



# Heat Removal for First Wall Components

Lane B. Carasik, Ph.D. – Happy Pride Month!



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# How'd I get here?



Middle Georgia College  
Associates in Math



Getty Images



THE UNIVERSITY OF  
**TENNESSEE**  
KNOXVILLE

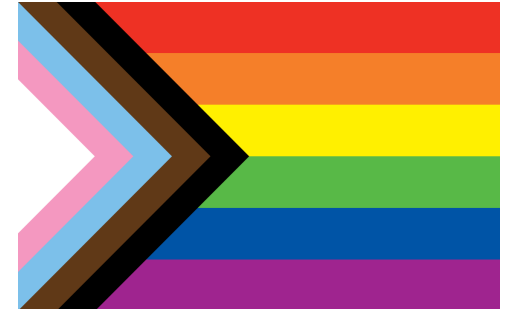
B.S. in Nuclear Engineering  
Undergrad research in  
experimental thermal  
hydraulics



knoxnews



Ph.D. in Nuclear Engineering  
Coursework in Aero, Mech,  
and Civil



# How'd I get here?



Summer REU (NSF CREATE)



THE UNIVERSITY OF  
TENNESSEE  
KNOXVILLE

Undergrad Research



Summer Intern



Summer Intern

Imperial College  
London

Visiting Researcher



Grad Intern



Grad Intern



Kairos Power

CFD Engineer



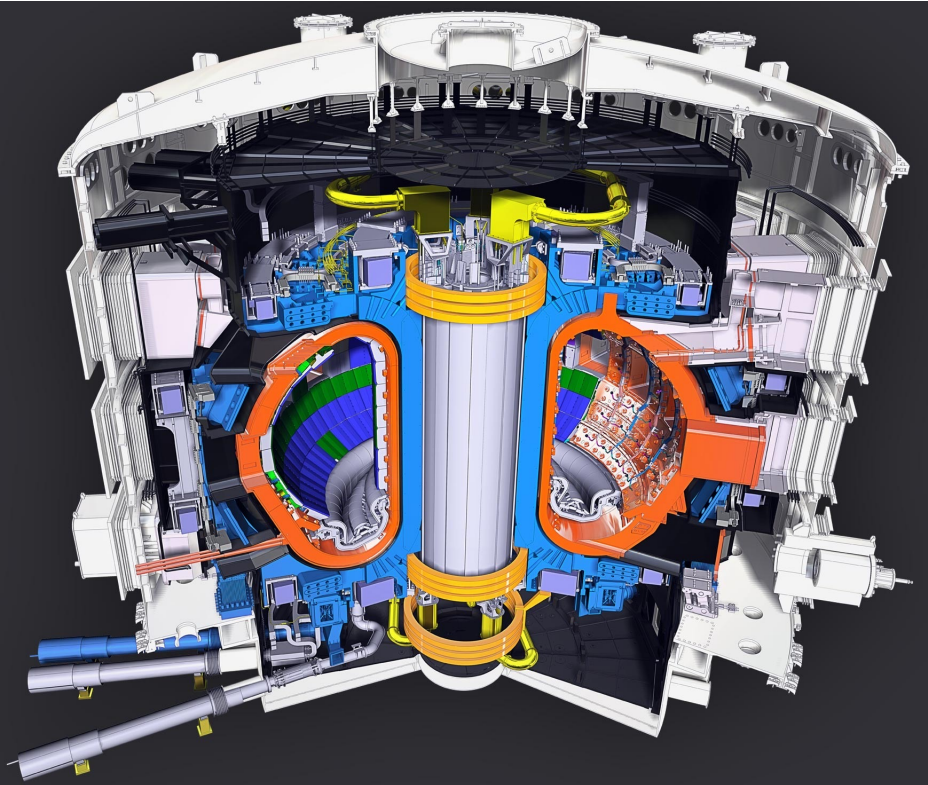
Nuclear Thermal  
Fluids Engineer

# Outline

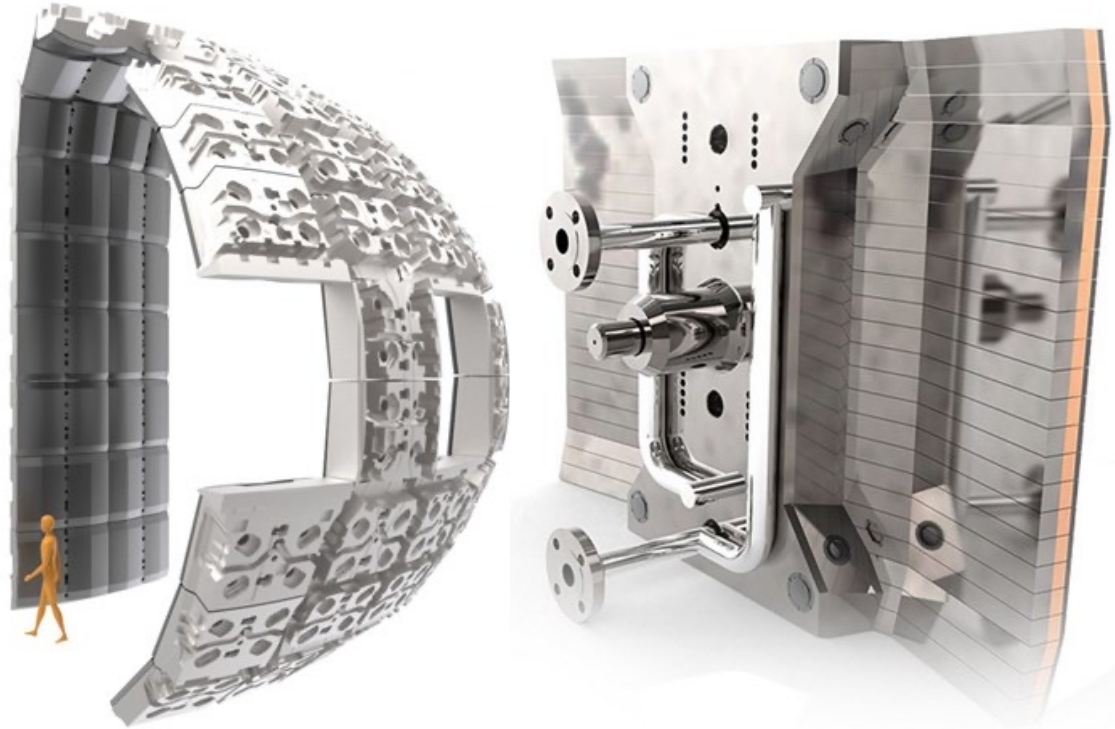
- First Wall Components within Fusion Devices
- System Level Viewpoint
- Thermodynamic Cycles
- Heat Removal Mechanisms



