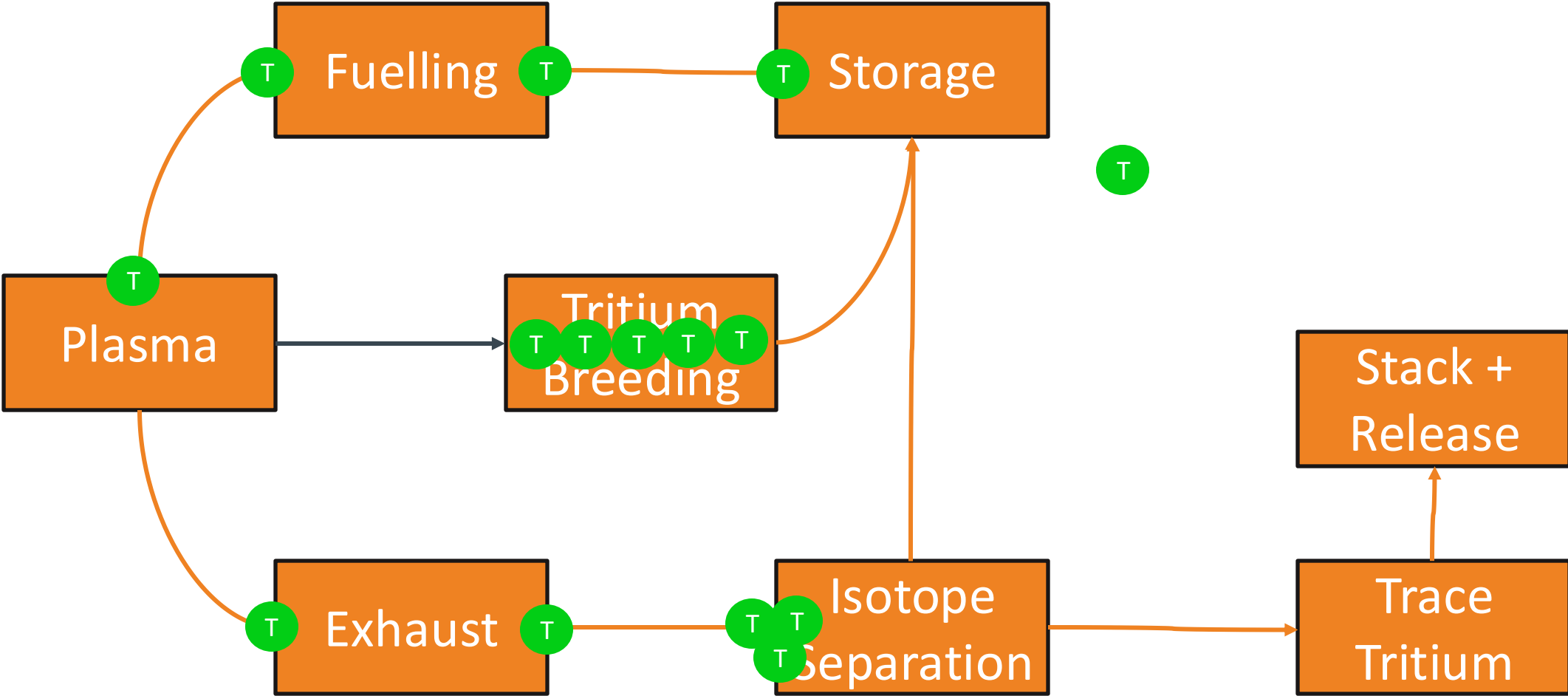
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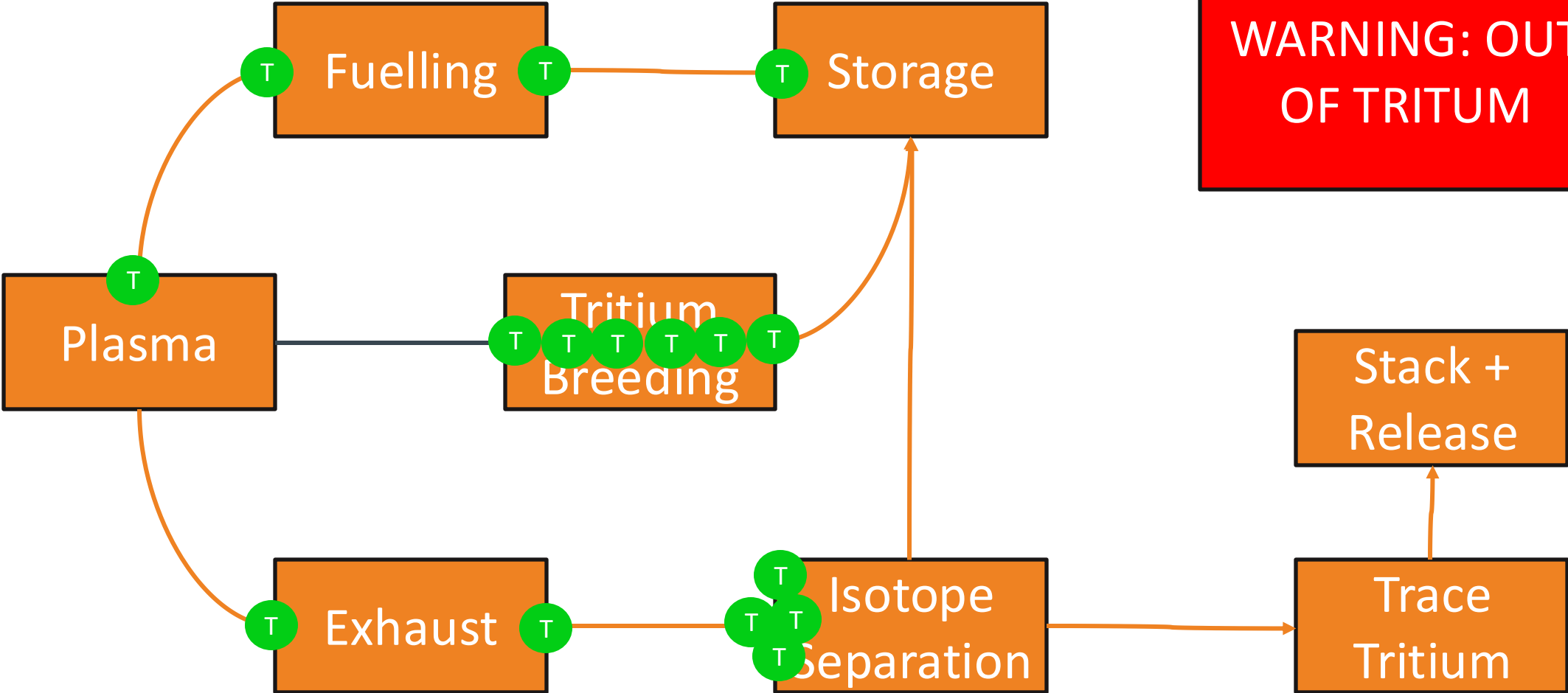
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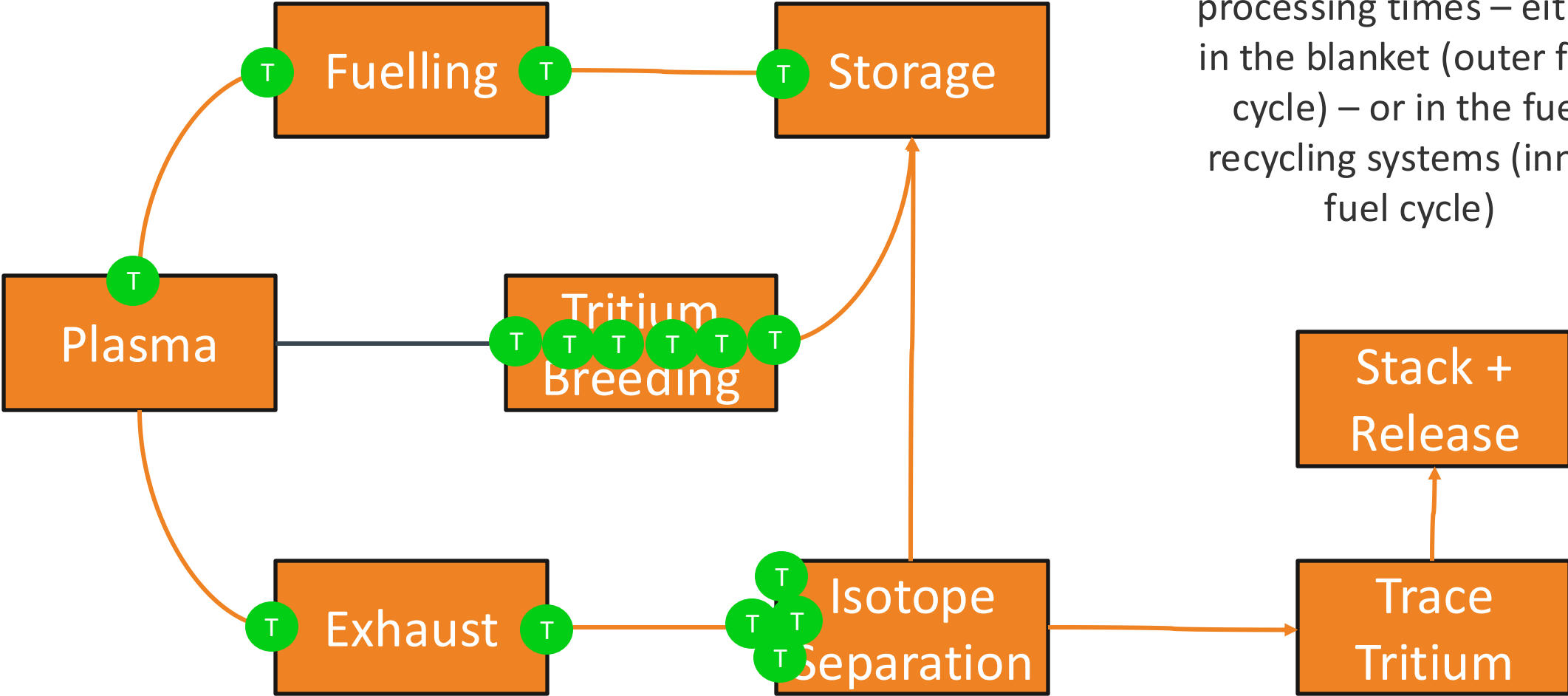
Startup Inventory



