Low Temperature Plasmas

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Introduction to Fusion Energy
and Plasma Physics Summer Course
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Day 9....on Zoom!
How did plasma physics find me?

• Originally from northern Wisconsin
  • Like to build things -> engineering school

• University of Wisconsin
  • Nuclear Engineering/Engineering Physics (BS 2006, MS 2008, PhD 2010)
  • Undergrad RA in Prof. Hershkowitz’s lab
  • Caught the plasma bug, never looked back

• Postdoctoral training
  • University of New Hampshire (2010-2012)
  • Los Alamos National Laboratory (2012-2013)

• University of Iowa Physics and Astronomy

Prof. Noah Hershkowitz
Irving Langmuir Professor (emeritus)
University of Wisconsin
Why plasma physics and engineering?

1. The thrill of “finding things out”

2. Can be both fundamental and practical
   • Fundamental questions: matter in extreme conditions, nonlinear dynamics, complexity, self-organization, etc., etc.
   • Translational: Fundamental discovery often leads to new devices on a short timescale
   • Contributes to solving pressing problems of society
   • This is especially true in Low Temperature Plasmas
Outline of this talk

• What is a low temperature plasma?
• What are they used for?
• Historical context
• Basic properties
  • Plasma potential
  • Sheaths
  • Plasma chemistry
What is a low temperature plasma?
Low temperature?
I thought plasmas were hot

- Low temperature plasmas (LTP) are roughly defined by:
  - Ion temperature: \( T_i \approx 0.03 \text{eV} \) (near room temp)
  - Electron temperature: \( T_e \approx 1 - 100 \text{eV} \) (few-to-several eV)

- **Highly non-equilibrium**
  - “Temperature” refers to a mean kinetic energy, rather than a thermodynamic temperature

- Plasma density can vary widely: \( n \approx 10^6 - 10^{16} \text{cm}^{-3} \)
- Ionization fraction is usually low: \( \frac{n_p}{n_n} < 1\% \)

How can electron and ion “temperatures” be so different?
Ions are much more massive than electrons

• Electron mass: \( m_e = 9.1 \times 10^{-31} \text{kg} \)
• Proton mass: \( m_i = 1.7 \times 10^{-27} \text{kg} \)
• Ratio: \( \frac{m_i}{m_e} \approx 1836 \)

About the same as:

• Ping pong ball: \( m_{pp} = 2.7 \text{ g} \)
• Bowling ball: \( m_{bb} = 5 \text{ kg} \)
• Ratio: \( \frac{m_{bb}}{m_{pp}} \approx 1850 \)
Electrons move much faster than ions

- Thermal speed: \( v_{T\alpha} = \sqrt{\frac{2k_BT\alpha}{m_\alpha}} \)

- Ratio: \( \frac{v_{Te}}{v_{Ti}} = \sqrt{\frac{m_iT_e}{m_eT_i}} \approx 250 \) (uses \( T_e = 1\text{eV} \))

About the same as

- A person walking: 3 miles/hr
- The speed of sound: 760 miles/hr

- Ratio: \( \frac{v_s}{v_w} \approx \frac{760}{3} \approx 250 \)
Equilibration rate depends on the mass ratio

- Reference collision rate (electron-electron)
  \[ \nu_0^e (s^{-1}) \approx \frac{5 \times 10^{-11} n (m^{-3})}{[T(eV)]^{3/2}} \]

- Mean free path: \( \lambda_{\varepsilon}^{\alpha \beta} = \frac{\nu_{T \alpha}}{\nu_{\varepsilon}^{\alpha \beta}} \)

- Time (or distance) to come to same temperature
  \[ \nu_{\varepsilon}^{ee} \approx \nu_0^e \]
  \[ \nu_{\varepsilon}^{ei} \approx \frac{m_e}{m_i} \nu_0^e \ll \nu_{\varepsilon}^{ee} \]
  \[ \nu_{\varepsilon}^{ii} \approx \sqrt{\frac{m_e}{m_i}} \left( \frac{T_e}{T_i} \right)^{3/2} \]
  \[ \nu_0^e \approx \nu_{\varepsilon}^{ee} \]
  \[ \lambda_{\varepsilon}^{ee} (m) \approx 1 \times 10^{16} \frac{[T_e(eV)]^2}{n(m^{-3})} \]
  \[ \lambda_{\varepsilon}^{ei} \approx \frac{m_i}{m_e} \lambda_{\varepsilon}^{ee} \gg \lambda_{\varepsilon}^{ee} \]
  \[ \lambda_{\varepsilon}^{ii} \approx \left( \frac{T_i}{T_e} \right)^2 \lambda_{\varepsilon}^{ee} \ll \lambda_{\varepsilon}^{ee} \]
Electrons escape the plasma before they come into equilibrium with ions