# First Wall Components within Fusion Devices

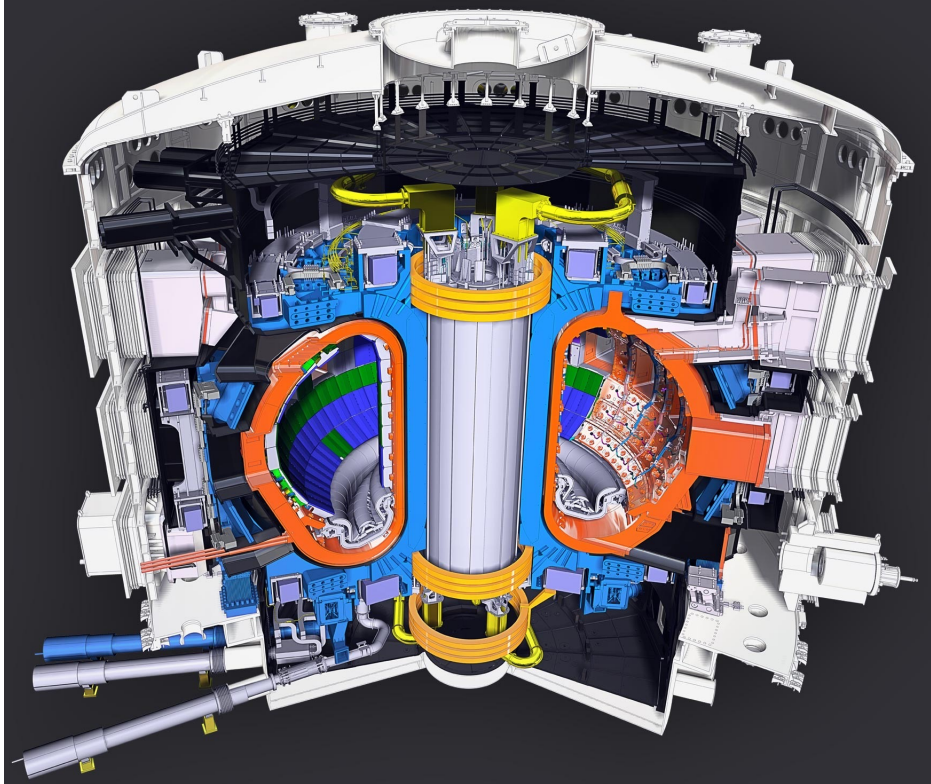


ITER<sup>1</sup>

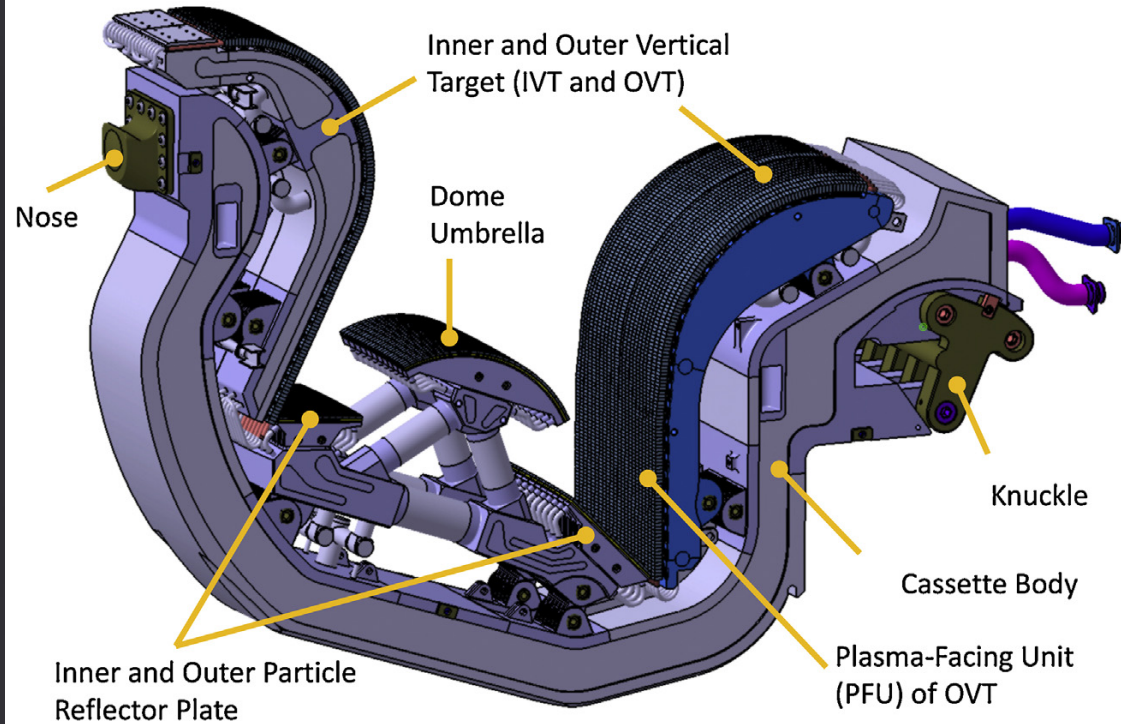


ITER Blanket<sup>1</sup>  
(~4.5 MW/m<sup>2</sup>)

# First Wall Components within Fusion Devices



ITER<sup>1</sup>

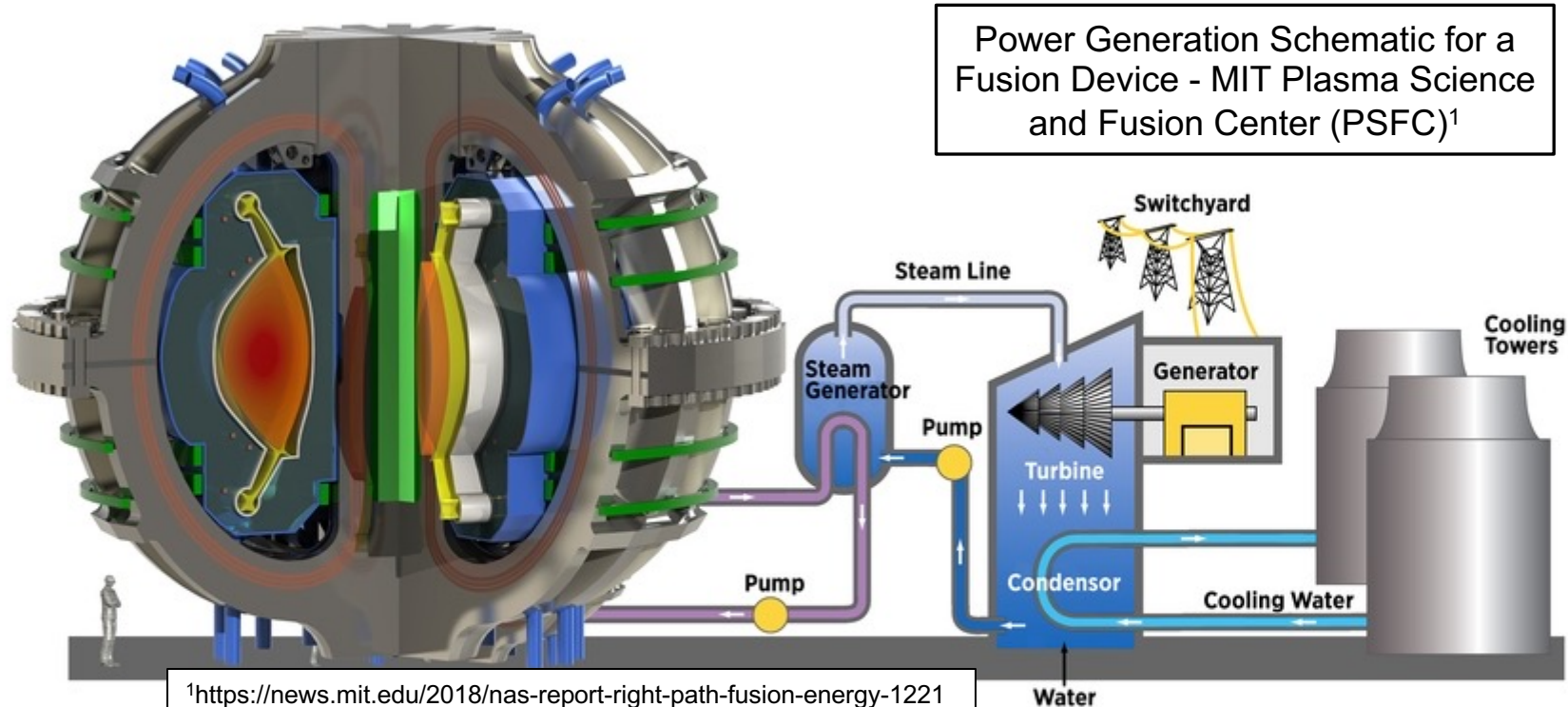


ITER Plasma Facing Component<sup>1</sup>  
Divertor (Up to 20 MW/m<sup>2</sup>)



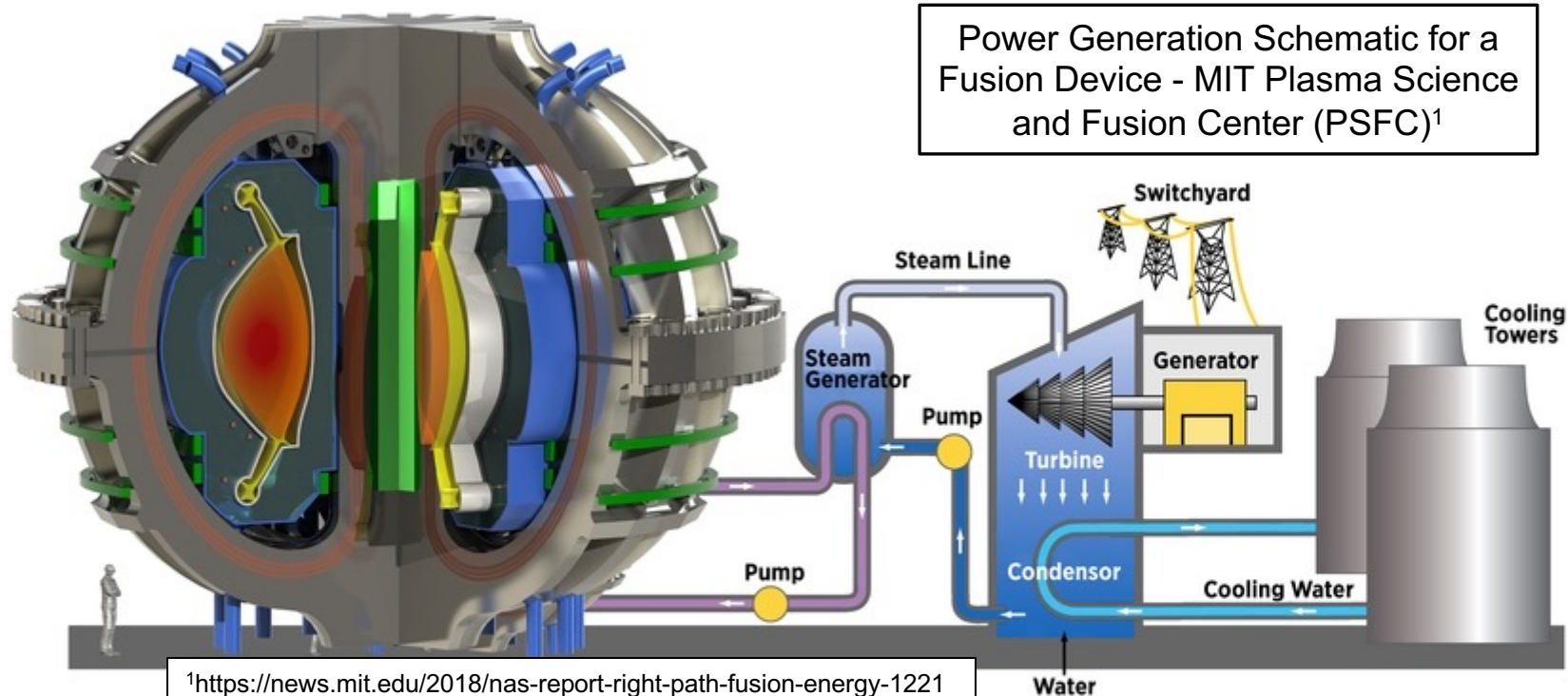
# Why Heat Removal? (System Level)