Startup Inventory



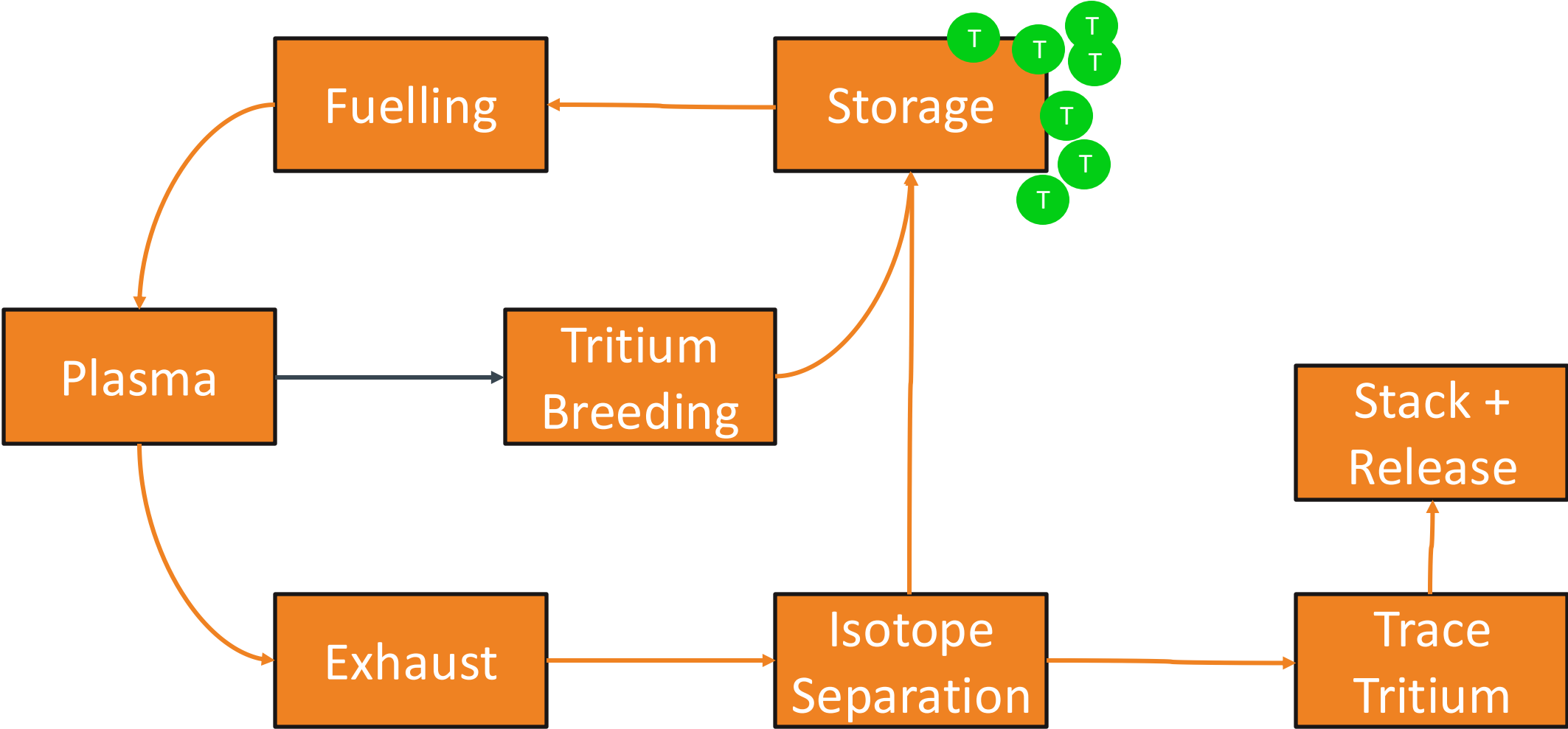
Startup Inventory



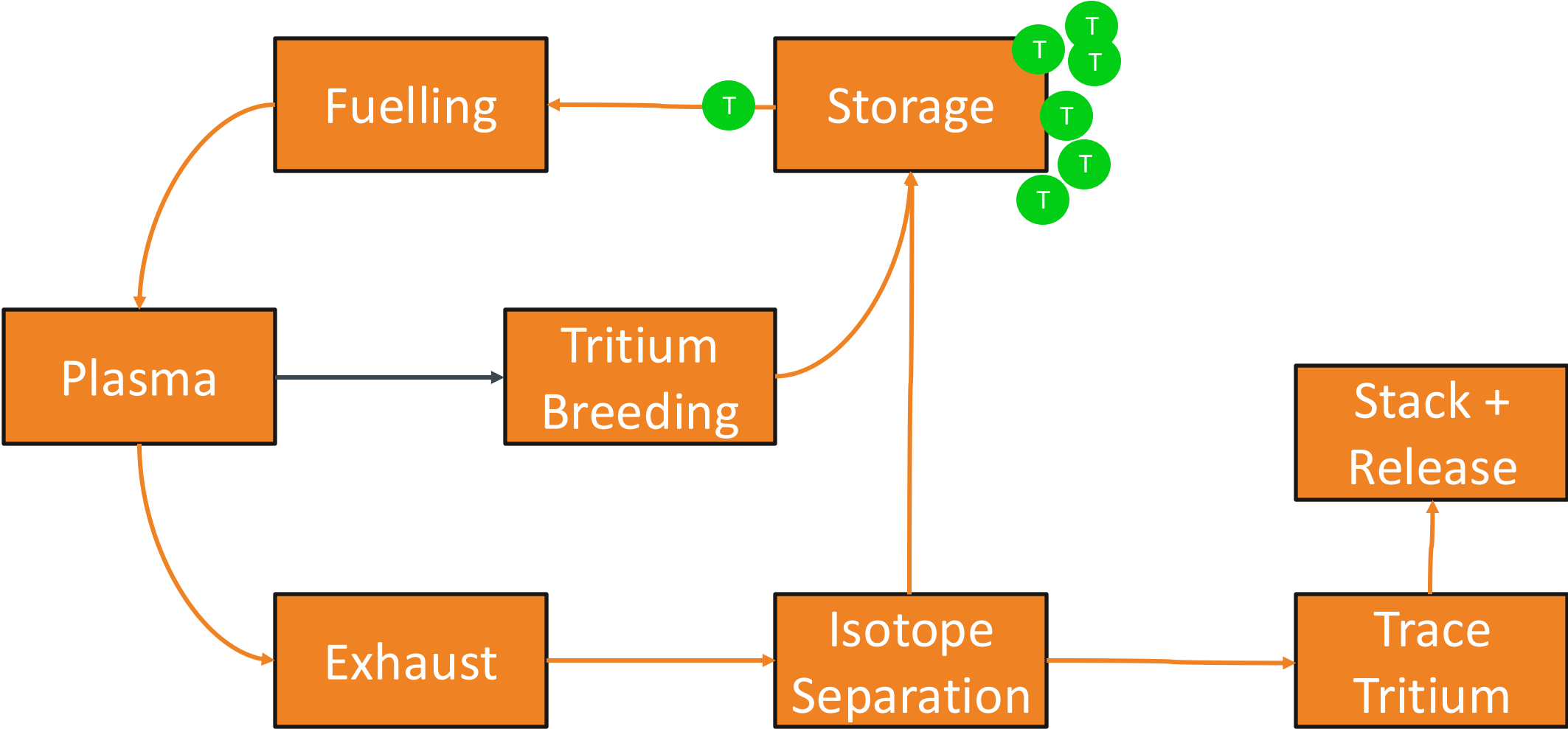
Need to improve processing times – either in the blanket (outer fuel cycle) – or in the fuel recycling systems (inner fuel cycle)