- Example: $T_e = 1\, eV$, $T_i = 0.03\, eV$, $n = 1 \times 10^{16} \, m^{-3}$, $L \approx 1 \, m$, protons
- Mean free paths:
  \[
  \lambda_{ee} \approx 1 \, m \quad \text{Marginal e-e equilibration}
  \]
  \[
  \lambda_{ei} \approx 2000 \, m \quad \text{No e-i equilibration}
  \]
  \[
  \lambda_{ii} \approx 0.001 \, m \quad \text{Excellent i-i equilibration}
  \]
- Ions equilibrate with the neutrals
  - Similar to ion-ion mean free path
  - Details depend on pressure

\[
\lambda_{ei} \gg L
\]
\[
\lambda_{ee} \sim L
\]
\[
\lambda_{ii} \sim \lambda_{in} \ll L
\]
LTPs are (relatively) easy to make

• Power source couples energy to electrons  
  • Many ways to do this (a few examples to come)

• Energetic electrons partially ionize gas  
  • Ionization energies ~10s eV

• A simple example is an incandescent light bulb  
  • Electrons emitted from a hot filament

• Unlike fusion plasmas, LTPs are typically not “confined” by a magnetic field  
  • Electrons and ions are lost to boundaries

Image from: https://www.1000bulbs.com
dc filament-driven discharges

- Heat material that emits electrons when hot
  - Thermionic emission
  - Filament
- dc bias the filament with respect to the plasma
  - Electrons gain energy from the electric field between the filament and plasma
- Energetic electrons ionize gas

Images from Dr. Ryan Hood, U Iowa
dc electric field-driven discharges

- Electric field accelerates electrons to high energy
  - Electrons ionize neutrals
- Examples
  - Arcs and sparks
  - Glow discharges

Image from:
https://commons.wikimedia.org/wiki/File:Glow_discharge_regions.jpg

Image from:
https://upload.wikimedia.org/wikipedia/commons/e/eb/Staccoto_Lightning.jpg
ac capacitively coupled discharges

- Electrodes biased with ac voltage
  - MHz frequency electrostatically couples power to electrons
  - Advantageous because electric fields are confined to edges (sheaths)

Figure from: Pattyn, Kovacevic, Strunskus, Lecas and Bernt, Scientific Reports 6, 10886 (2019).
ac inductively coupled discharges

• Antenna biased with ac voltage
  • 10s of MHz to couple with electrons
• Energy coupled by electromagnetic induction
  • ac current in the coils induces an ac magnetic field inside the tube
  • produces an electromotive force (electric field) inside the plasma
  • Energetic electrons ionize neutrals

Image from: https://www.trumpf.com

Plus, many more ways to make LTPs
Low temperature plasmas are tremendously useful.
Utilize directed energy of charged particles

• Electric fields accelerate ions (or electrons) in a preferred direction

• Directed kinetic energy is used to do useful things
  • Etching semiconductors
    • >$1 T industry!
  • Surface modification of materials
  • Plasma-based propulsion
  • Ion and electron sources
  • Sources of fusion neutrons
    • neutron imaging
    • producing medical isotopes
    • (not for net energy production)
Non-equilibrium (high temperature electrons) enables unique chemistry

- “Normal” chemistry only accesses reactions near room temperature
  - New reactions at eV e\(^{-}\) energy
  - Plasma catalysis
- Many applications
  - Materials processing
  - Climate
  - Chemical production
  - Medicine
  - Energy
  - Agriculture

Figure from: Zhang et al, “Plasma activation of methane for hydrogen production in a N\(_2\) rotating gliding arc warm plasma: A chemical kinetics study,” Chemical Engineering Journal 345, 67 (2018).
Science Challenges in Low-Temperature Plasma Science and Engineering:

Enabling a Future Based on Electricity Through Non-Equilibrium Plasma Chemistry

Workshop held at the National Science Foundation | Arlington, VA | 22-23 August 2016

Historical context
Joseph J. Thomson (1856-1940)

- Cavendish Professor of Physics, Cambridge
- Discovered
  - electrons, isotopes, Thomson scattering, plum pudding model, mass spectrometer, waveguides
- Renowned educator
  - Advised 6 physics Nobel laureates and 2 chemistry Nobel laureates
  - Ex: Rutherford, Bohr, Born, Oppenheimer, Bragg

(1906) for work on the conduction of electricity in gasses
Thomson discovered the electron

- Needed high vacuum
  - otherwise a plasma would form!
- Cathode rays were influenced by electric and magnetic fields
  - Measured charge-to-mass ratio
- Understanding collisions was essential
  - Showed size is 1000x smaller than the atom (first subatomic particle)
- Plum pudding model of the atom
  - corpuscles emanating in a sea of positive charge

From: wikipedia.org
Irving Langmuir (1881-1957)

• Industrial physicist at General Electric
• Founded
  – Plasma physics
  – Surface chemistry
  – Atmospheric control (cloud seeding)
• Invented gas-filled incandescent lamp, Langmuir probe diagnostic, Langmuir (plasma) wave, hydrogen welding

1932: For his discoveries and inventions in surface chemistry
Langmuir named “plasma” and “sheath”


“Except near the electrodes, where there are sheaths containing very few electrons, the ionized gas contains ions and electrons in about equal numbers, so that the resultant space charge is very small. We shall use the name plasma to describe this region containing balanced charges of ions and electrons.”

Langmuir named “plasma” and “sheath”

History of “Plasmas”

Sir,—There has been speculation as to the basis for Irving Langmuir’s use of the word “plasma” to describe a particular region of an electrical discharge in a gas. I was working for Langmuir when he made this innovation about 1927, and, much later, described it in a letter to a friend at the General Electric Research and Development Center, dated April 20, 1967. The pertinent extract from this letter follows.

“Meanwhile word had come down from on high that we had better do something about mercury-arc rectifiers and converters, which were the big new things in the electrical engineering world, although at that time they were all in glass bulbs. So Langmuir began to study mercury vapor discharges. He shortly invented his probe, I did the experimental work and most of the mathematics, and we soon accumulated a lot of data about ion densities and velocity distributions in the mercury arc columns, in the positive columns of Geissler tubes, and in gas-filled thermionic tubes.

“We noticed the similarity of the discharge structures they revealed. Langmuir pointed out the importance and probable wide bearing of this fact. We struggled to find a name for it. For all members of the team realized that the credit for a discovery goes not to the man who makes it, but to the man who names it. Witness the name of our continent. We tossed around names like ‘uniform discharge’, ‘homogeneous discharge’, ‘equilibrium discharge’; and for the dark or light regions surrounding electrodes, names like ‘auras’, ‘haloes’, and so forth. But one day Langmuir came in triumphantly and said he had it. He pointed out that the ‘equilibrium’ part of the discharge acted as a sort of sub-stratum carrying particles of special kinds, like high-velocity electrons from thermionic filaments, molecules and ions of gas impurities. This reminds him of the way blood plasma carries around red and white corpuscles and germs. So he proposed to call our ‘uniform discharge’ a ‘plasma’. Of course we all agreed.

“But then we were in for it. For a long time we were pestered by requests from medical journals for reprints of our articles. This happens to me this day. The scientific world of physics and chemistry looked askance at this uncouth word and were slow to accept it in their vocabulary. The engineering world treated it as a GE trade name. Then all of a sudden, long after I had left the laboratory, to my pleased surprise, everybody started to talk about plasmas. This happened not long before they became thermonuclear and so government subsidized. That finally put the seal of respectability on plasmas.”