- Cooling of components to prevent degradation and damage
- Why use fusion devices at all?
  - Power generation



# System Level – Power Cycles

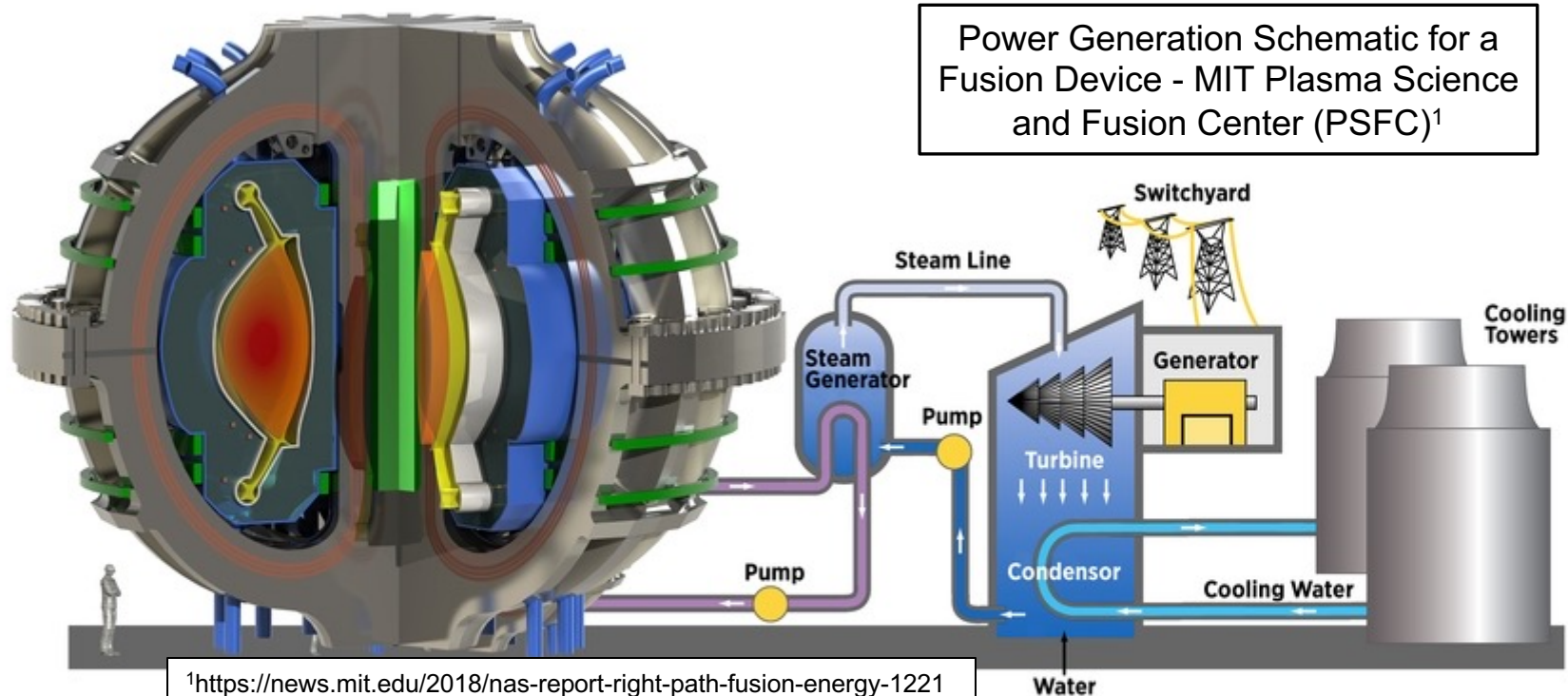
- Based on a Rankine power cycle that is built upon centuries of experience in the power industry





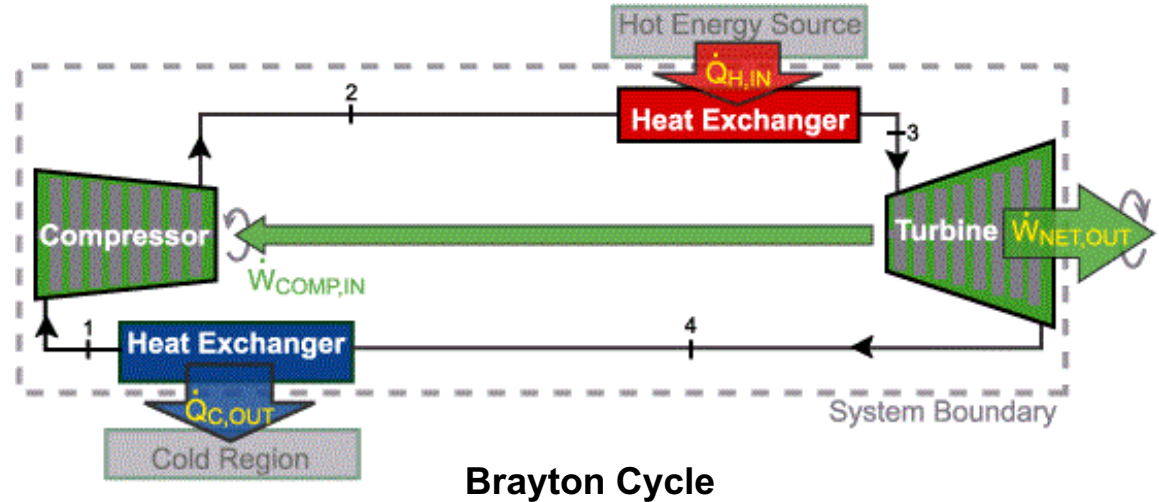
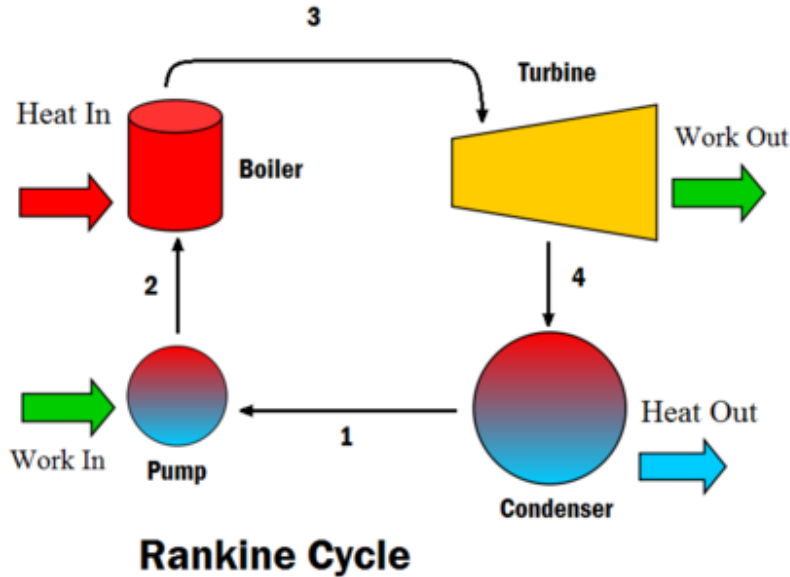
# System Level – Power Cycles

- Regardless of heat source, most power generation results in the same thing, boiling water to spin a turbine that causes a generator to create electricity.



# System Level – Power Cycles

- There are two common forms of power cycles that could be used:
  - Rankine (Liquid - water based)
  - Brayton (Gas – helium or others based)



<sup>1</sup>[https://energyeducation.ca/encyclopedia/Rankine\\_cycle](https://energyeducation.ca/encyclopedia/Rankine_cycle)

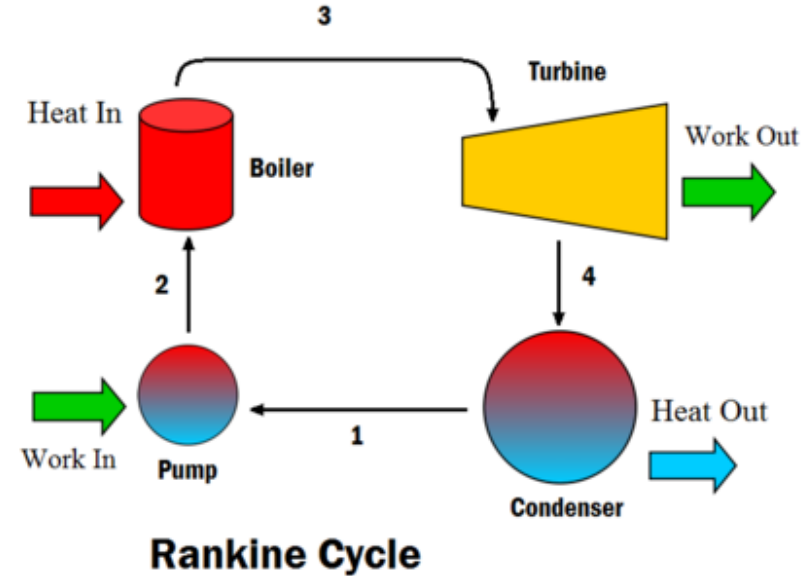
<sup>2</sup>[https://energyeducation.ca/encyclopedia/Brayton\\_cycle](https://energyeducation.ca/encyclopedia/Brayton_cycle)

# System Level – Power Cycles

- Both cycles can be made more complex with reheating, regeneration, multiple stages, to maximize the amount of useful energy out vs. heat in

$$\eta = \frac{\text{Work Out}}{\text{Heat In}} = \frac{W}{Q_H} = \frac{Q_H - Q_C}{Q_H} \text{ or } = \frac{T_H - T_C}{T_H}$$

- In this case, the  $T_H$  or  $T_C$  are the hottest and coldest reservoir temperatures.