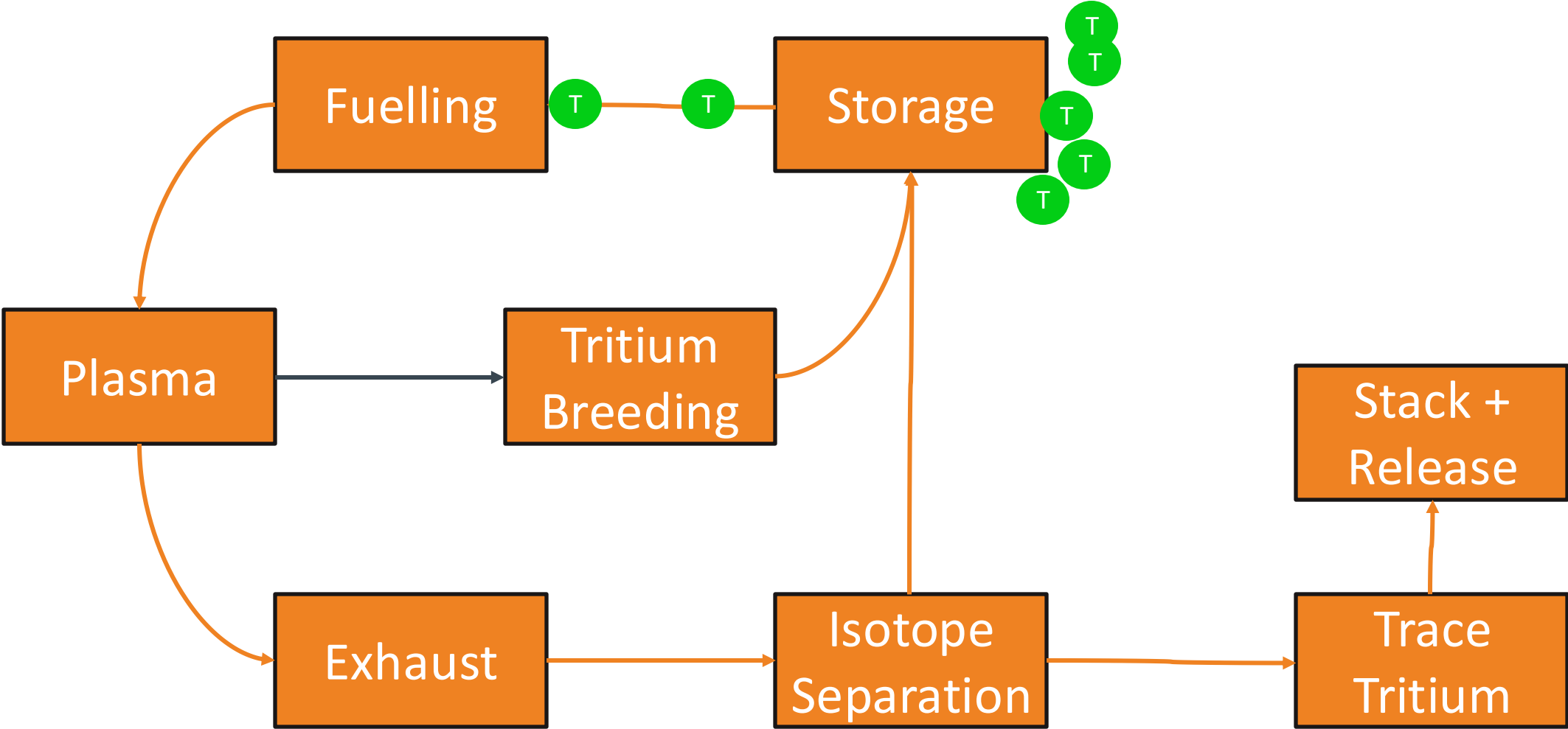
Startup Inventory



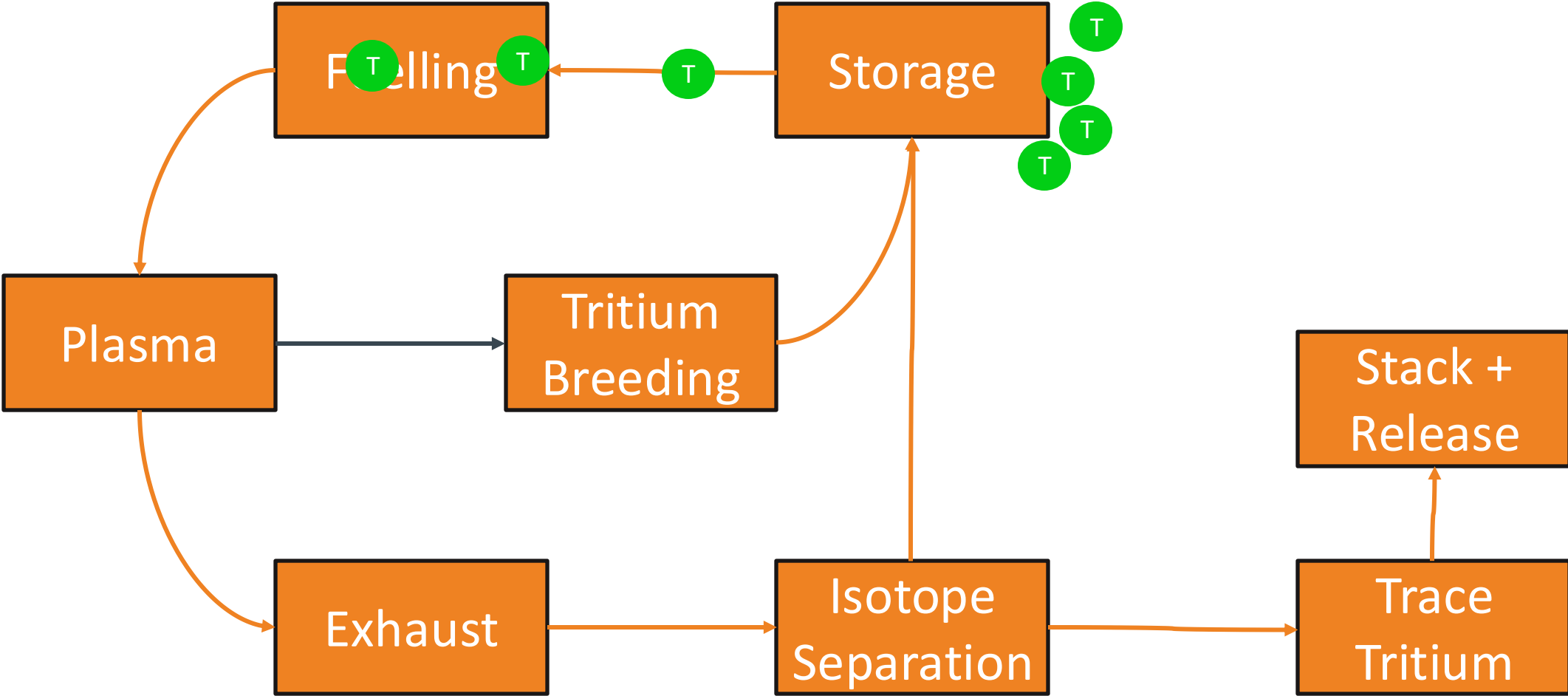
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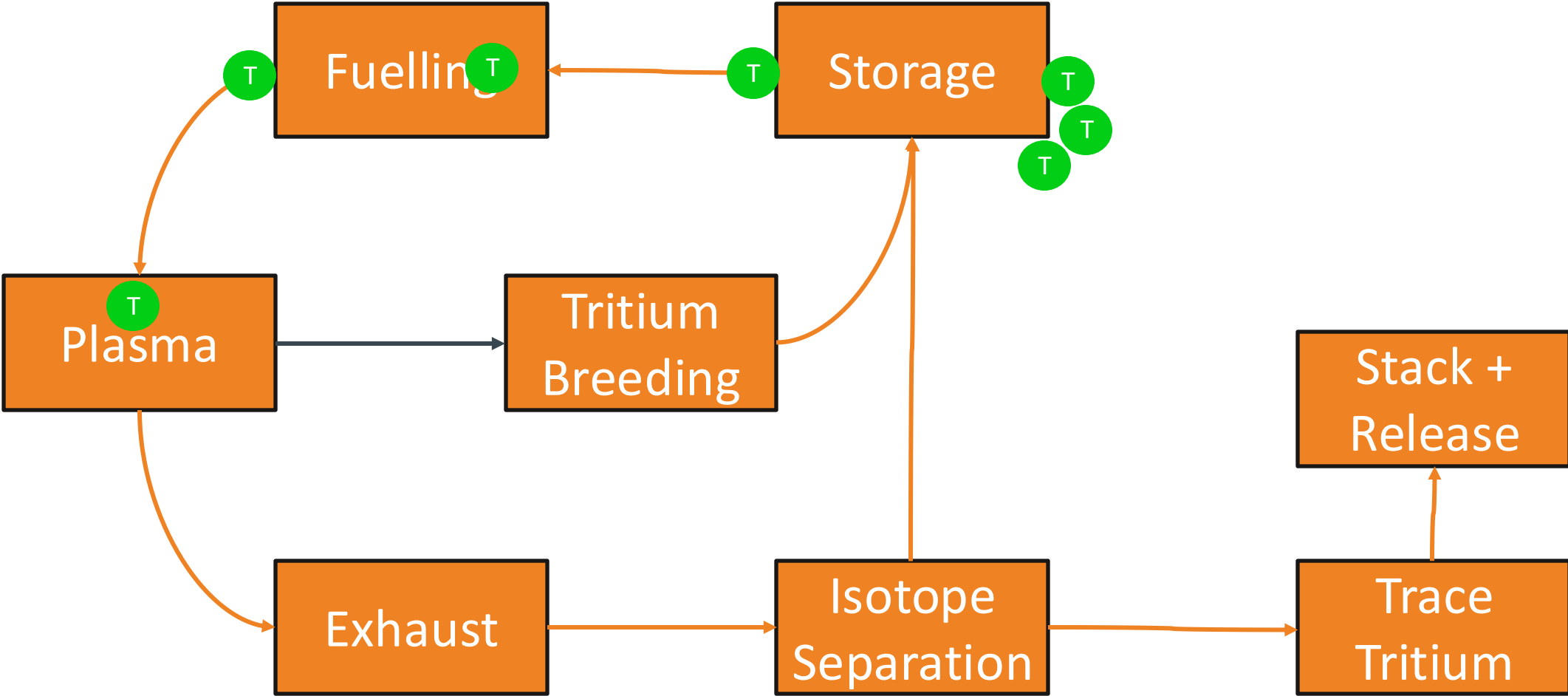