Yours faithfully,

Harold M. Mott-Smith

6909 SW Freeway,
Texas 77036
Basic Properties
Plasmas are surrounded by sheaths

- **Sheath**: A narrow region of strong electric field that acts to balance electron and ion currents lost from a plasma.
- **Recall**: electrons move faster than ions
  \[
  \frac{v_{Te}}{v_{Ti}} = \sqrt{\frac{m_i T_e}{m_e T_i}} \approx 250 \text{ (uses } T_e = 1 \text{eV)}
  \]
  - Electrons quickly charge the boundary surface.
- **Sheaths** are an electrostatic barrier that forms naturally as ions are attracted to the electron surface charge.
Sheaths are thin (almost like a skin)

• Polarization shields electric fields in plasmas

• Debye length: \( \lambda_{De} = \sqrt{\frac{\varepsilon_0 T_e}{en}} = 740 \sqrt{\frac{T_e(eV)}{n(cm^{-3})}} \) cm

• For our example: \( T_e = 1eV, \ n = 1 \times 10^{10} cm^{-3} \):
  \( \lambda_{De} \approx 7 \times 10^{-5} m \) (tiny!)

• More complete analysis of sheath thickness (Child-Langmuir):
  \[
  \frac{l_s}{\lambda_D} = \frac{\sqrt{2}}{3} \left( \frac{2e \Delta \phi}{T_e} \right)^{3/4}
  \]

• Plasma potential: \( \Delta \phi \)
  • Determined by balancing electron and ion currents to boundary
David Bohm (1917-1992)

• Leading theoretical physicist of the 20th century
  – Hidden variables theory of quantum mechanics, Aharonov-Bohm effect, neuropsychology, philosophy
  – In plasma: Bohm diffusion, Bohm criterion, plasmon, random-phase approximation

• His war work was in plasma
  – Refused security clearance
  – Thomson -> Oppenhiemer -> Bohm
Bohm criterion and the presheath

- Ions must flow into the sheath supersonically

\[ V \geq c_s = \sqrt{\frac{T_e}{m_i}} \]

Bohm, in Characteristics of Electrical Discharges in Magnetic Fields, 1949
Plasma potential is determined by balancing electron and ion currents

• Ions are lost at the Bohm flux:

\[ \Gamma_i \approx 0.6en \frac{T_e}{m_i} \]

• The 0.6 is from the presheath density drop

• Electrons are lost at a thermal flux, reduced by the sheath:

\[ \Gamma_e \approx en \frac{T_e}{2\pi m_e} e^{-\Delta\phi/T_e} \]

• Setting these equal gives the plasma potential:

\[ \Delta\phi = \frac{T_e}{e} \ln \left( \sqrt{\frac{\pi m_i}{8m_e}} \right) \approx 3V \]
Summary of sheath properties

• 2 length scales
  – Sheath $\sim \lambda_D$
  – Presheath $\sim \lambda_{in} \gg \lambda_D$

• Child-Langmuir law
  \[ \frac{s}{\lambda_D} = 0.79 \left( \frac{e \Delta \phi}{T_e} \right)^{3/4} \]

• Bohm criterion
  \[ V_i \geq c_s = \sqrt{T_e / m_i} \]
  – Presheath potential drop
  \[ e \Delta \phi_{ps} \geq T_e / 2 \]
Plasma interaction with electric fields, such as sheaths, are the basis for many technologies

- Manufacturing microelectronics
  - Directed ion energy enables etching of large aspect ratio trenches
  - Enabled Moore’s law
- Chemistry at the surface is also important
  - Reactive ion etching


Figure from: http://www.appliedmaterials.com/
Sheaths also influence the chemistry

- Transfer of plasma-generated reactive species to materials
  - Happens through a sheath
- The sheath influences the electron energy distribution function (EEDF) in the plasma
  - The EEDF controls the rate of chemical reactions
  - Modeling the EEDF is a complex plasma-boundary interaction problem, mediated by the sheath

Plasmas enable more of our modern way of life than you might realize.
Low Temperature Plasma is an exciting area of science and engineering

• Fast-paced
  • Short timeframe from fundamental research to engineering implementation
• Interdisciplinary
  • Source design must be coupled with desired outcome
• Important
  • Enables many modern technologies
  • And you may not have even realized it!
Suggested References

Principles of Plasma Discharges and Materials Processing
Second Edition

Michael A. Lieberman and Allan J. Lichtenberg