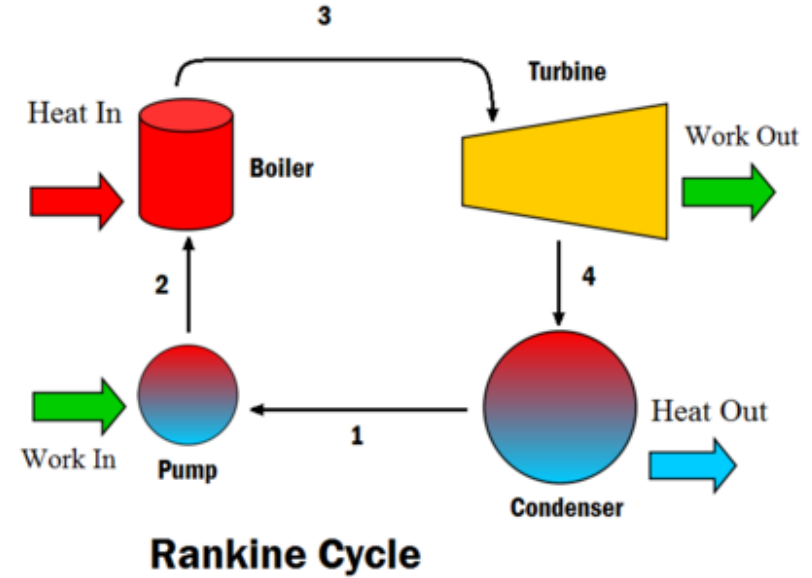


# System Level – Power Cycles

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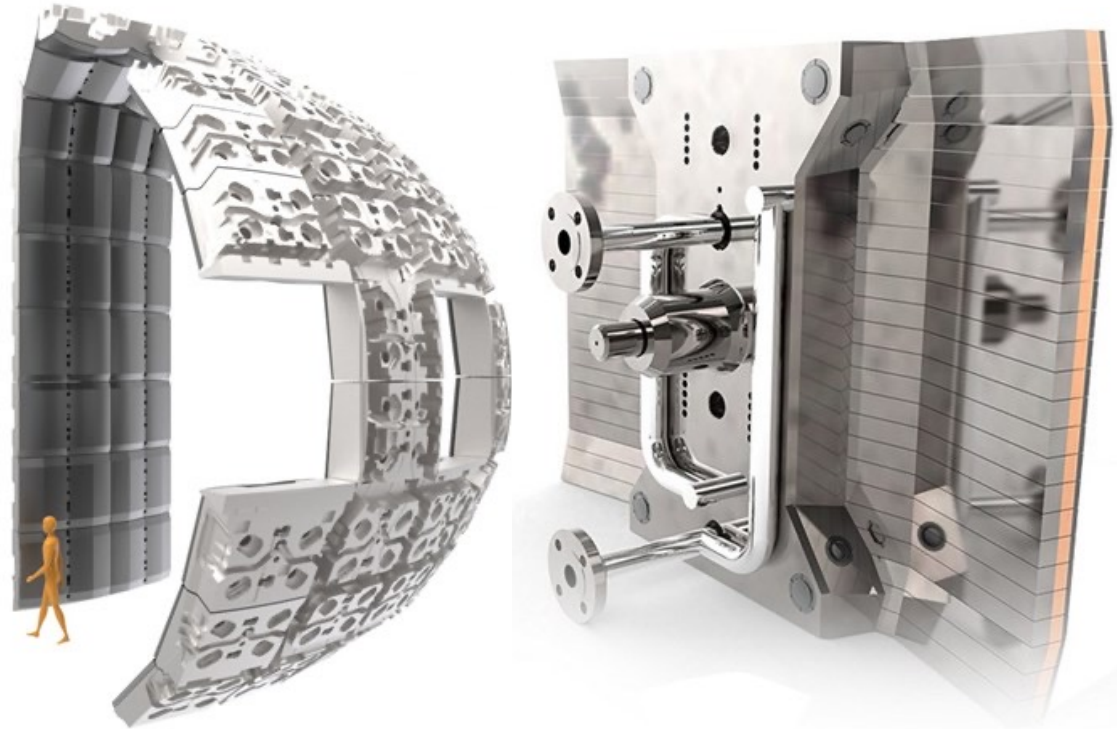
- We have a few options, lower our final heat sink temperature, raise our operating temperature, but never can reach ideal efficiencies.





# System Level – Heat Removal

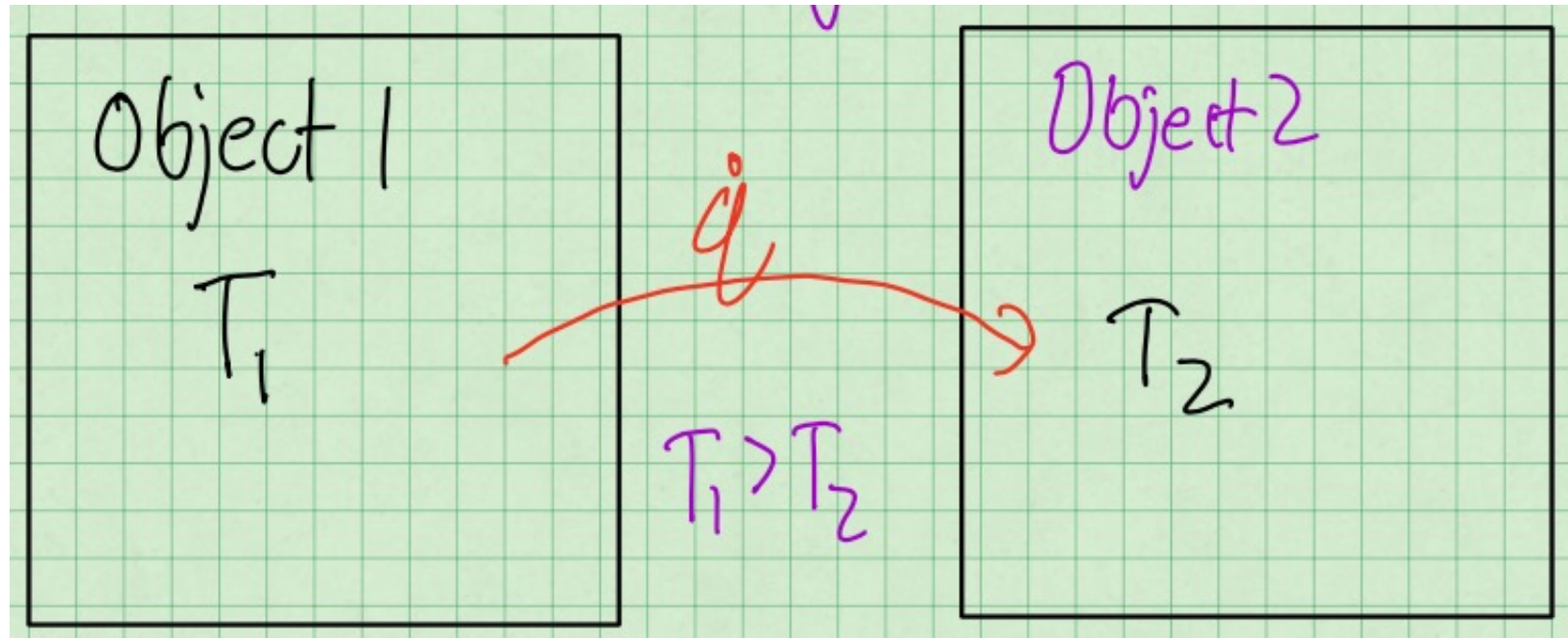
- A critical aspect of heat removal is ensuring we understand the heat transfer mechanisms that control our power cycle capabilities, material temperature limits, and instrumentation/control schemes.