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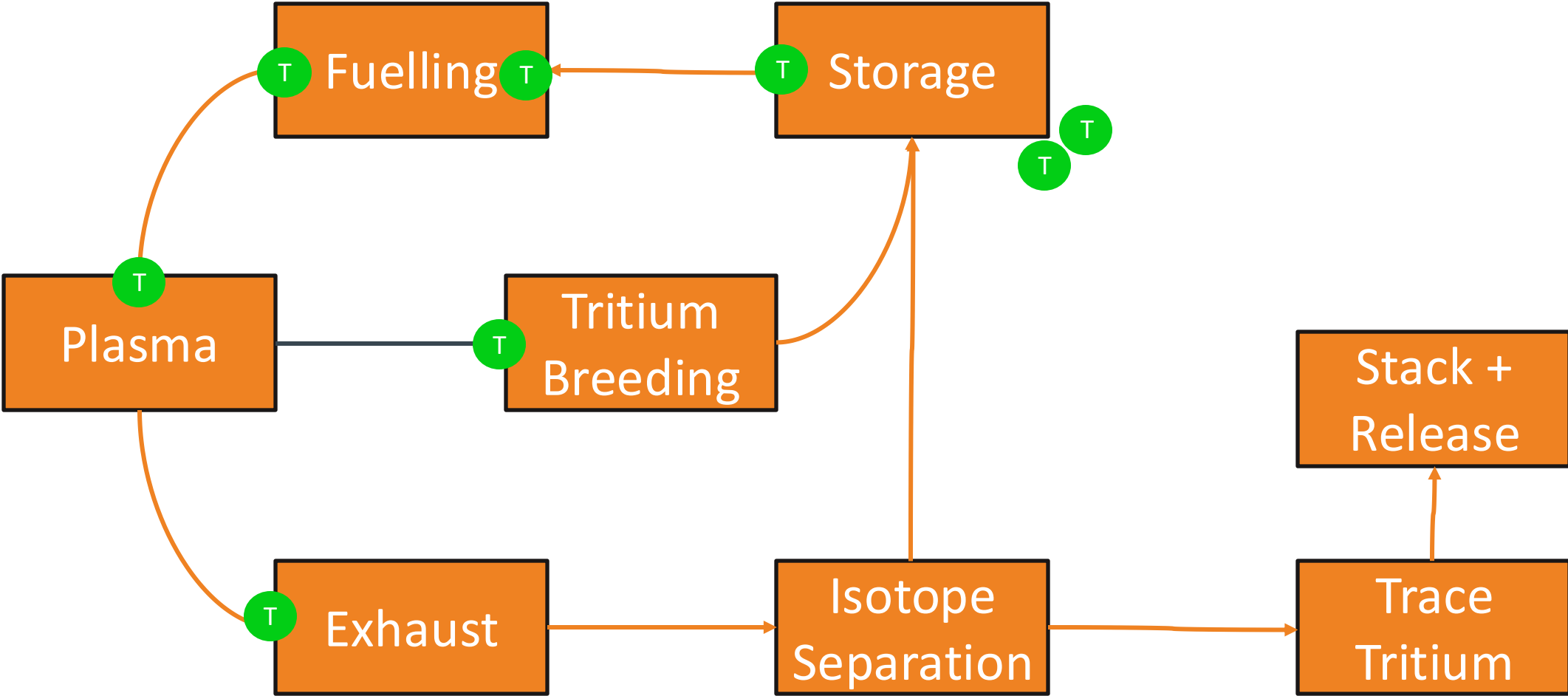
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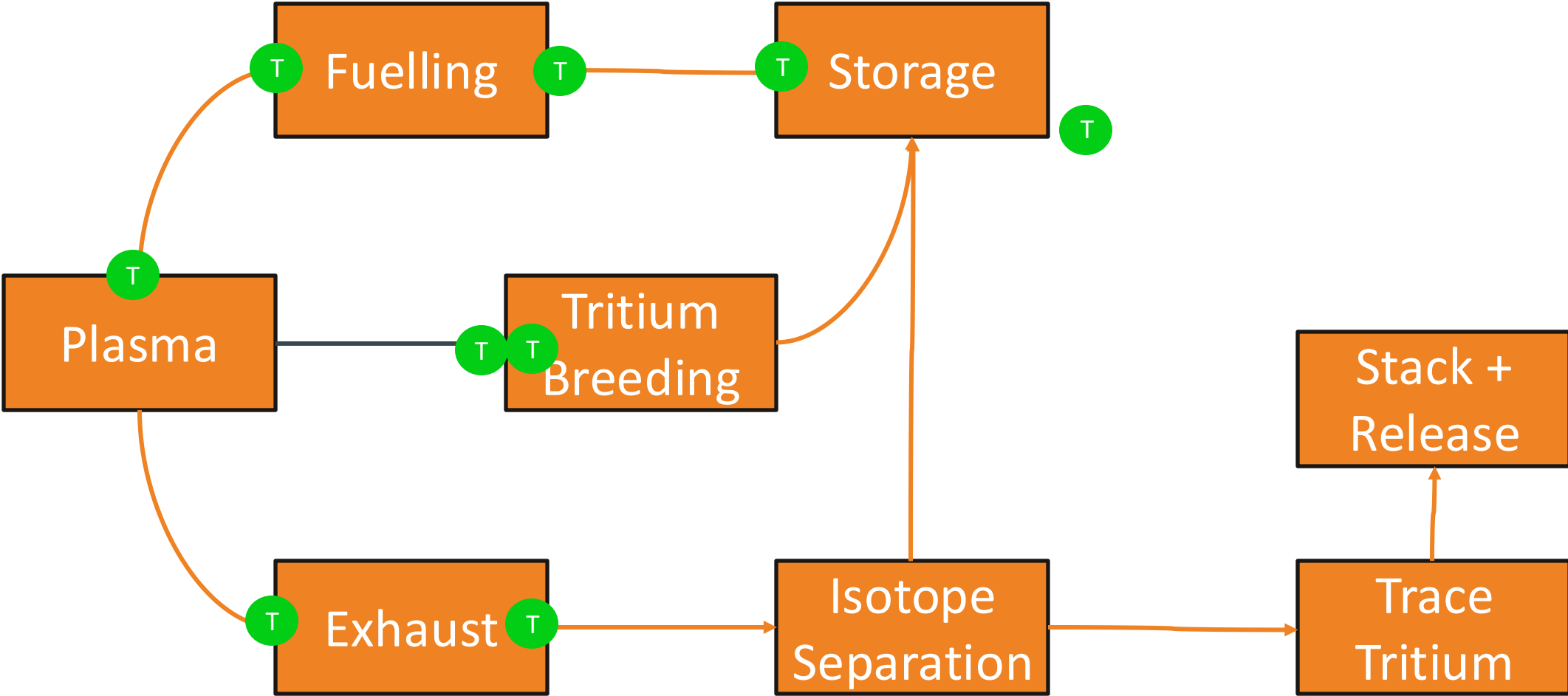