ITER Blanket<sup>1</sup>  
(~4.5 MW/m<sup>2</sup>)

# Heat Transfer Mechanisms

- Heat Transfer occurs whenever there is a temperature difference between two objects



# Heat Transfer Mechanisms

- Heat Transfer occurs in:
  - Solids – Gases – Fluids – Plasmas - Others?
  - As a continuum (not individual particles)

Very General Conservation of Energy Equation

Enthalpy Formulation

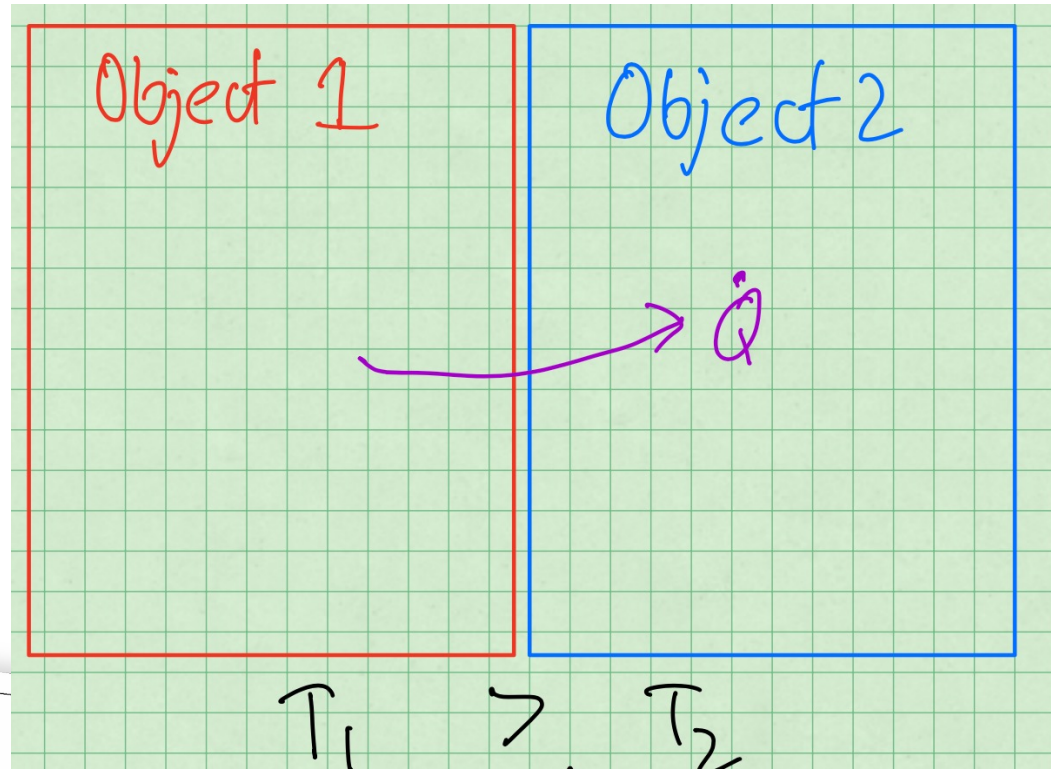
$$\rho \frac{Dh}{Dt} = \nabla \cdot (k \nabla T) + \dot{q}''' + \frac{DP}{Dt} + \mu \Phi$$

Temperature Formulation

$$\rho c_p \frac{DT}{Dt} = \nabla \cdot (k \nabla T) + \dot{q}''' + \beta T \frac{DP}{Dt} + \mu \Phi$$

# Heat Transfer Mechanisms

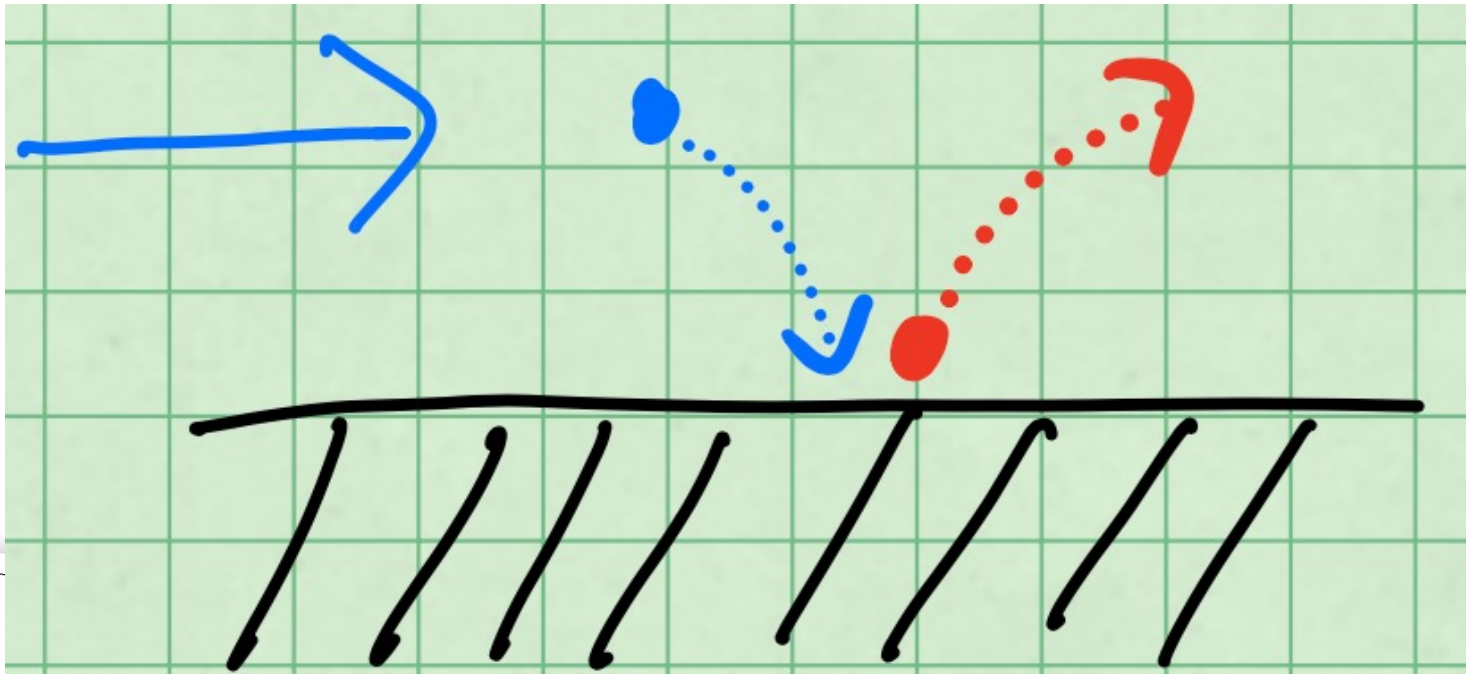
- Heat Conduction (Conductive Heat Transfer)+
  - Can occur within an object and between objects in contact
  - Described using diffusion operator ( $\nabla k \nabla T$ ) in the energy equation





# Heat Transfer Mechanisms

- Heat Convection (Convective Heat Transfer)
  - Involves a fluid (liquid or gas) that is moving due to mechanical or buoyant forces
  - Described using the advective operator ( $\vec{V} \cdot \nabla T$  or  $\vec{V} \cdot \nabla h$ )



# Heat Transfer Mechanisms

- Heat Convection (Convective Heat Transfer)
  - Two forms (Free/Natural and Forced Convection)

$$\dot{Q}_{conv.} = HTC(T_{wall} - T_{bulk})SA$$

$$HTC = \frac{kNu}{D_h}$$

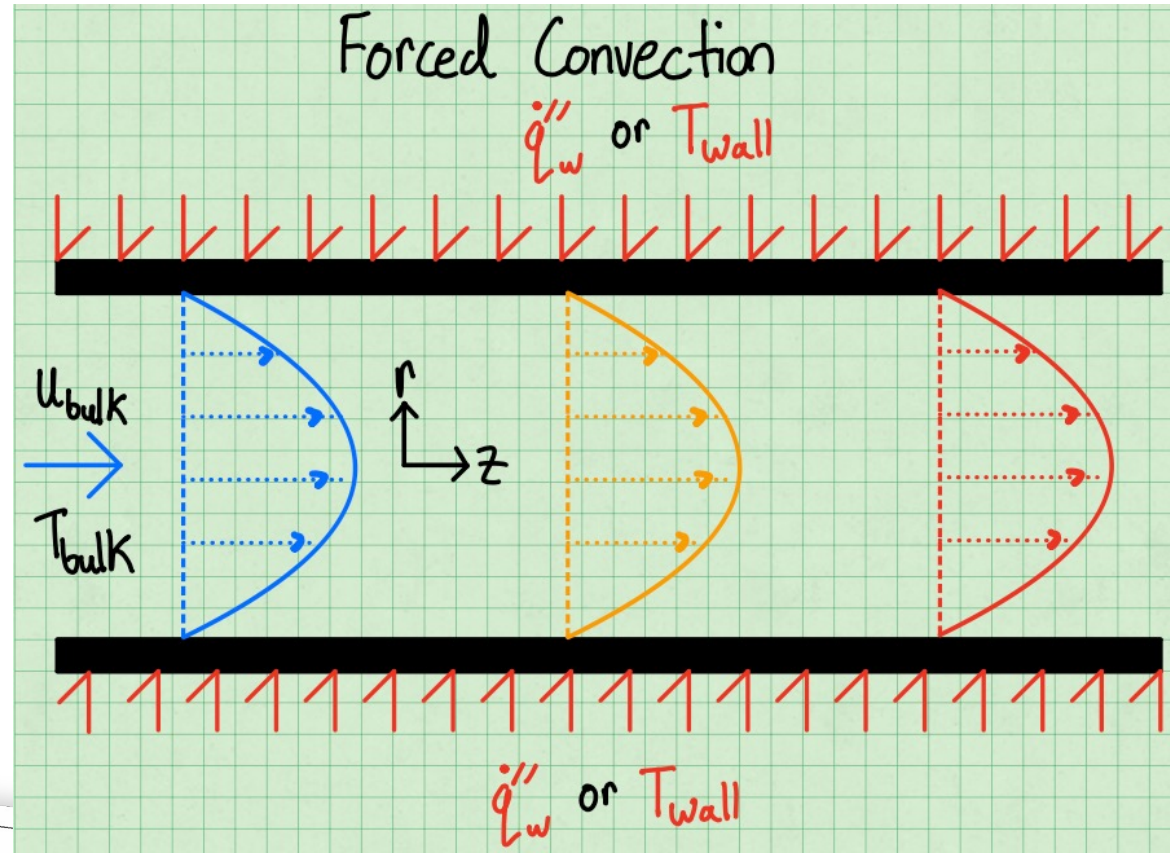
$HTC$  – Heat Transfer Coefficient

$T_{wall}$  – Wall Temperature ( $^{\circ}C$ )

$T_{bulk}$  – Bulk Fluid Temperature ( $^{\circ}C$ )

$SA$  – Surface Area ( $m^2$ )

$Nu$  – Nondimensional  
Heat Transfer Coefficient or  
Nusselt Number



# Heat Transfer Mechanisms

- Heat Convection (Convective Heat Transfer)
  - Two forms (Free/Natural and Forced Convection)

$$\dot{Q}_{conv.} = HTC(T_{wall} - T_{\infty})SA$$

$HTC$  – Heat Transfer Coefficient

$$HTC = \frac{kNu}{L_c}$$

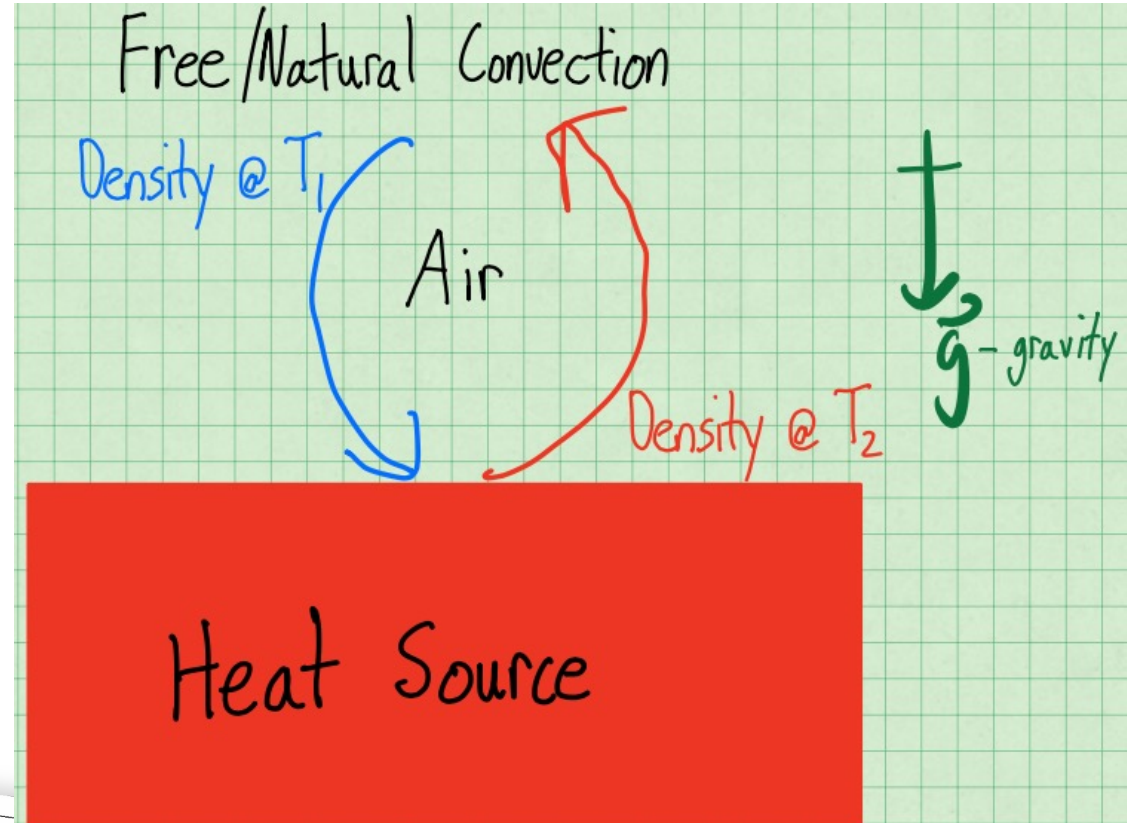
$T_{wall}$  – Wall Temperature ( $^{\circ}C$ )

$T_{\infty}$  – Ambient Temperature ( $^{\circ}C$ )

$SA$  – Surface Area ( $m^2$ )

$Nu$  – Nondimensional

Heat Transfer Coefficient or  
Nusselt Number



# Heat Transfer Mechanisms

- Heat Radiation (Irradiative Heat Transfer)

$$\dot{Q}_{rad} = \varepsilon \sigma (T_{Hot}^4 - T_{Cold}^4) SA$$

$\varepsilon$  – Emissivity (–)

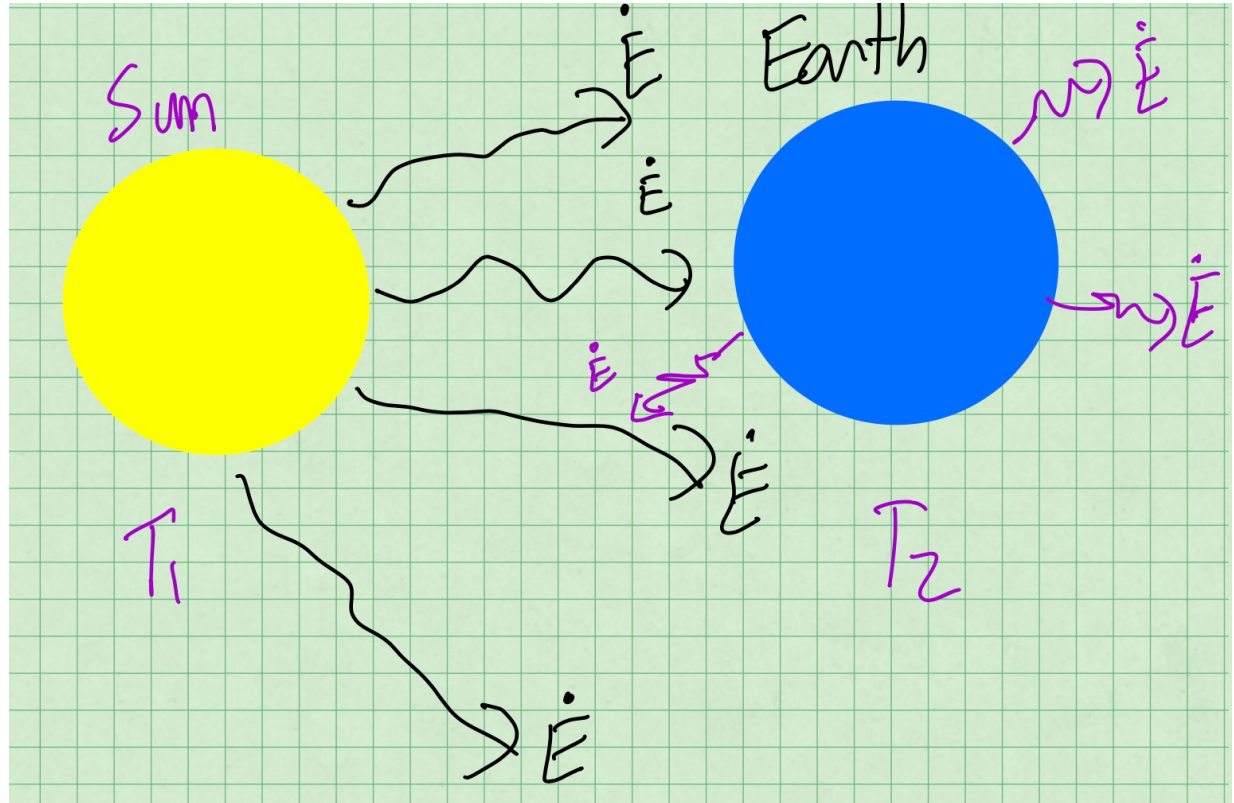
$\sigma$  – Stefan – Boltzmann  
Constant

$$= 5.6703 \times 10^{-8} \frac{W}{m^2 \cdot K^4}$$

$T_{Hot}$  – Hotter Surface (K)

$T_{Cold}$  – Colder Surface (K)

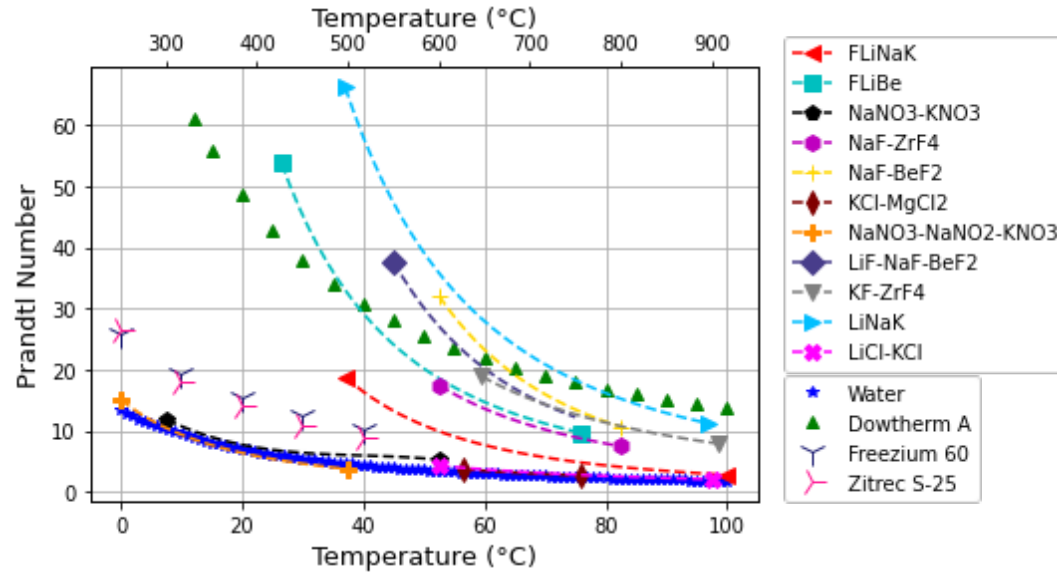
$SA$  – Surface Area ( $m^2$ )





# Scaling and Similitude for Fusion Device Design

We take advantage of surrogate fluids for molten salts, liquid metals, and gases to do scaled heat transfer and fluid dynamics experiments.