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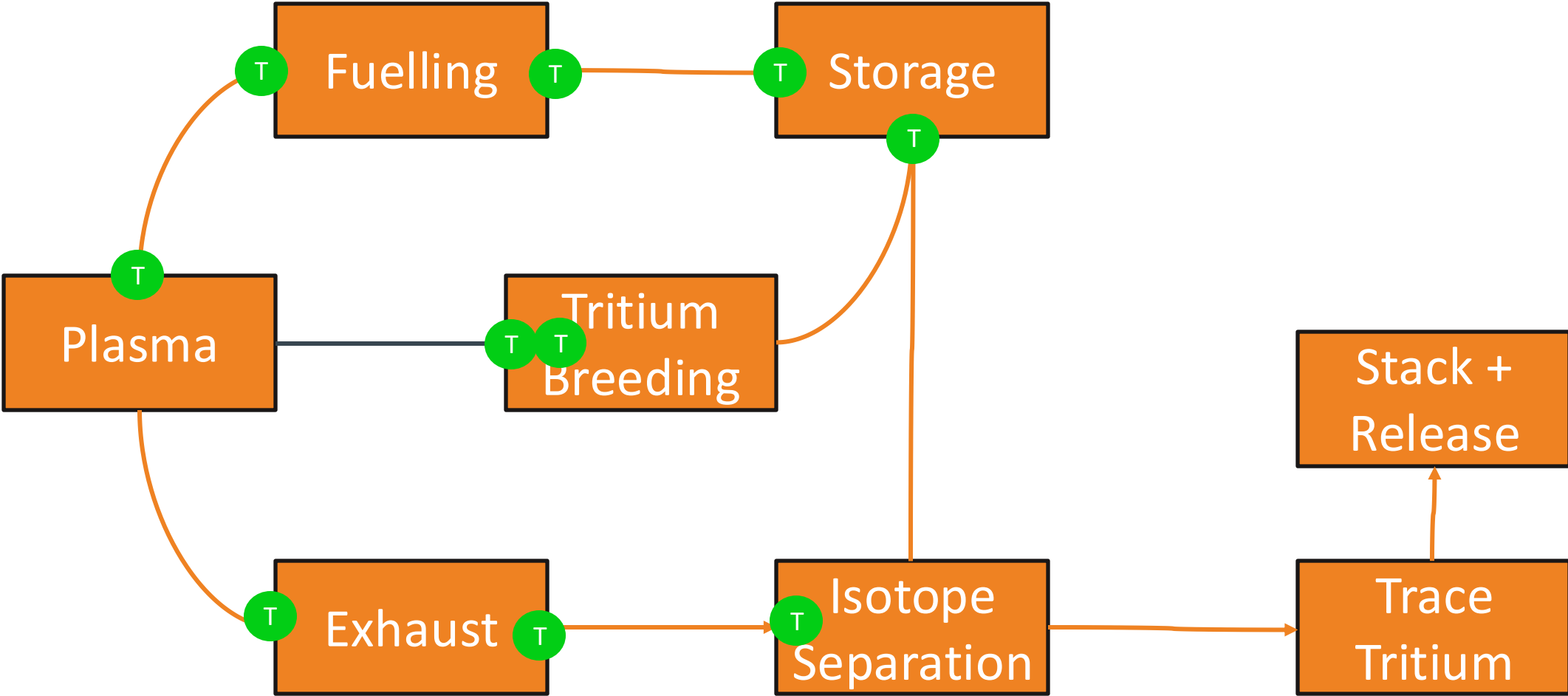
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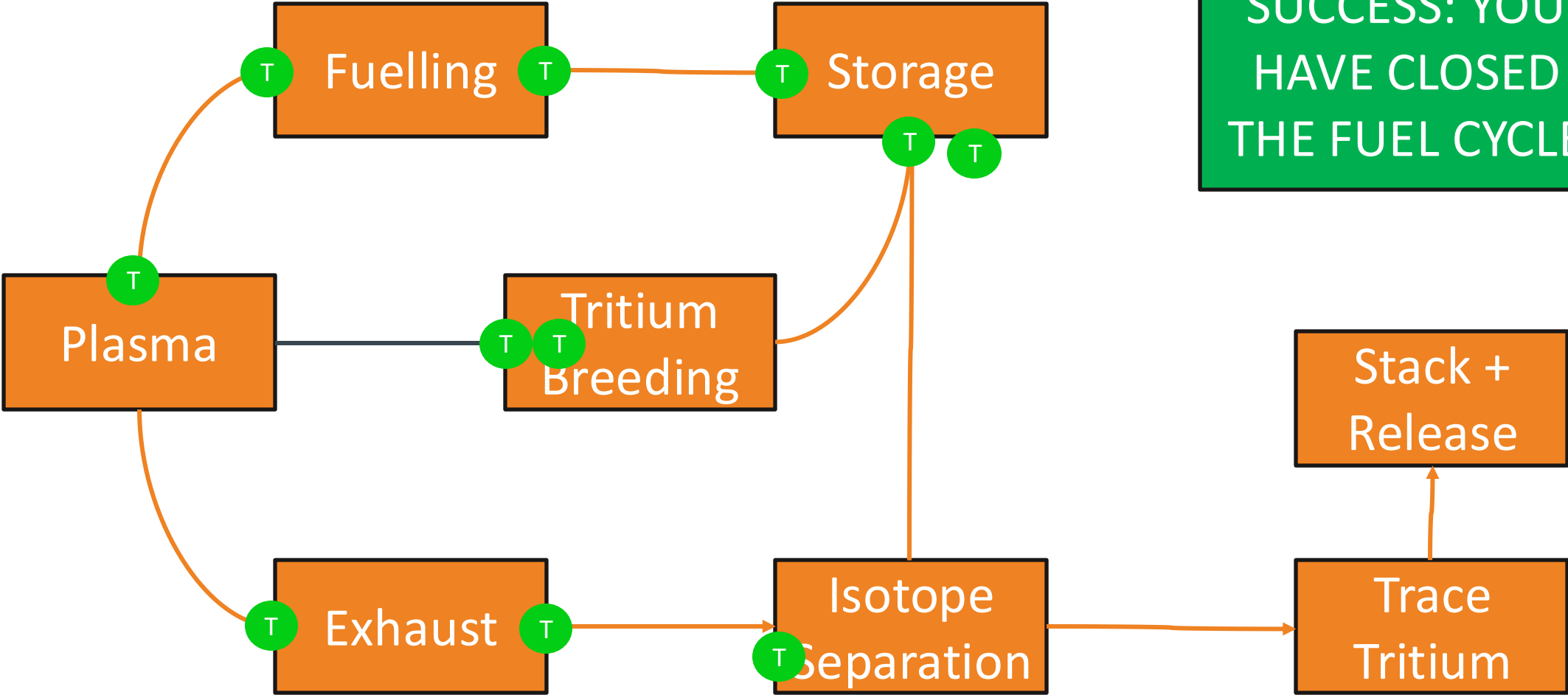
Startup Inventory