Relevant scaling parameters/non-dimensional numbers:

$$Pr = \frac{\mu c_p}{k}$$

$$Re = \frac{\rho u D_h}{\mu}$$

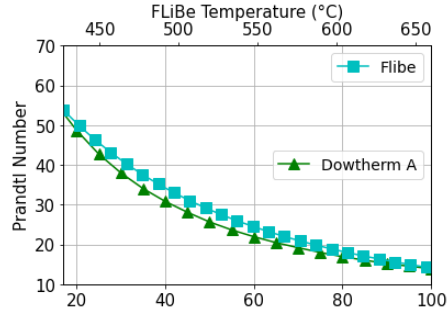
Desired Heat Transfer/Fluid Dynamics Design Information:

$$HTC = f(Re, Pr, Geom, BCs)$$

$$dP = f(Re, Geom)$$

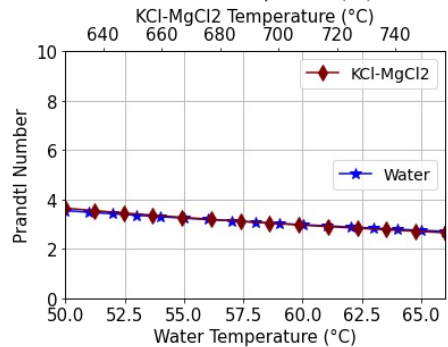
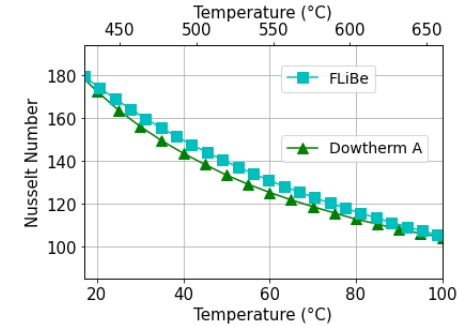


# Similitude of Heat Transfer



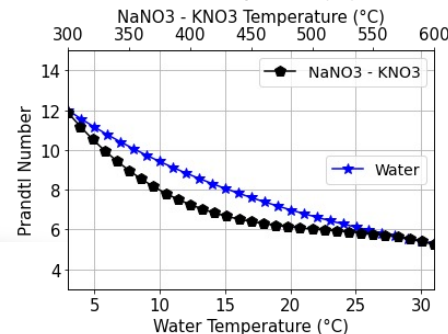
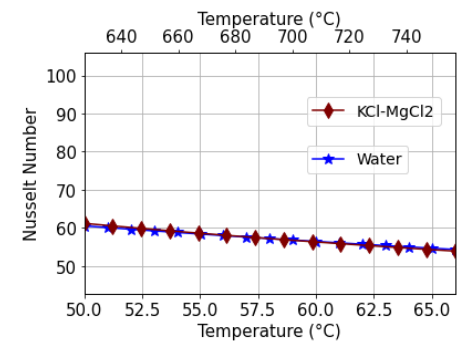
Fluorides:

$$Re = \frac{\rho u D_h}{\mu}$$



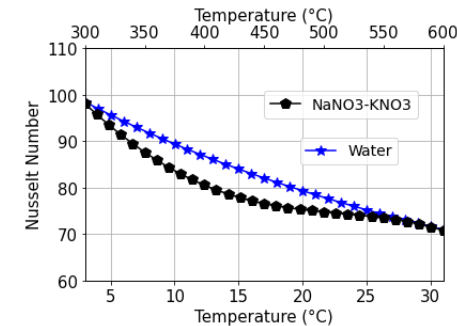
Chlorides:

$$Pr = \frac{\mu c_p}{k}$$



Nitrates:

$$HTC = f(Re, Pr, Geom, BCs)$$



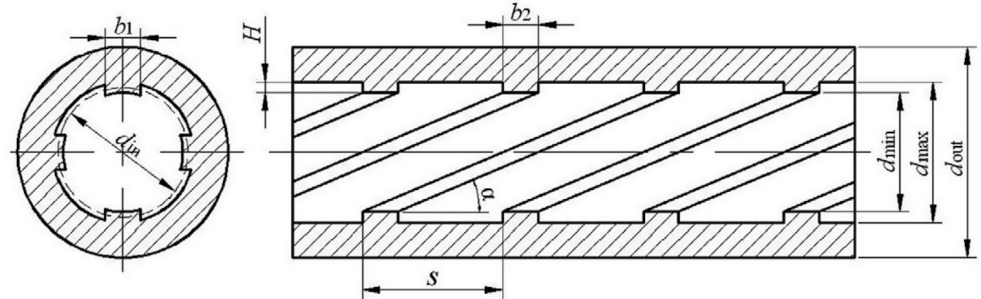
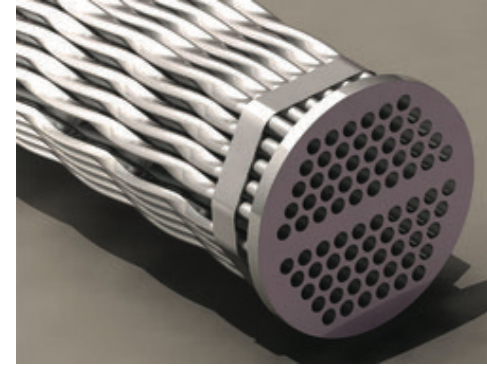
# Types of Heat Transfer Enhancements

Different forms of heat transfer enhancements exists:

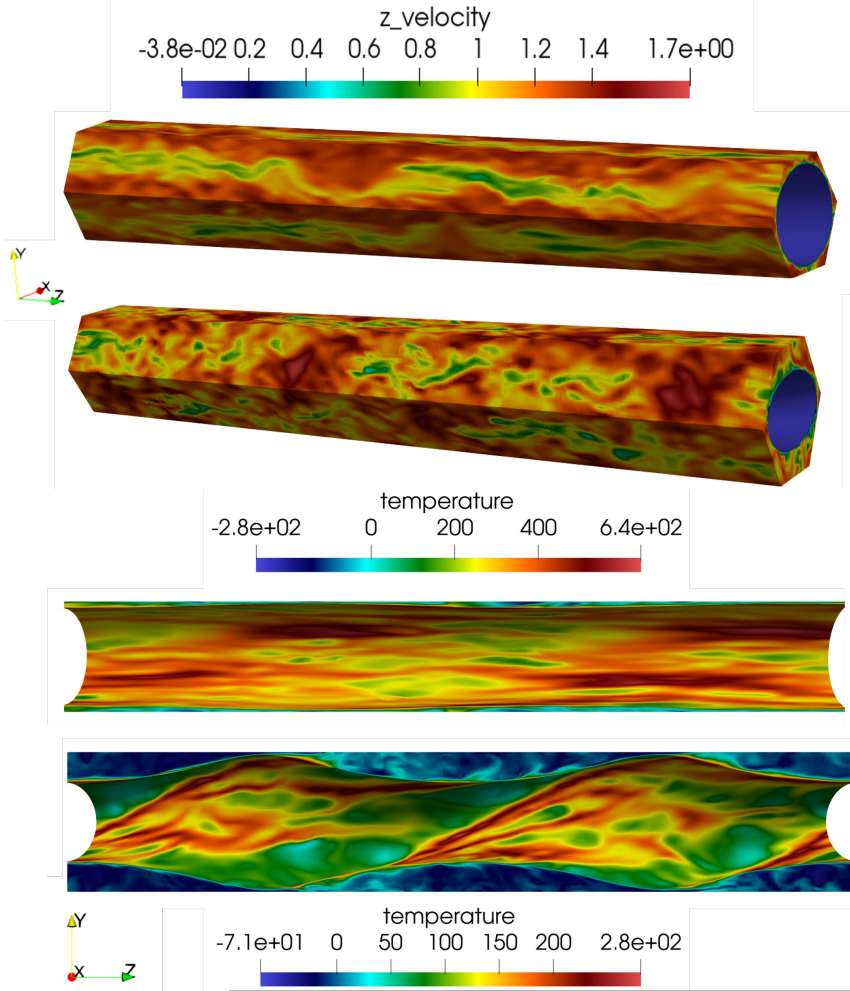
- Swirl flow inserts
- Internal fins
- Rifling/grooves
- Others