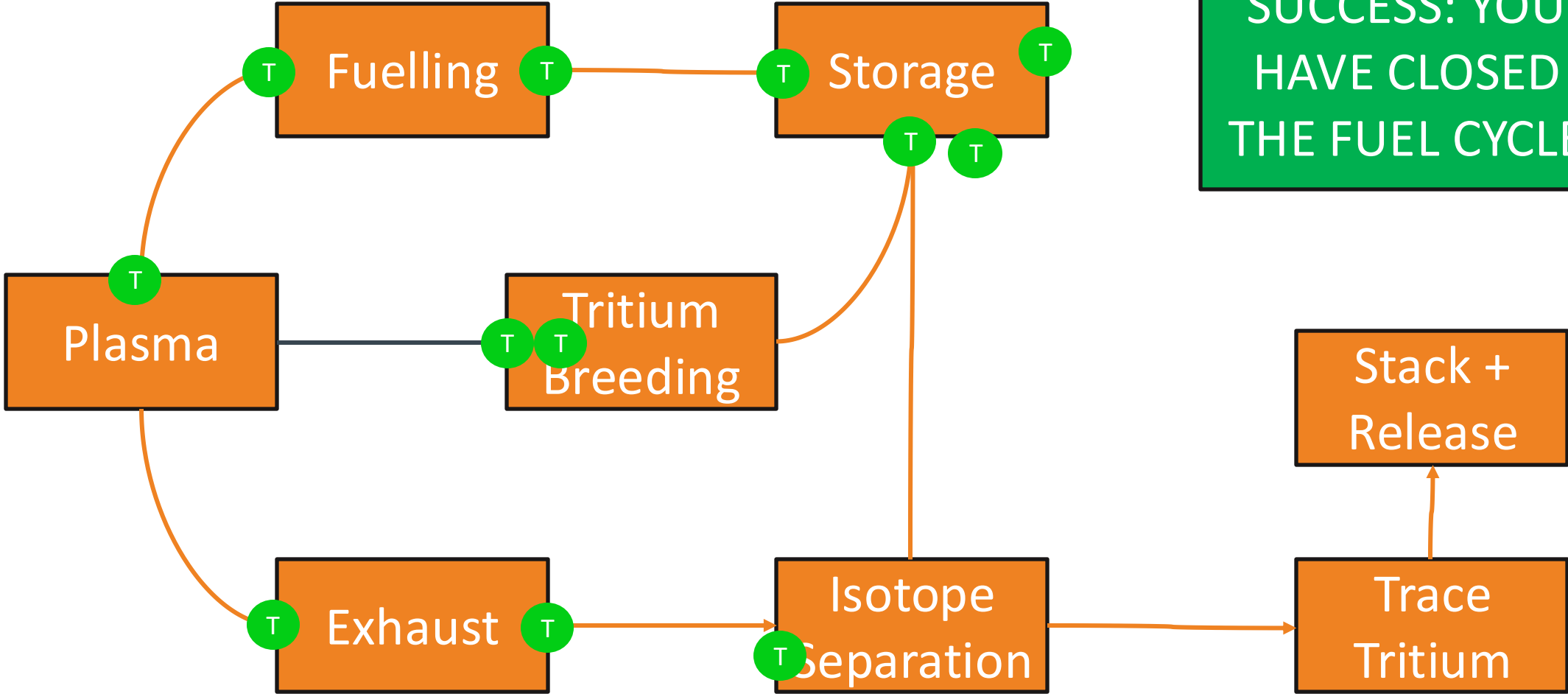
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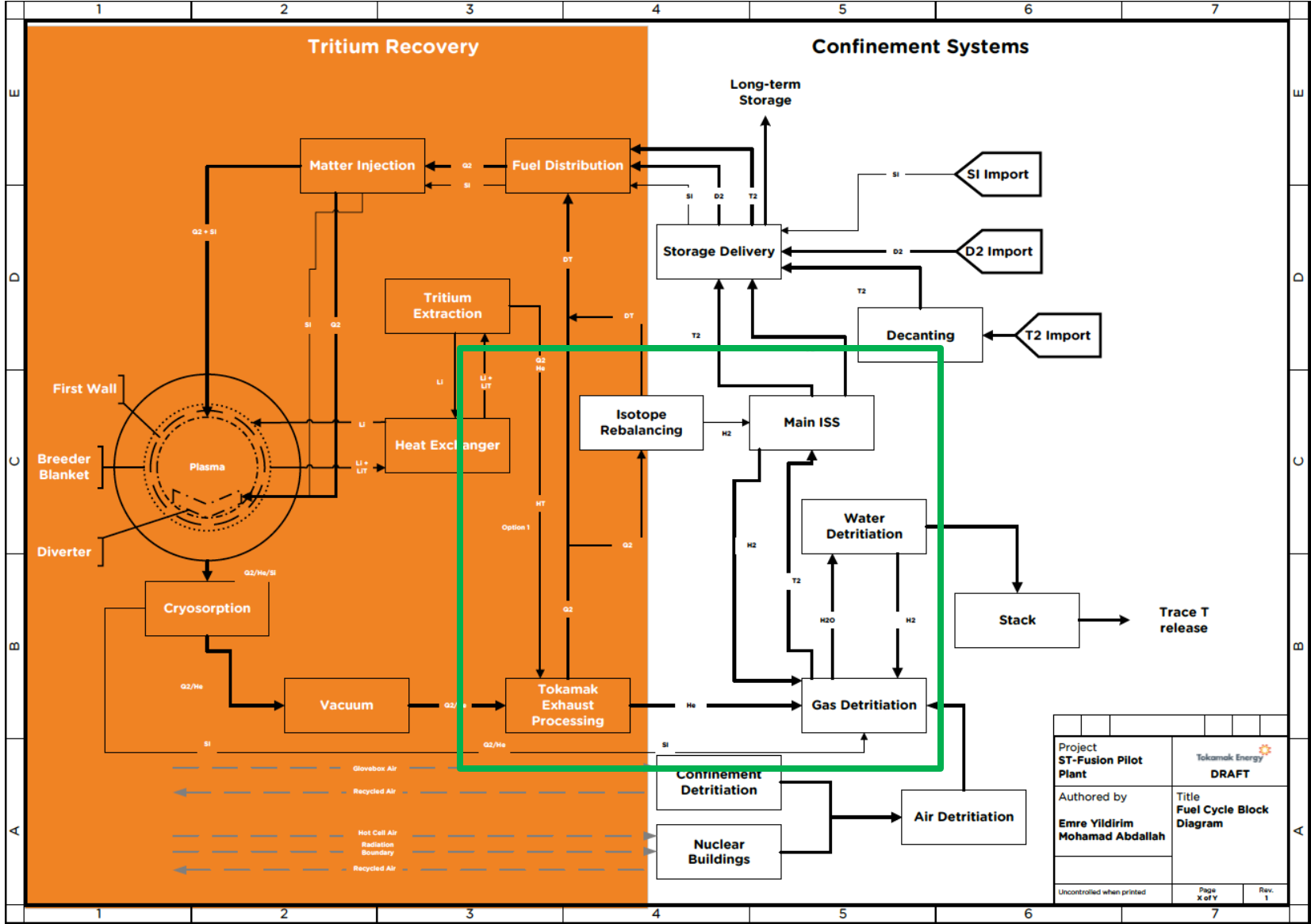
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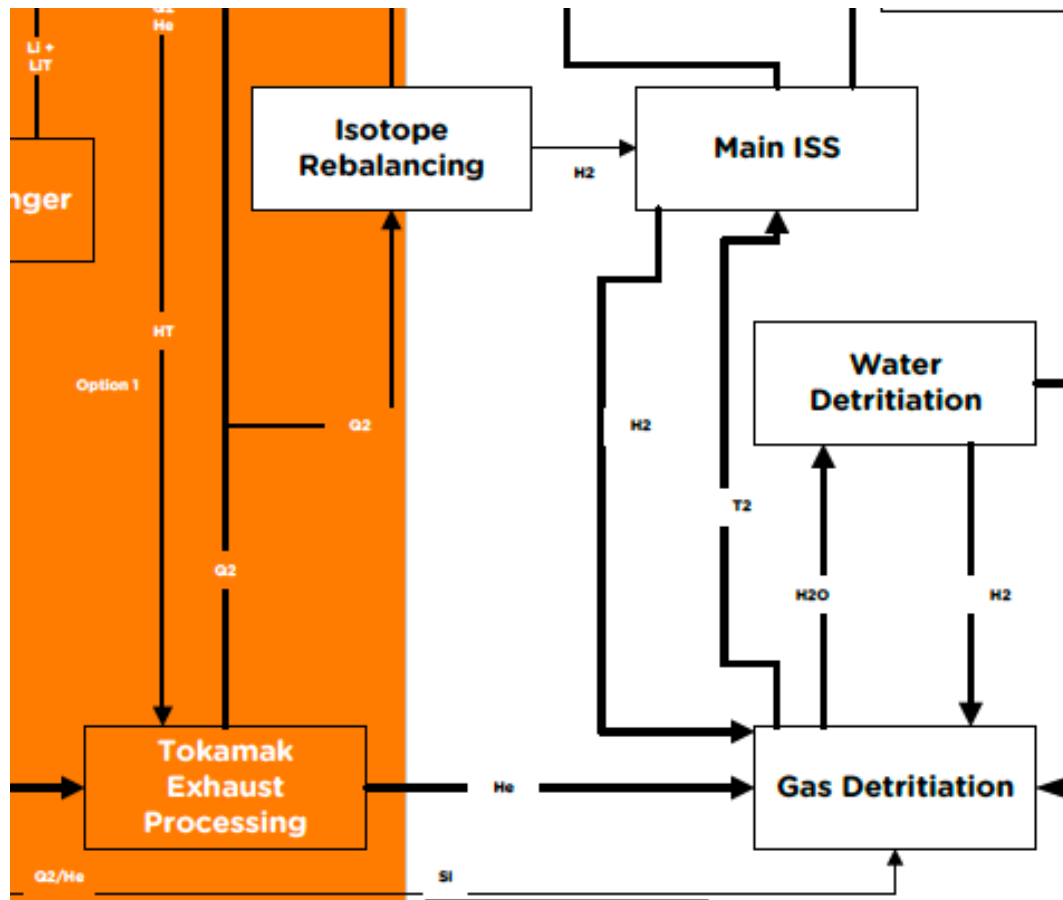
Startup Inventory



Speeding up processing times - example

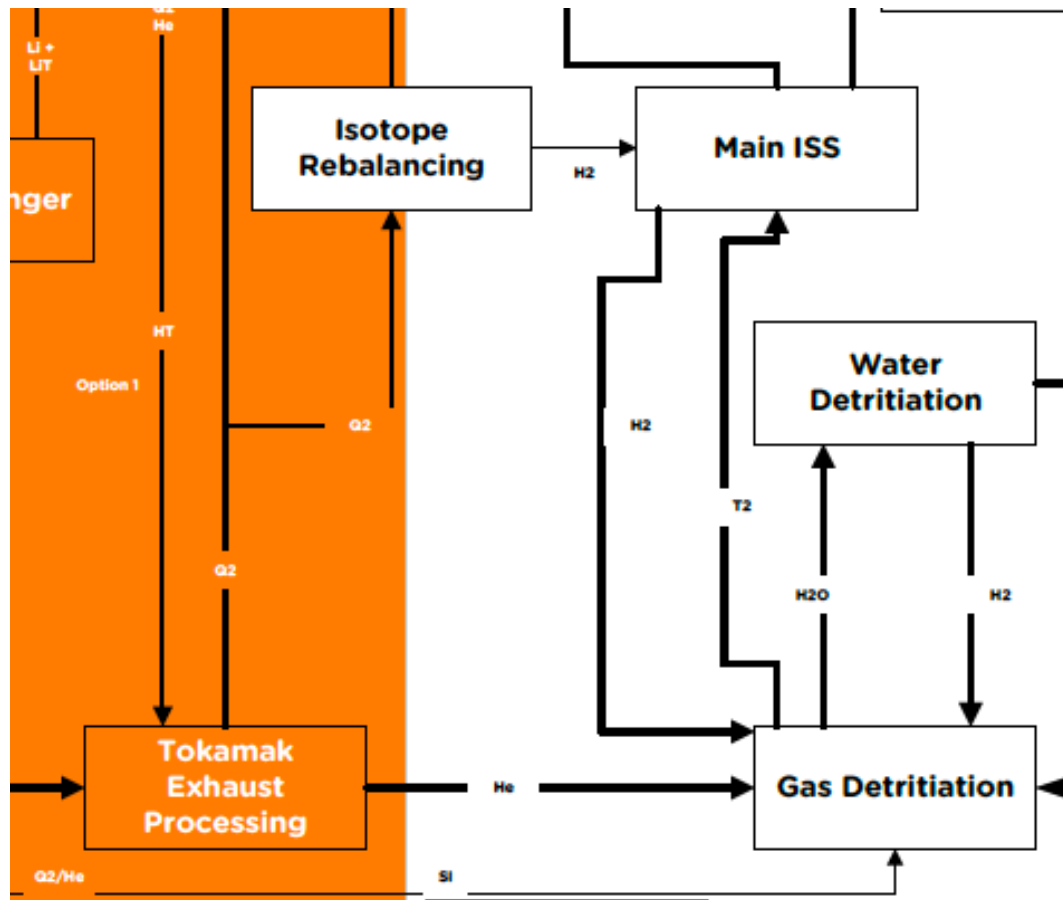


Speeding up processing times



- Instead of having a single isotope separation function have a fast-rebalancing loop and one for trace tritium recovery
- This is sometimes called *direct internal recycling*
 - Allow only hydrogen isotopes to permeate through the fast loop to directly inject them back into the plasma
 - Do minimal isotope balancing in this loop to maintain ~50:50 D:T
 - Protium build up is usually the limiting factor

Speeding up processing times



- Some of the technologies discussed earlier can be utilized here:
- Palladium diffusers, metal foil pumps – remove hydrogen isotopes
- Cryo distillation columns, or TCAP can be used to rebalance – the more of this you need to do the slower your processing time!



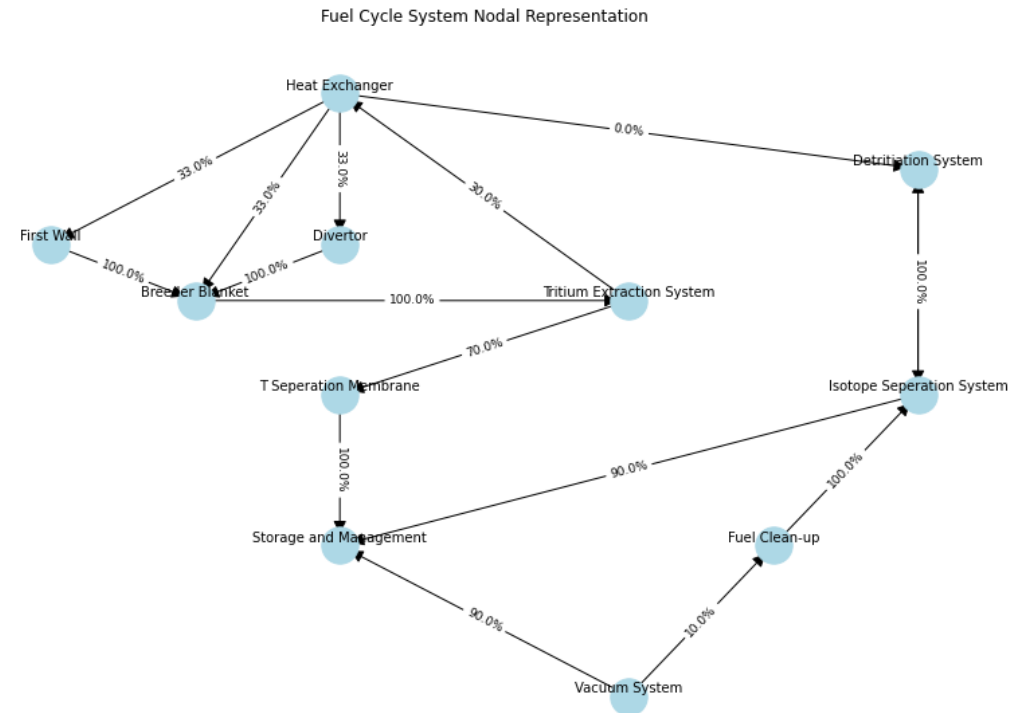
How we explore those

- We can explore these with *inventory models*
- These gives us an overview of the system to understand architecture, efficiency and processing time effects on key high level parameters
- They use the function
$$\frac{dI_i}{dt} = \sum_{j \neq i} \left(\frac{I_j}{\tau_j} \right)_i - (1 + \varepsilon_i) \frac{I_i}{\tau_i} - \lambda I_i + S_i$$
 - I_i — Tritium inventory in component i
 - τ_i — Residence time in component i
 - ε_i — Non-radioactive loss fraction in component i
 - λ — Tritium decay constant
 - S_i — Source term (e.g., tritium produced in blanket)
 - $(I_j / \tau_j)_i$ — Flow of tritium from component j to component i

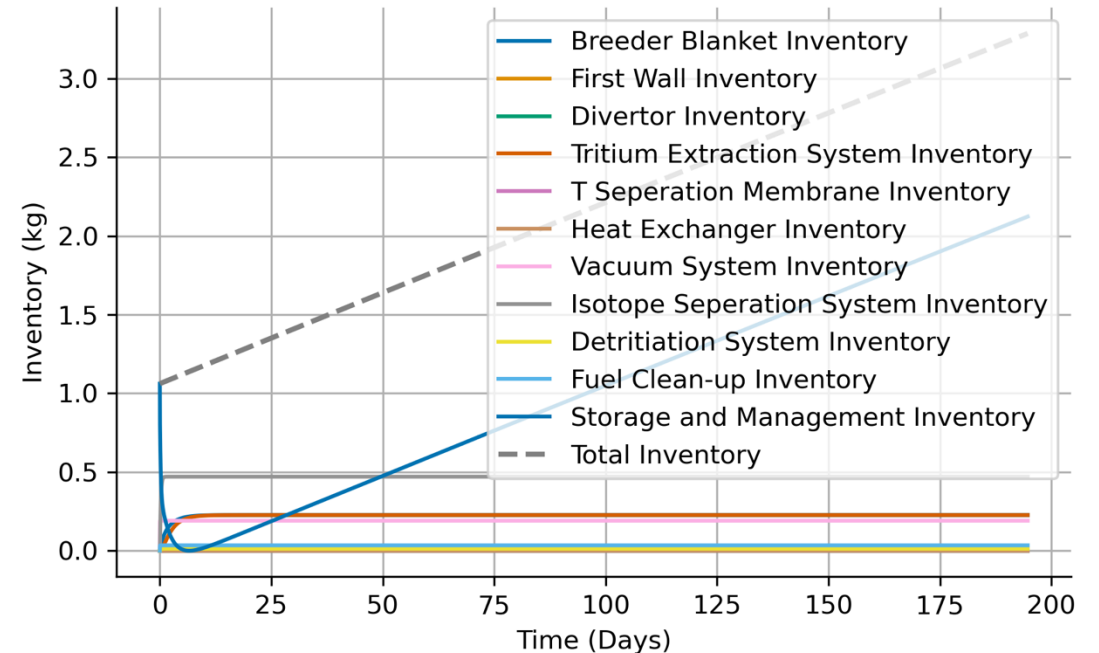


How we explore those

- We can use this model to explore parameters such as required startup inventory for different residence/processing times for different components or for higher level parameters like TBR requirements, or effect of fuelling efficiency
- We can see in this outputs based on an architecture similar to Meschini *et al.* *Modeling and analysis of the tritium fuel cycle for ARC- and STEP-class D-T fusion power plants.* Nuclear Fusion, Vol. 63, 126005 (2023).