Performance quantified by the  
Thermal Performance Factor

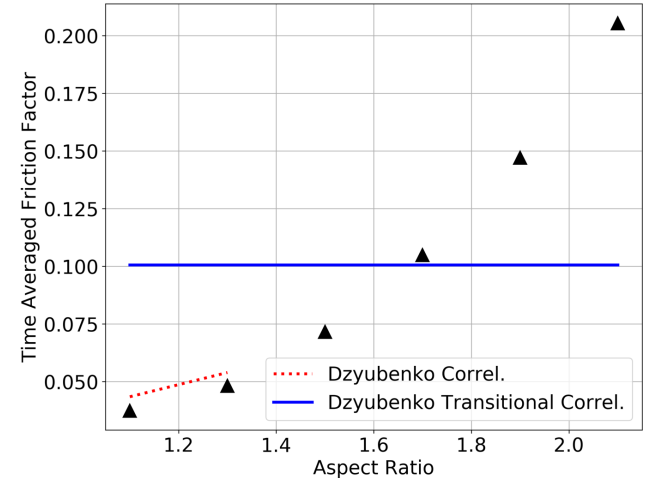
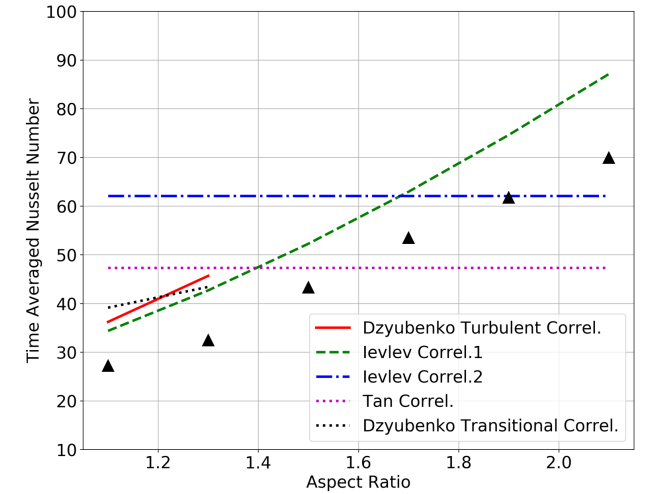
$$T.P.F. = \eta = \frac{\left(\frac{Nu}{Nu_0}\right)}{\left(\frac{f}{f_0}\right)^{1/3}}$$



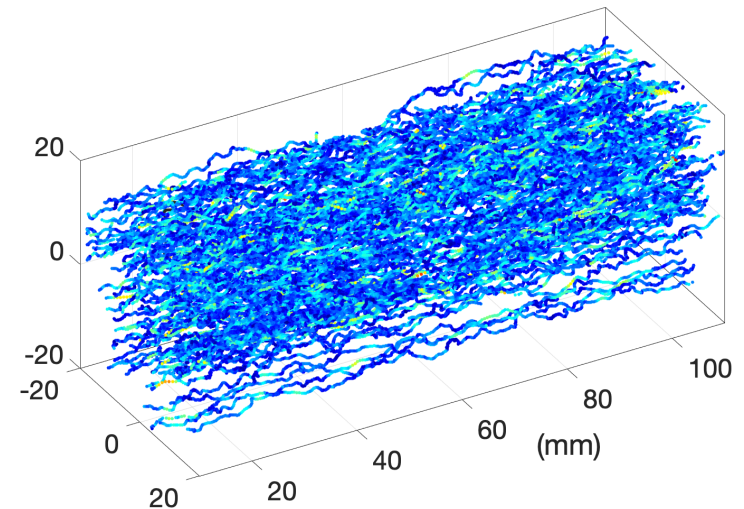
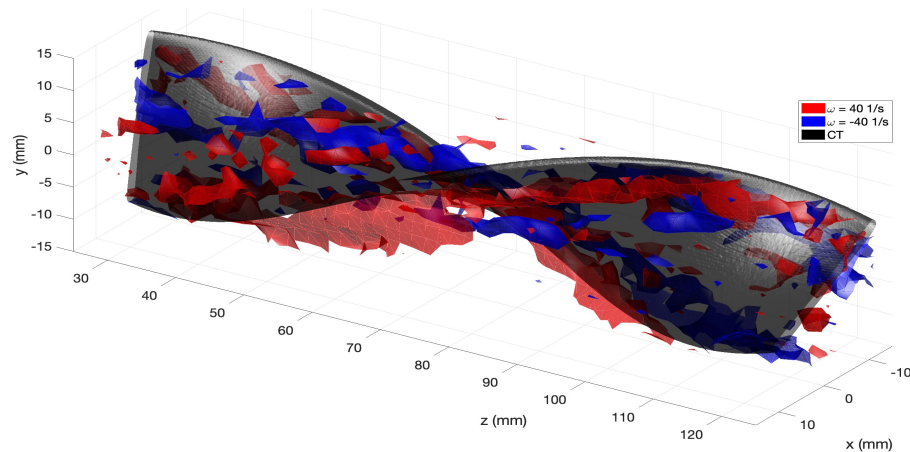
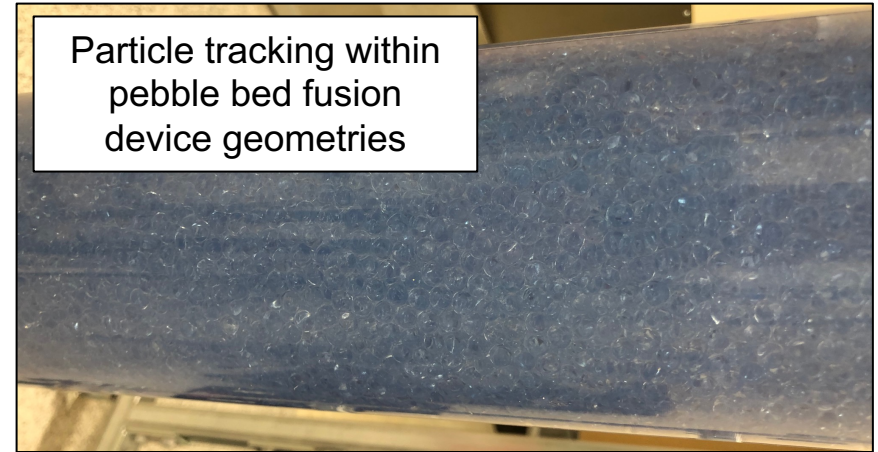
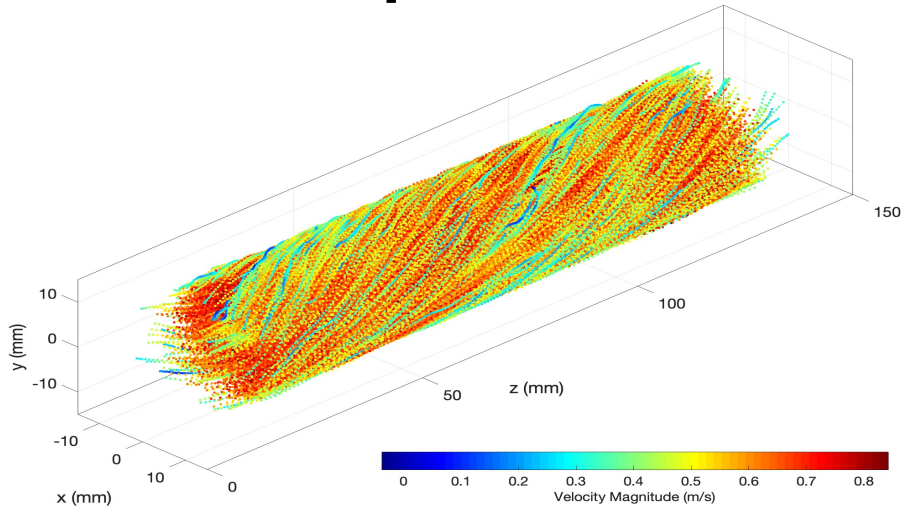
# Computational Fluid Dynamics



**Data Reduction  
(Triangles are  
Simulations)**



# Experimental Fluid Dynamics





# Acknowledgements

Thanks to Dr. Arturo Dominguez, PFURO, DOE FES, US DOE Office of Science, and more for this opportunity!

# Thank you!



# Additional Resources

## *Heat Transfer Lectures:*

[https://www.youtube.com/playlist?list=PLZOZfX\\_TaWAHZOgn8CRjpqRElp5Dd-GaY](https://www.youtube.com/playlist?list=PLZOZfX_TaWAHZOgn8CRjpqRElp5Dd-GaY)

## *Fluid Mechanics Lectures:*

[https://www.youtube.com/watch?v=clVwKynHpB0&list=PLZOZfX\\_TaWAGocs2k5QmTL44OKOI7rn34&ab\\_channel=CPPMechEngTutorials](https://www.youtube.com/watch?v=clVwKynHpB0&list=PLZOZfX_TaWAGocs2k5QmTL44OKOI7rn34&ab_channel=CPPMechEngTutorials)

[https://www.youtube.com/watch?v=PXjZ7xEAqsU&list=PLZOZfX\\_TaWAH0baRhA8OosWVbEsJK5sPe&ab\\_channel=CPPMechEngTutorials](https://www.youtube.com/watch?v=PXjZ7xEAqsU&list=PLZOZfX_TaWAH0baRhA8OosWVbEsJK5sPe&ab_channel=CPPMechEngTutorials)

[https://www.youtube.com/watch?v=kxhTMc8tyEo&list=PLZOZfX\\_TaWAE7uM59dIBr-rH73WTJCcp\\_&ab\\_channel=CPPMechEngTutorials](https://www.youtube.com/watch?v=kxhTMc8tyEo&list=PLZOZfX_TaWAE7uM59dIBr-rH73WTJCcp_&ab_channel=CPPMechEngTutorials)