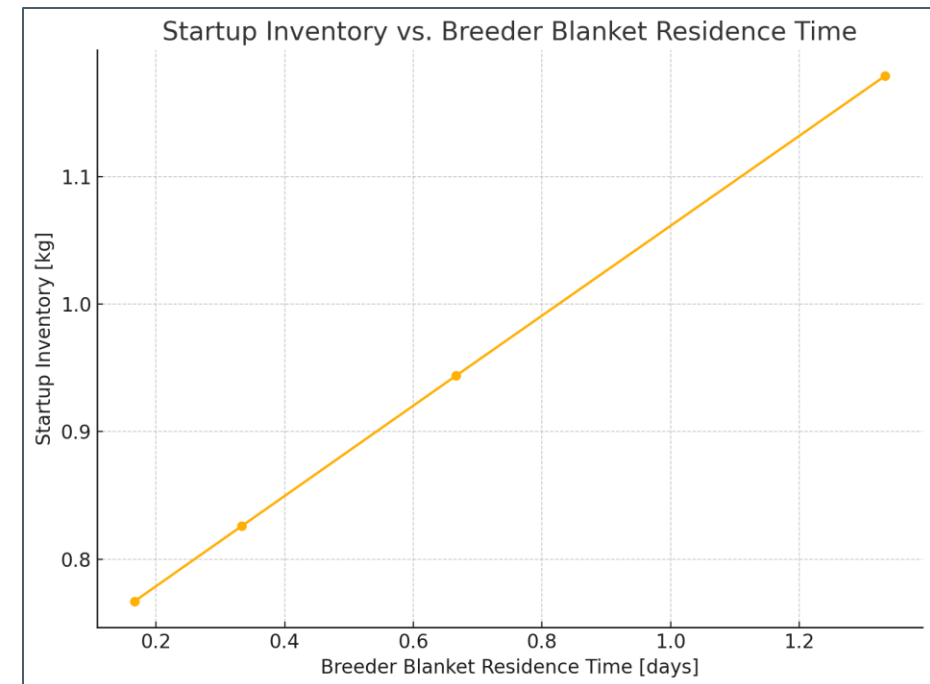
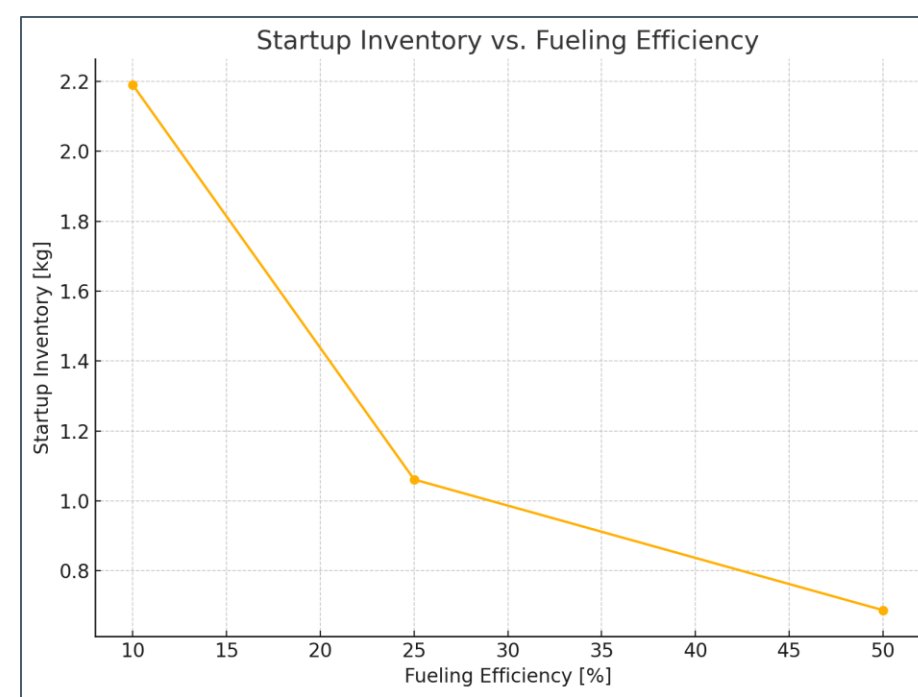


Inventory Evolution in Fuel Cycle Components



How we explore those

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Beyond inventory models

- Inventory models are just the start!
- Closer to realization -> process engineering
 - Models to understand the full flow in and out of all components
 - Greater detail to understand off normal events and scheduled and unscheduled downtime
- Models must better match real world values – building components to validate these models is the only way to do this
- We need to better understand unknowns, e.g.:
 - How much protium build-up will there be – directly affects how much we can recycle fast in DIR
 - In long pulse conditions – what will be retained in plasma-facing material



Tritium Accountancy Systems

- Tritium accountancy is key for two main reasons:
 - Operation – tritium accountancy allows us to understand where fuel is, how efficiently our systems are working and that we are getting the fuel we need where we need it
 - Regulation – for licensing, safety and compliance as tritium is a controlled substance
- Current status:
 - Mass balance audits that are performed over days and are not real time
 - And usually requires modelling to correct for any decay
- Where we need to be
 - Real-time inventory tracking, with minimal uncertainty
 - Need localized accountancy and digital integration with plant monitoring and control
- **ONE SLIDE DOES NOT DO THE IMPORTANCE OF THIS JUSTICE!!**



Who works on the fuel cycle?

What disciplines make up the fuel cycle

- This is an area with a variety of backgrounds
- Chemical Engineers
 - Isotope separation (e.g., cryodistillation, TCAP), Purification systems, catalytic reactors, flow control
- Materials Scientists
 - Permeation barriers, hydrides, getter materials, Compatibility with liquid metals, irradiation effects
- Vacuum Engineers
 - Cryopumps, turbomolecular systems, conductance optimization, Divertor and exhaust system design
- Plasma Physicists
 - Burn fraction modeling, fuelling dynamics, Plasma-surface interaction and exhaust behavior



What disciplines make up the fuel cycle

- This is an area with a variety of backgrounds
- Control System Engineers,
 - Real-time tritium monitoring and automation
- Radiation Chemists
 - Polymer degradation, radiolysis, tritiated compound behavior, Organic contamination and mitigation strategies
- Health Physicists / Safety Specialists
 - Tritium containment and release modelling, Dose assessment, safety cases, environmental protection





Finale!

Key Takeaways

- The fuel cycle looks to purify deuterium and tritium, and get rid of waste exhaust with minimal losses
- Tritium is scarce, so we must be constantly making more in the fusion device
- Hydrogen isotopes mainly behave similarly but the radioactive nature of tritium means it must be carefully controlled, and additional limits must be placed on it
- There are many technologies that make up fuel cycle system, and they use different properties of hydrogen isotopes to manipulate the compositions through different systems
- The challenges of tritium and the fuel cycle are plentiful but there is a variety of people working on it (hopefully including some of you!)



THANK YOU AND ANY QUESTIONS?

CONTACT US:



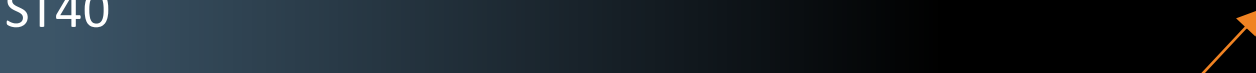
EMRE YILDIRIM

emre.yildirim@tokamakenergy.com



www.tokamakenergy.com

This is what a 100 million °C fusion plasma looks like in ST40

An orange arrow originates from the bottom of the text and points towards the central image of the fusion plasma.